



Oct. 22, 2021



Bio-integrated Materials Science (Online Lectures)

Bio-integrated Metallic Nanomaterials

Lecture 7

Prof. Jung Heon Lee

Metals in Periodic Table

hydrogen 1 H 1.0079																	helium 2 He 4.0026						
lithium 3 Li 6.941	beryllium 4 Be 9.0122																	boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg 24.305																	aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80						
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29						
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]					
francium 87 Fr [223]	radium 88 Ra [226]	89-102 * *	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnium 110 Uun [271]	ununium 111 Uuu [272]	ununium 112 Uub [277]	ununquadium 114 Uuq [289]										

* Lanthanide series

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

** Actinide series



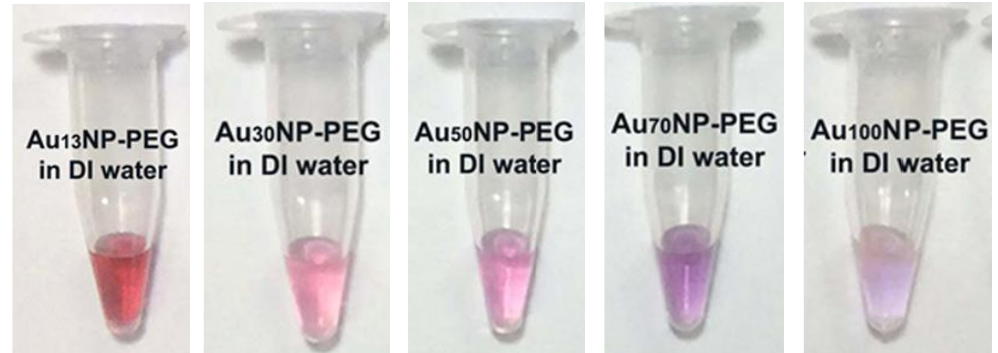
Gold in nanoscale



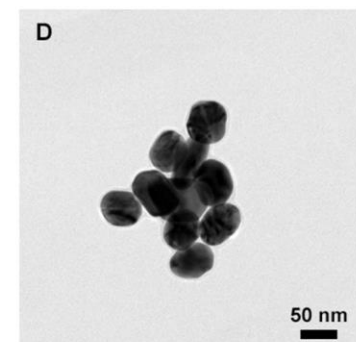
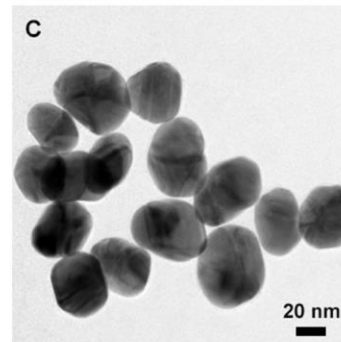
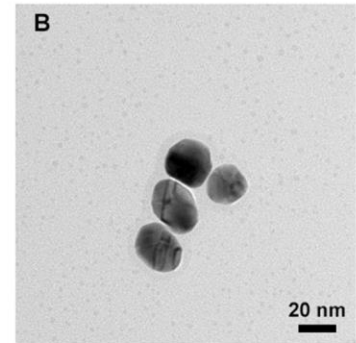
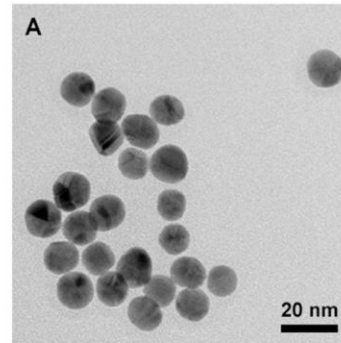
Bulk Gold

- Atomic number 79
- Electron rich
- Chemically inert
- Stable (noncorrosive)
- Great biological compatibility
- Low toxicity
- \$\$

Nano size

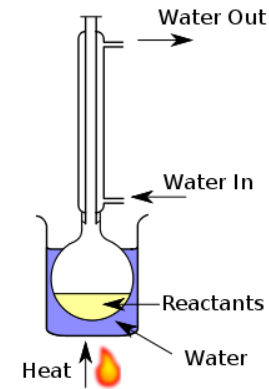
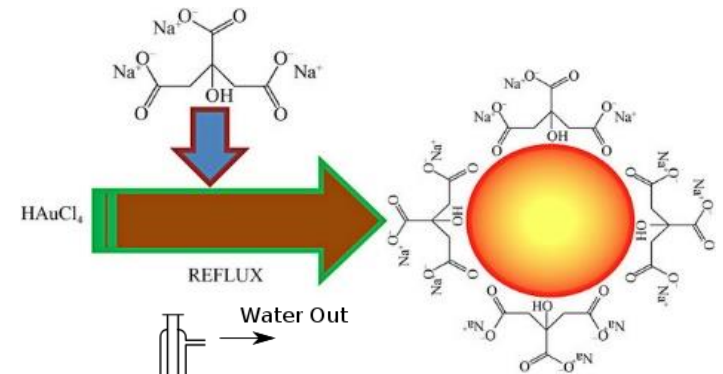


Gold nanoparticles

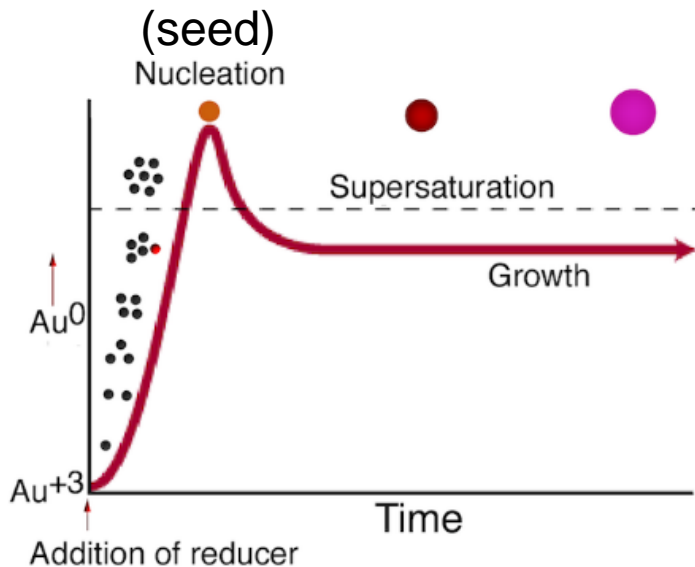


Synthesis of gold nanoparticles

- Citrate reduction method
- Developed by Turkevitch in 1951
- Sodium citrate + HAuCl_4
- High yield and monodisperse
- Cheap, easy
- Citrate helps to keep the stability of gold nanoparticles



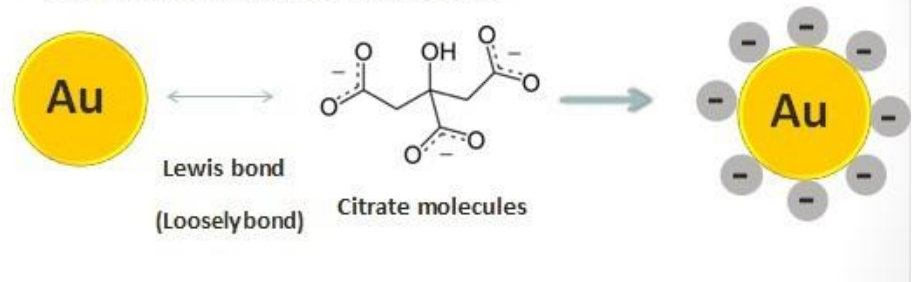
Reflux process



Nucleation and growth

Colloidal stability

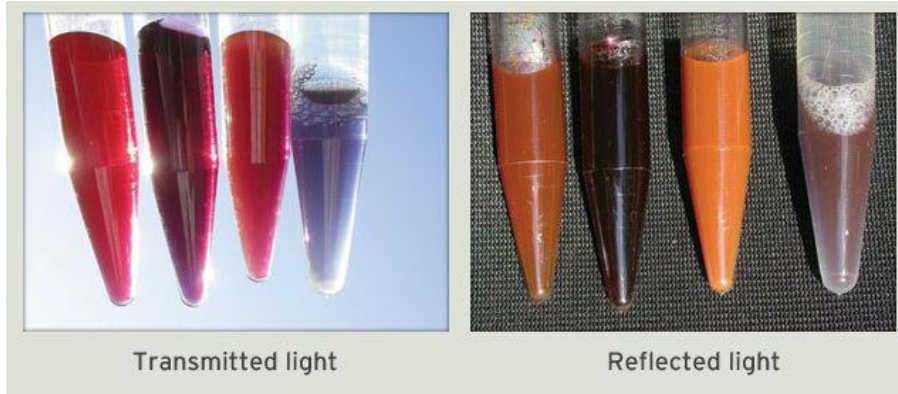
- Citrate stabilized gold nanoparticles



Extinction

Extinction = absorbance + scattering

↳ Absorb light ↳ Reflect light

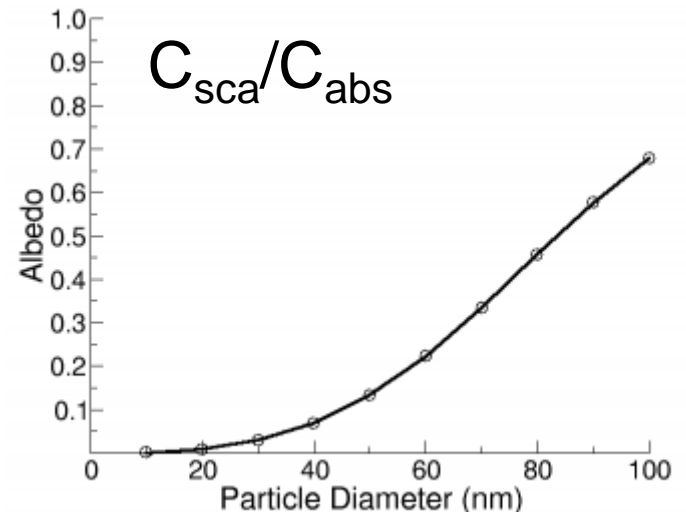


Color of identical gold nanoparticles with transmitted and reflected light

$$C_{abs} = 4\pi k a^3 \text{Im} \left[\frac{\varepsilon - \varepsilon_m}{\varepsilon + 2\varepsilon_m} \right]$$

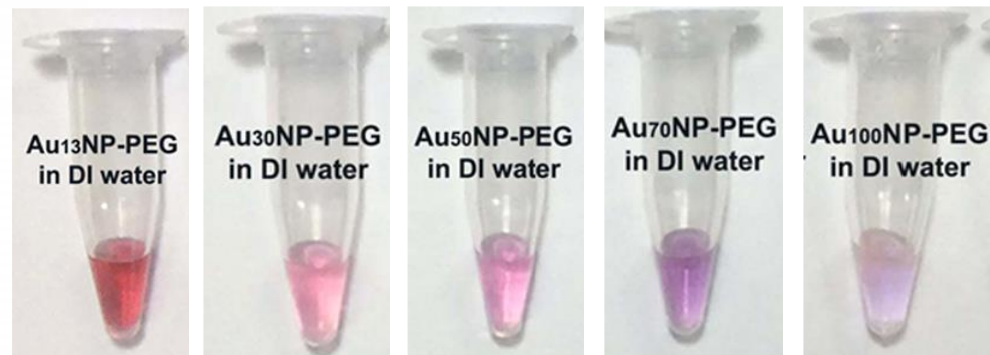
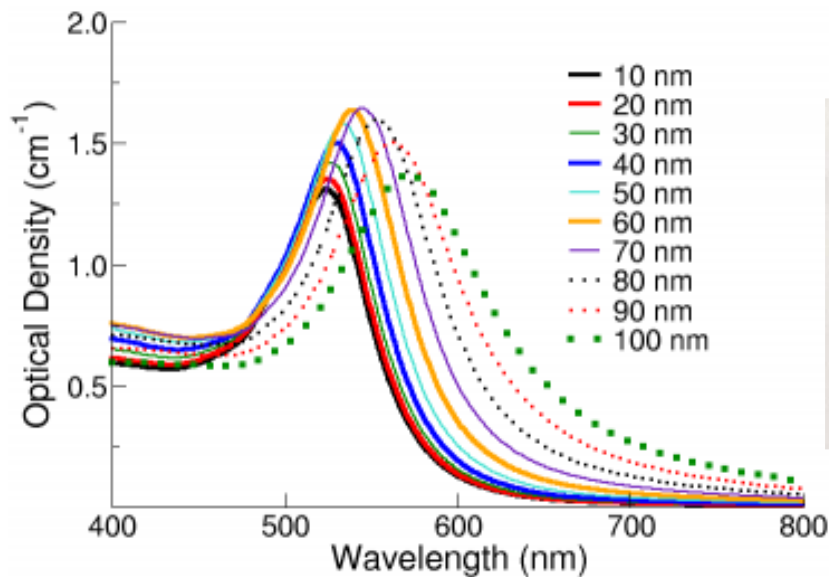
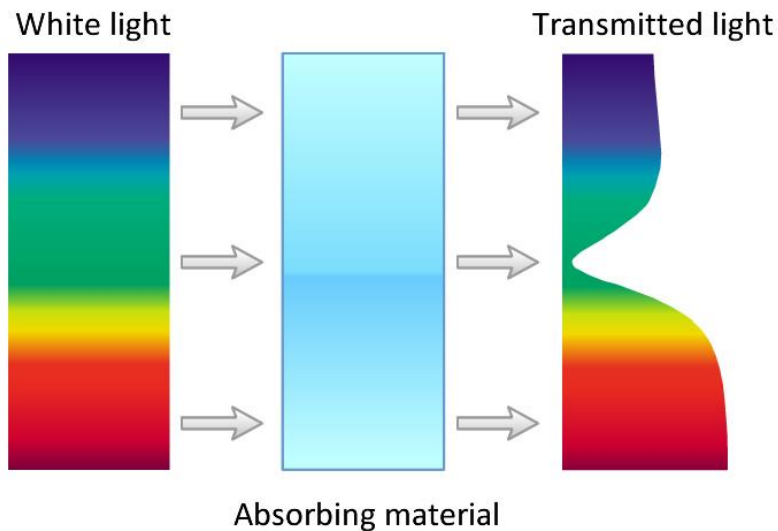
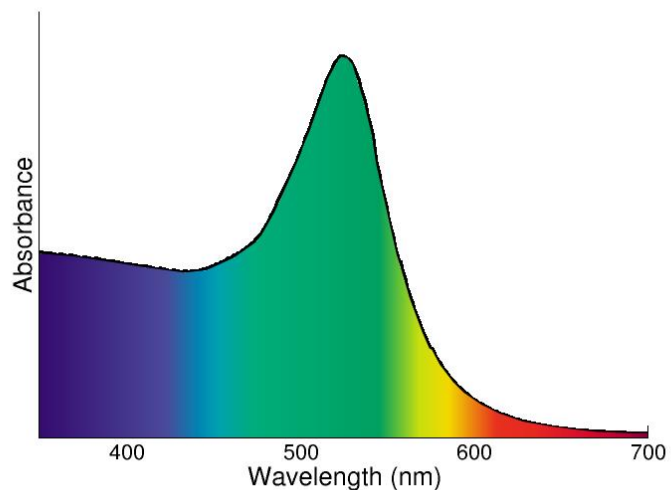
$$C_{sca} = \frac{8\pi}{3} k^4 a^6 \left| \frac{\varepsilon - \varepsilon_m}{\varepsilon + 2\varepsilon_m} \right|^2$$

- Absorbance is dominant when a is small
- Scattering is dominant when a is large



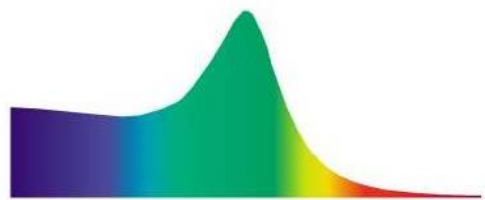
Why is gold nanoparticle red?

Absorbance of gold nanoparticle



Gold nanoparticles are not always red!

Scattering spectra of a nanoparticle



White light



Reflected light



Absorbing material



Ag Nanoprisms
~100 nm

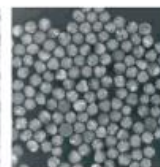
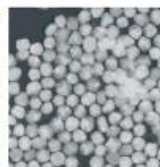
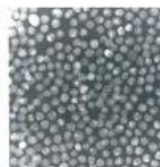
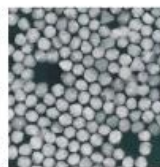
Au Spheres
~100 nm

Au Spheres
~50 nm

Ag Spheres
~100 nm

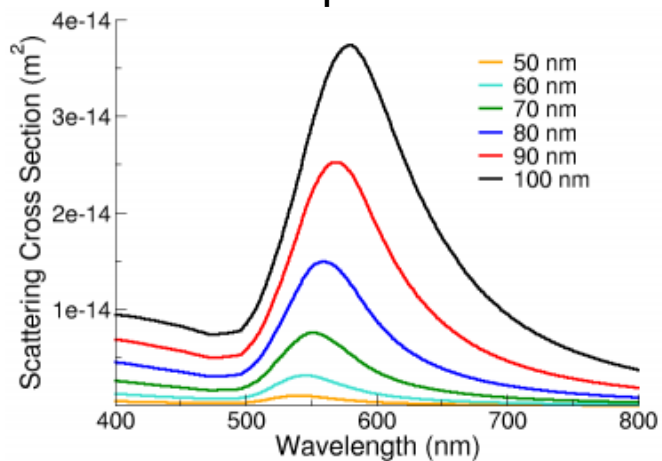
Ag Spheres
~80 nm

Ag Spheres
~40 nm



200nm (same for all the images)

Scattering spectra of gold nanoparticle

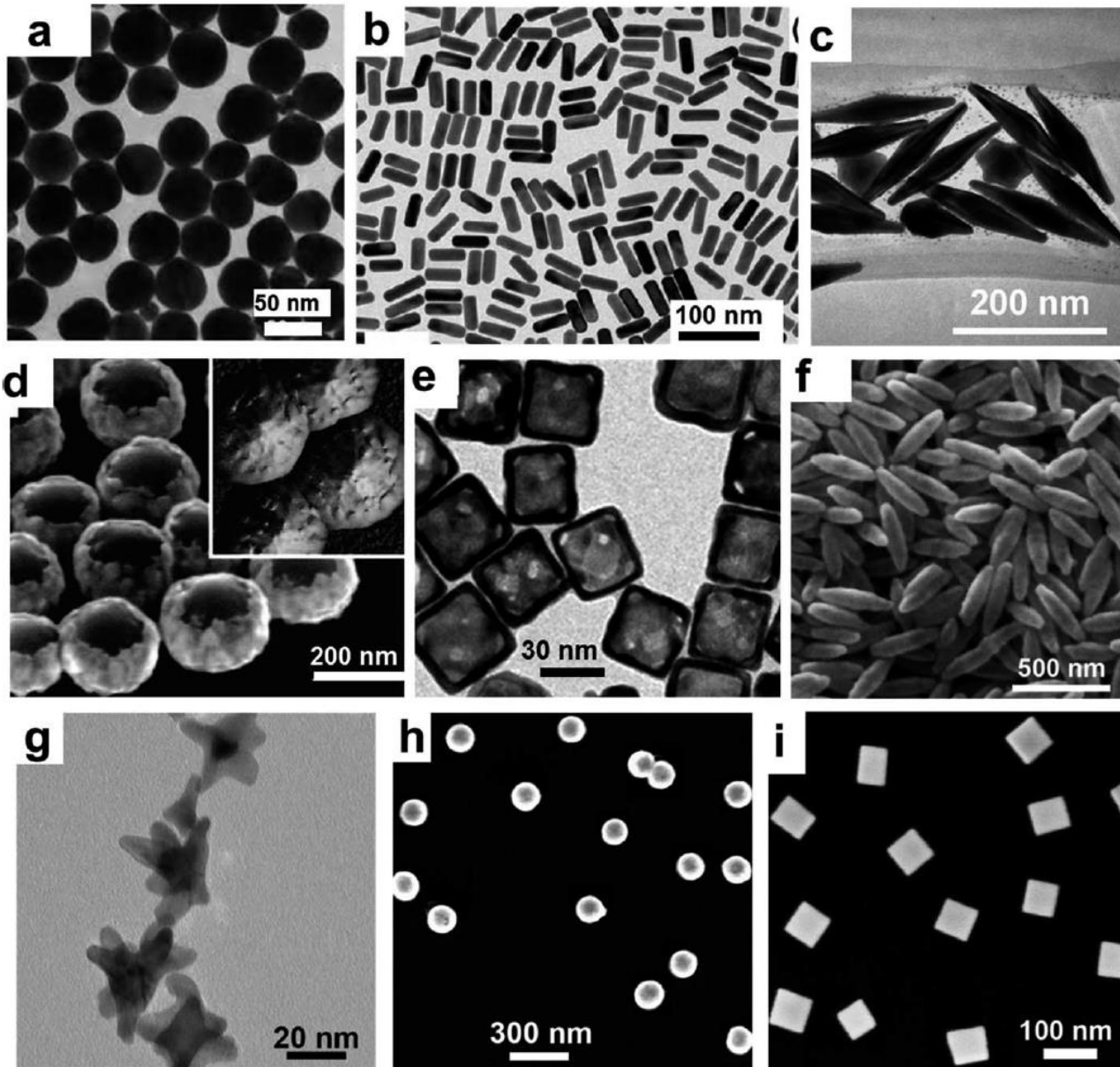


Lycurgus cup



Why does it look different?

Gold materials of different shapes



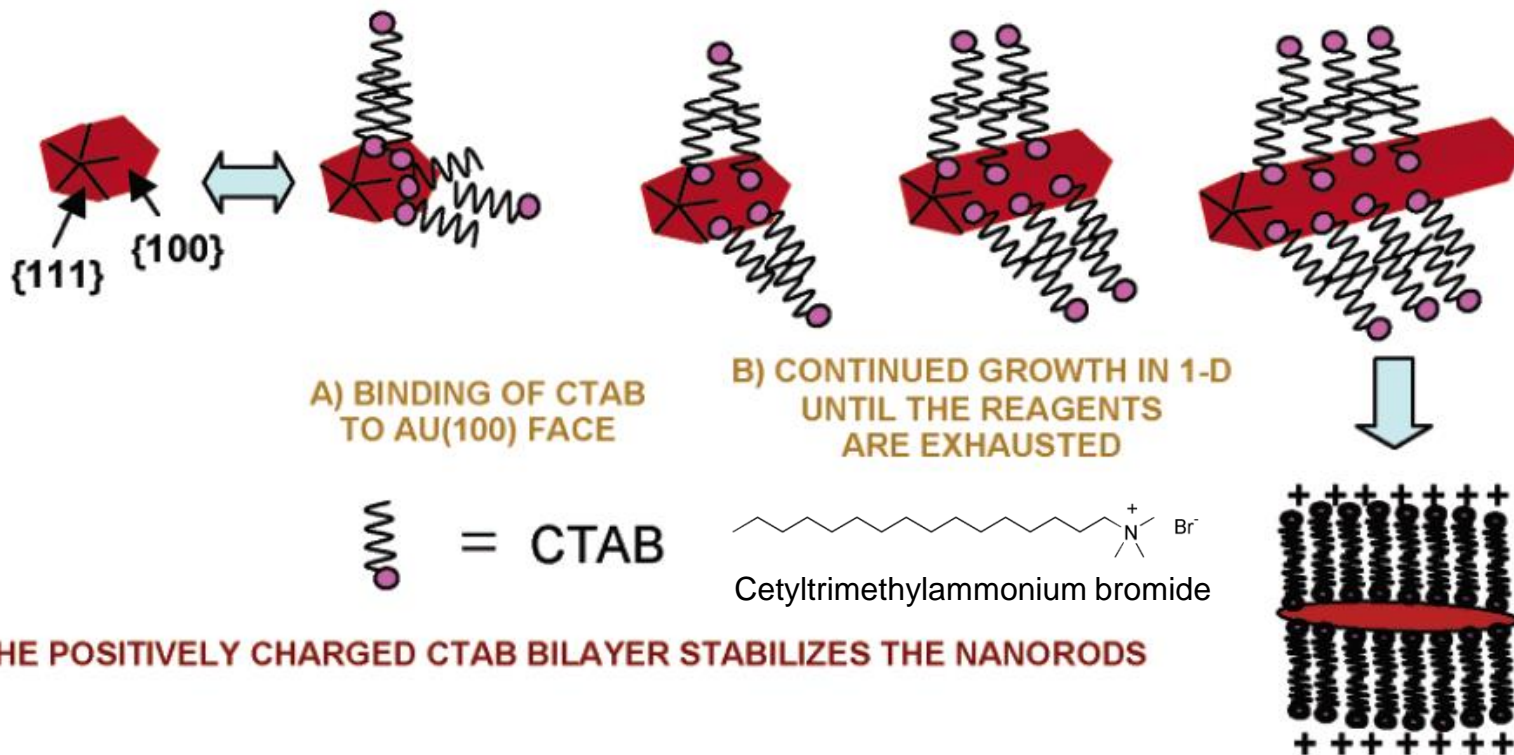
→ Each NP has different properties

Growth mechanism of gold nanorods

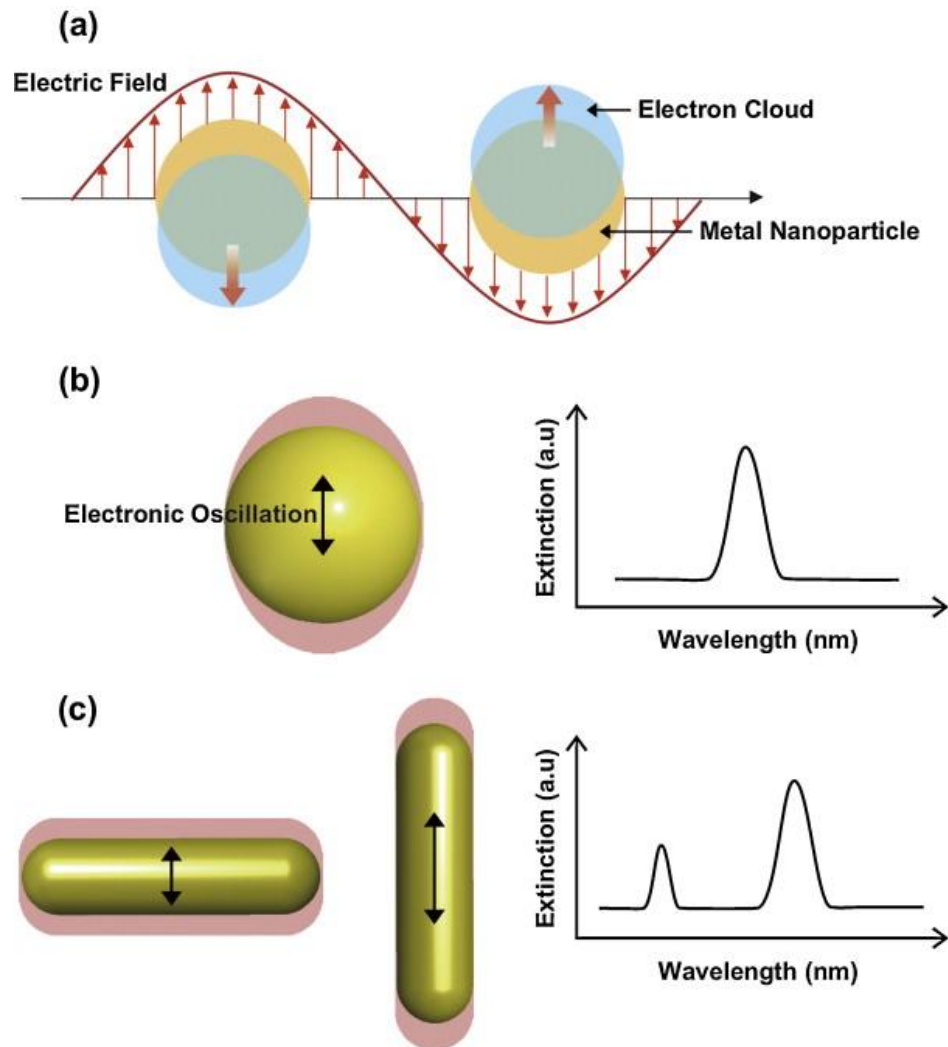
STEP 1: SYMMETRY BREAKING IN FCC METALS



STEP 2: PREFERENTIAL SURFACTANT BINDING TO SPECIFIC CRYSTAL FACES

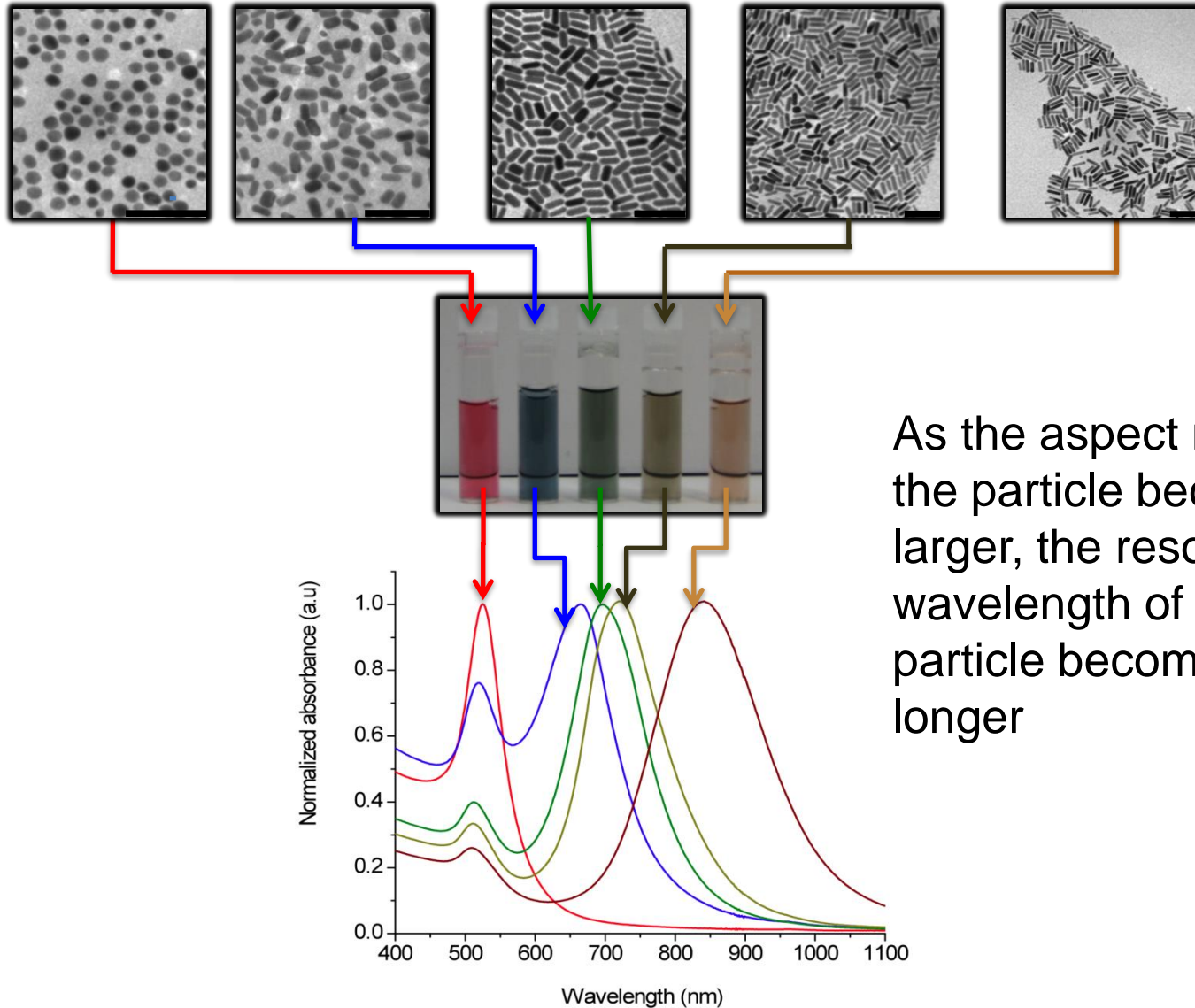


Surface plasmons of nanoparticles



Size and shape of the nanoparticle affect their resonance frequency

Shape variation of gold nanoparticles

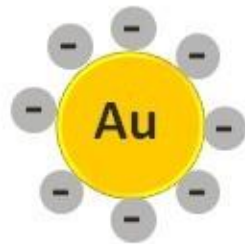


As the aspect ratio of the particle becomes larger, the resonance wavelength of the particle becomes longer

Aggregation of gold nanoparticles

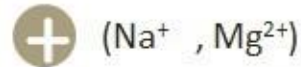
- Dispersed gold nanoparticles will have a red color in solution.
- If the particles aggregate, the solution will appear blue/purple and can progress to a clear solution with black precipitates.
- Why? Plasmonic coupling!

Colloidal stability

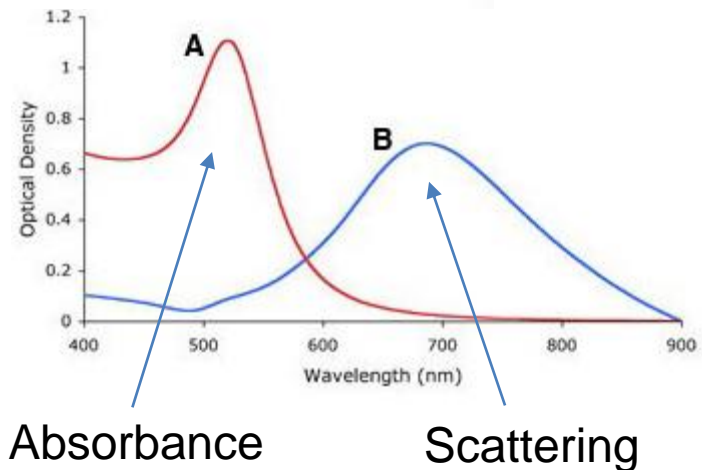
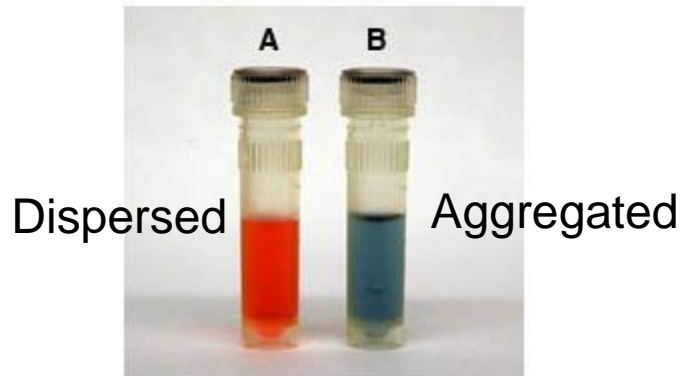


Salt buffer

+ Ex: NaCl or $MgCl_2$



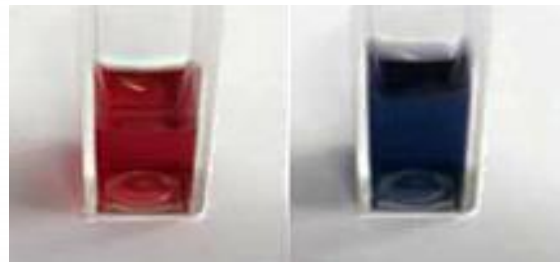
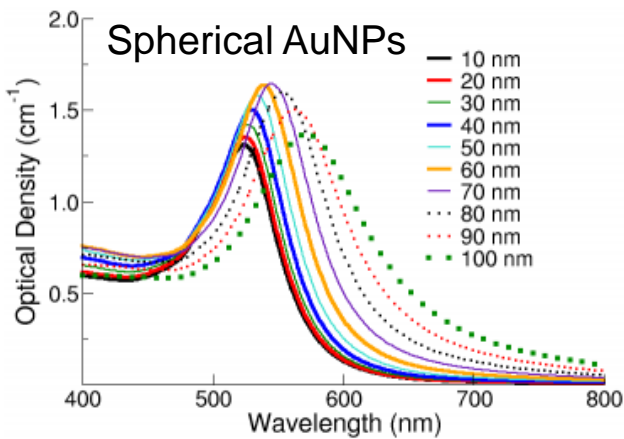
→ Aggregate



Metallic nanoparticles

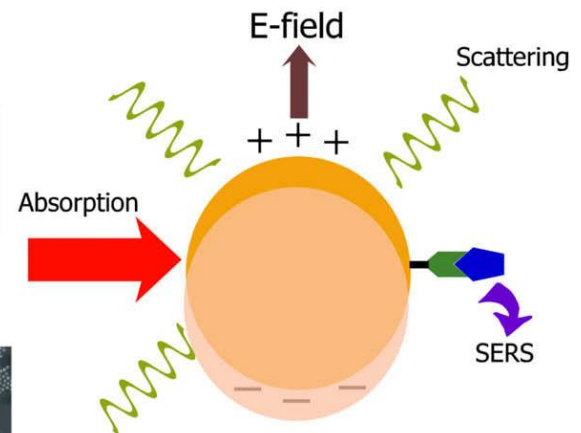
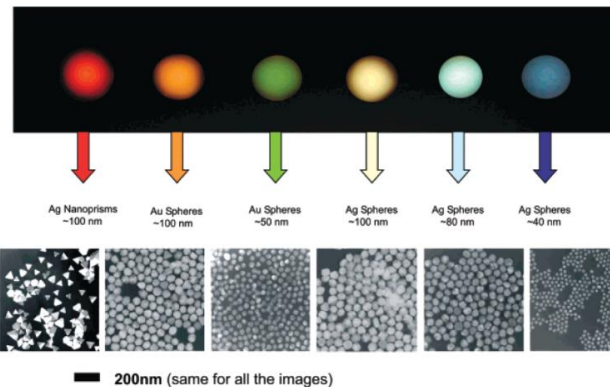
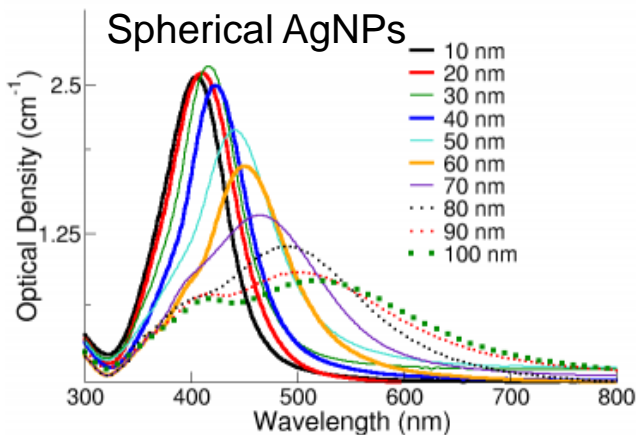
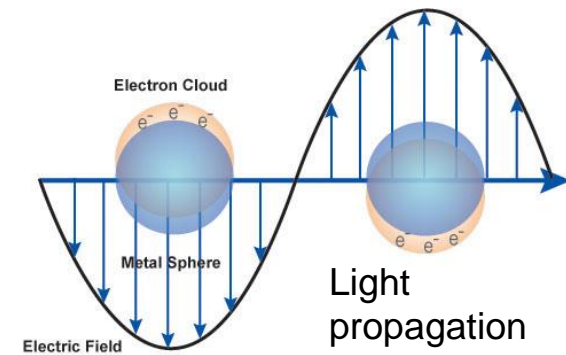
Surface plasmon resonance

- Free electrons oscillate collectively at the interface of metal and surrounding medium in resonance with external EM fields
- Dyes, SERS, fluorescence amplification, colorimetric sensing
- Can be applied in various biomedical imaging



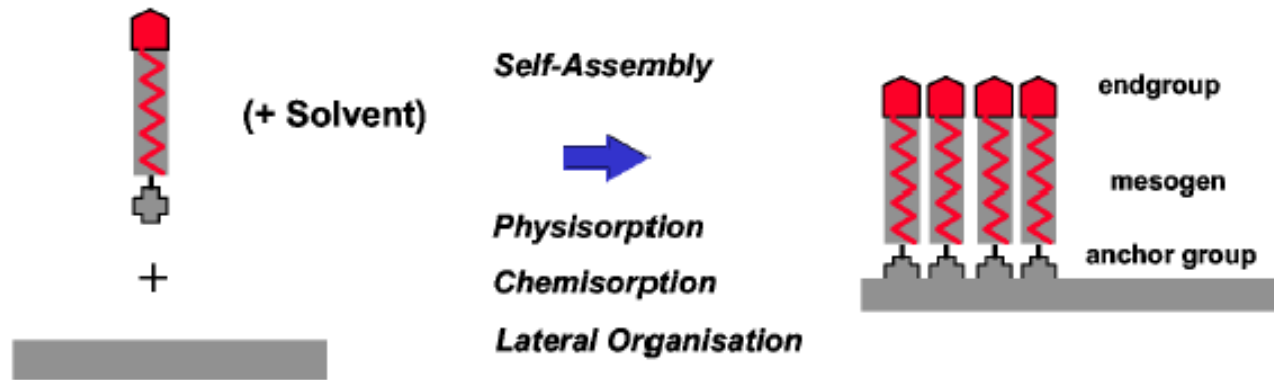
Dispersed

Aggregated

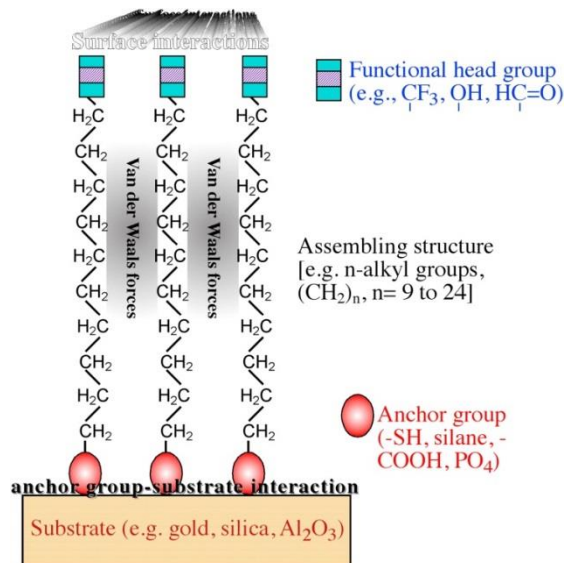


Self assembled monolayers (SAM)

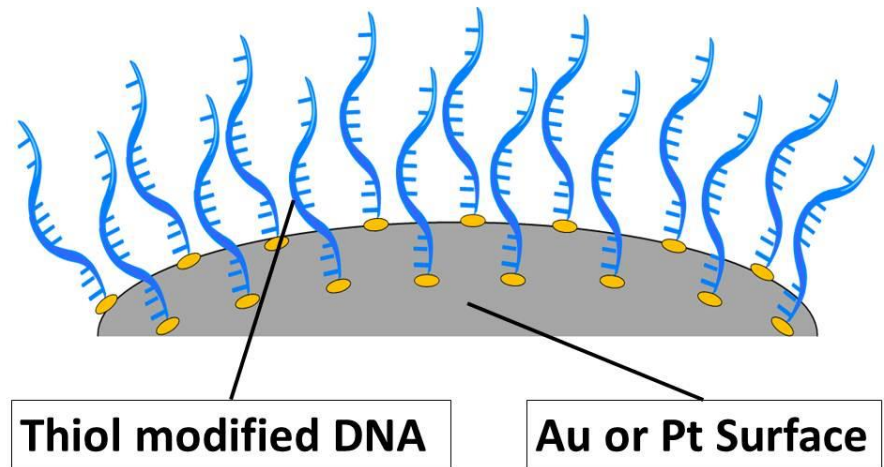
SAM (self-assembled monolayers): adsorption of molecules from solution onto solid substrates to form ordered molecular monolayers (e.g. alkylthiols on gold)



Gold-thiol chemistry

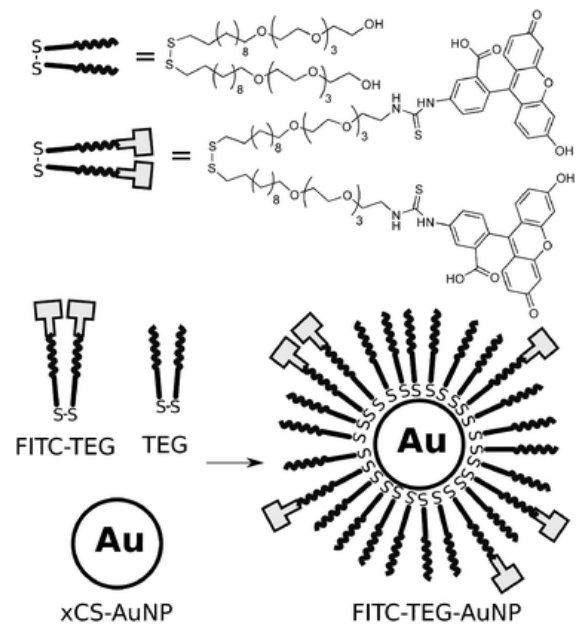
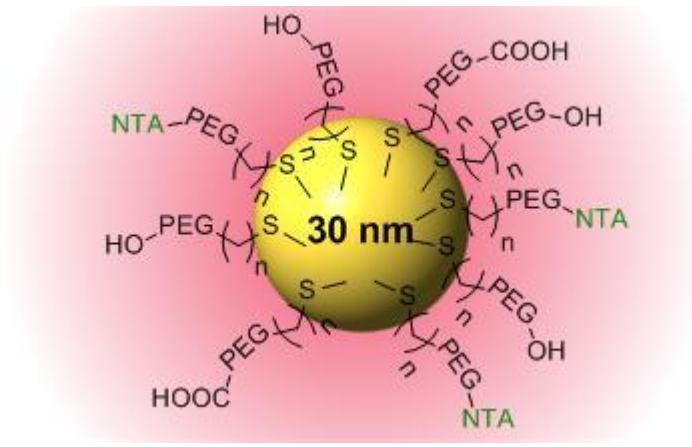


SAM conjugation

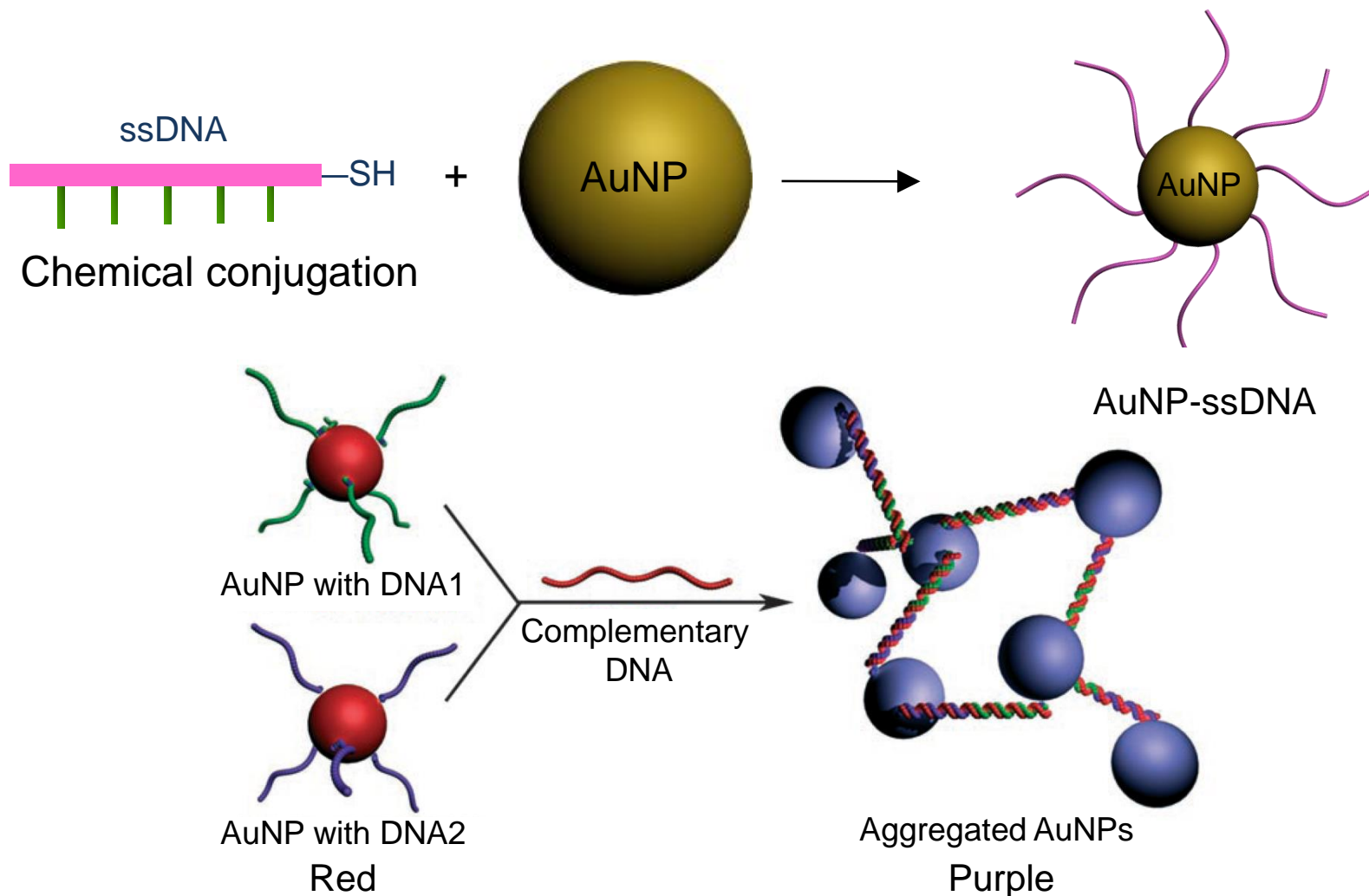


Chemical conjugation of AuNPs

Use SAM (Self assembled monolayer) on the surface of AuNP surface



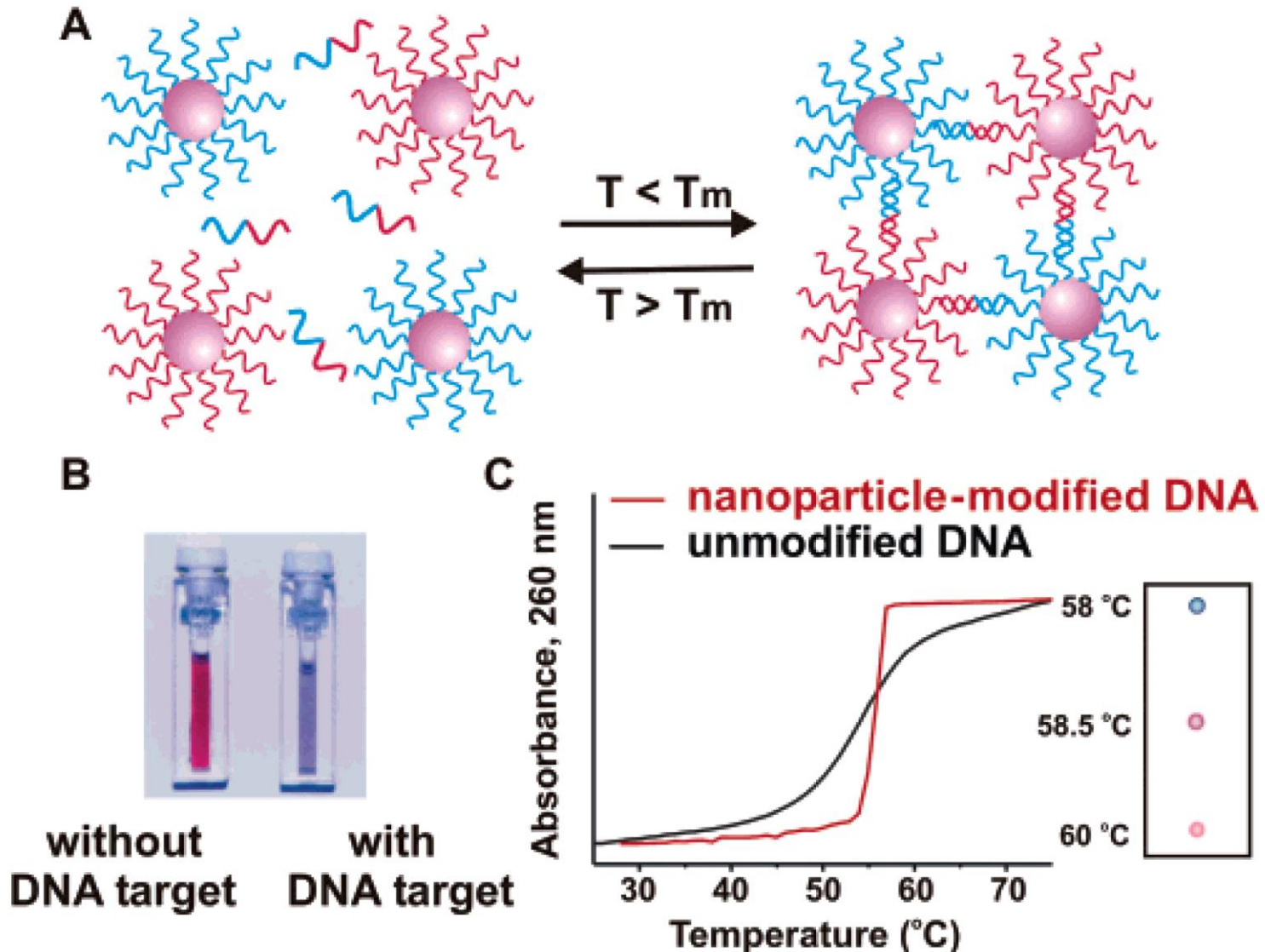
Chemical conjugation between AuNP and ssDNA



Single stranded (ss) DNA can be chemically attached on AuNP based on SAM

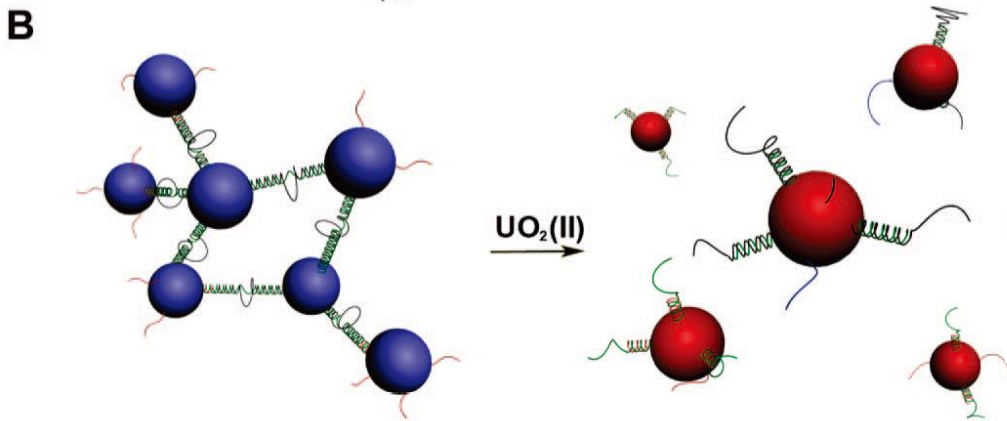
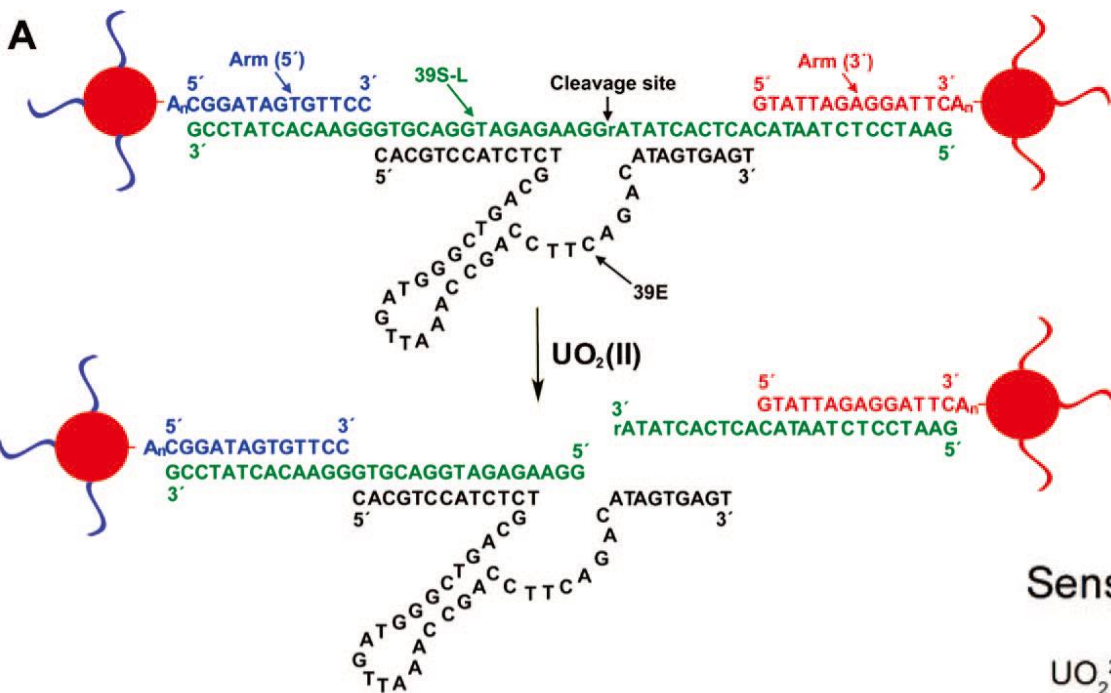
DNA conjugation on AuNPs

- The behavior of AuNPs can be controlled by the functionalized DNA



AuNP-DNA system for metal ion sensing

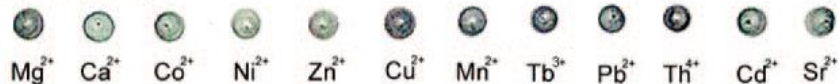
Metal ion specific aptamers integrated to AuNP system for colorimetric detection of metal ions



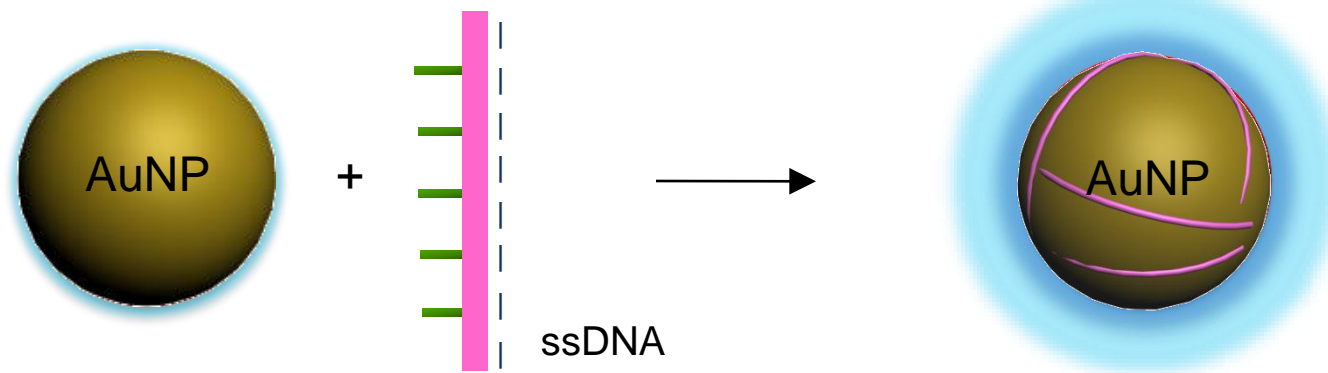
Sensitivity



Selectivity

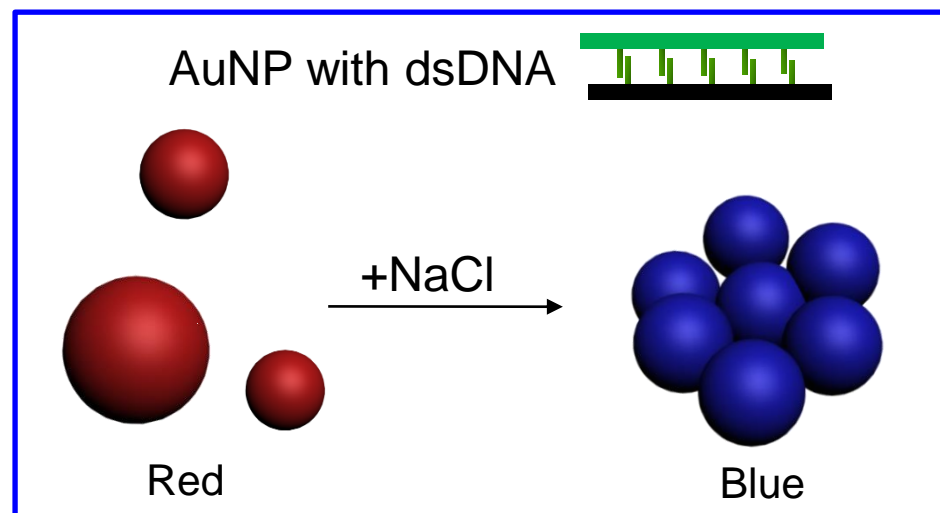
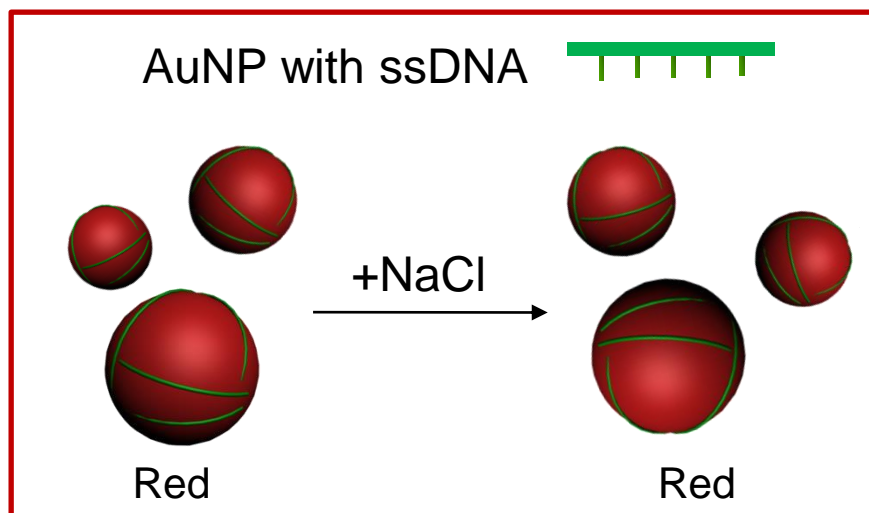


Physical interaction between AuNP and ssDNA



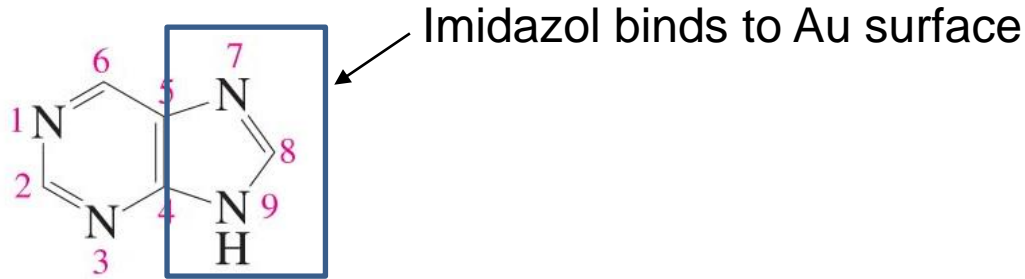
Physical interaction

AuNP with enhanced stability

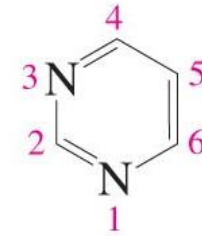


ssDNA can bind to AuNPs, but double stranded DNA cannot
ssDNA can improve the stability of AuNPs

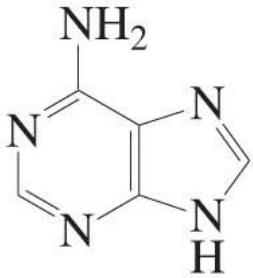
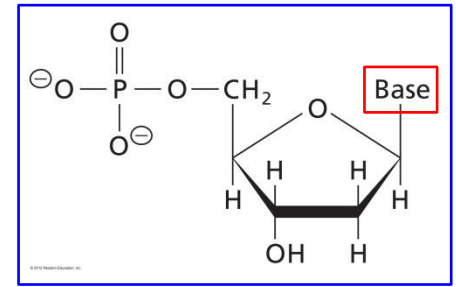
Bases



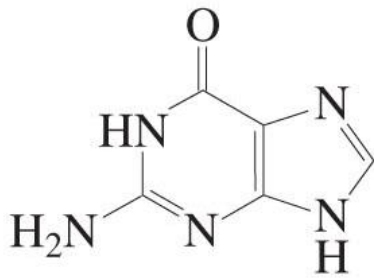
purine : Two rings with four "N"



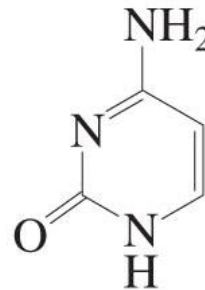
pyrimidine : One ring with two "N"



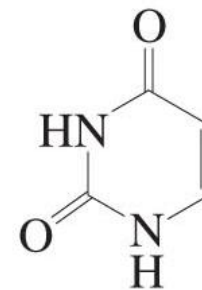
adenine



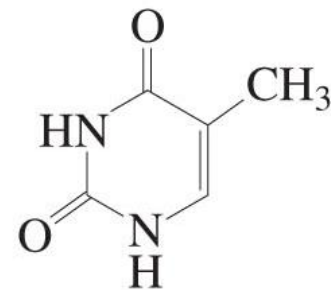
guanine



cytosine



uracil

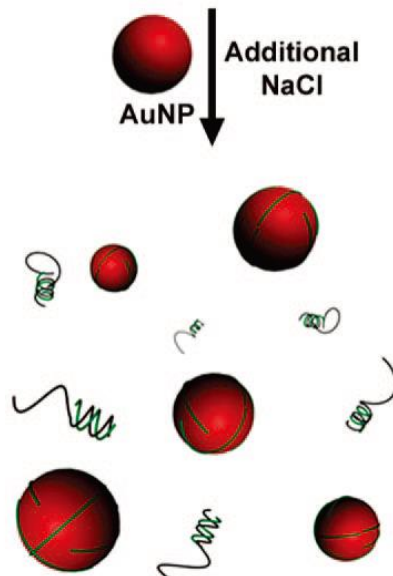
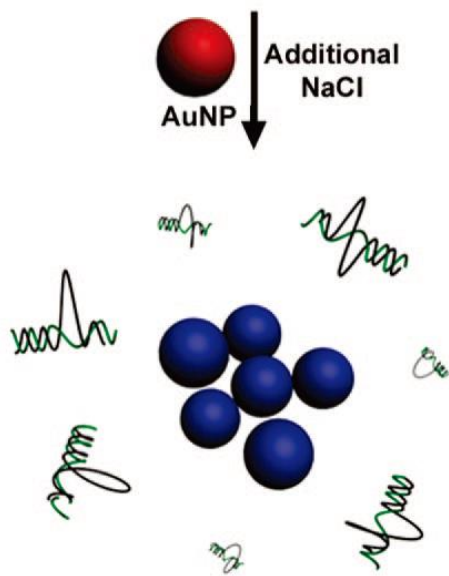
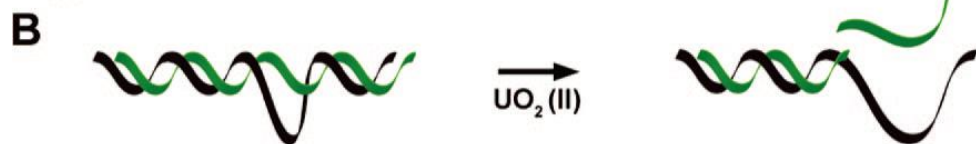
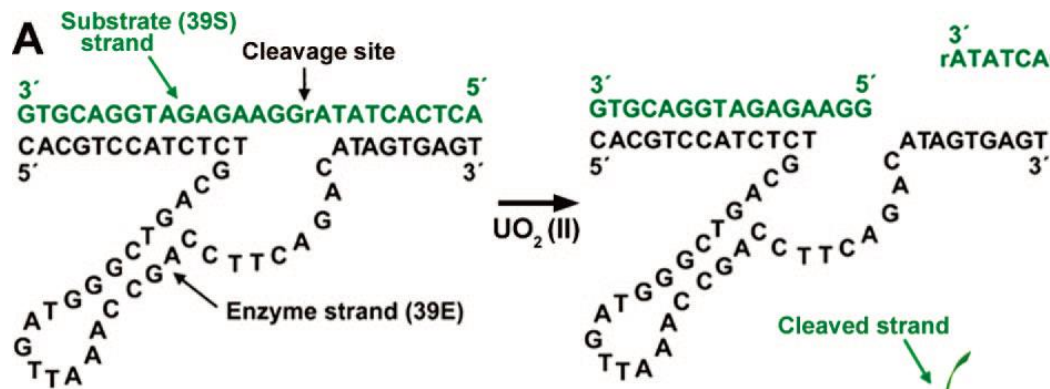


thymine

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- Adenine (A), guanine (G), cytosine (C), and thymine (T) are found in DNA.
- Adenine, guanine, cytosine, and uracil (U) are found in RNA
- Bases have ring structure → hydrophobic

Colorimetric sensors for detection of metal ions

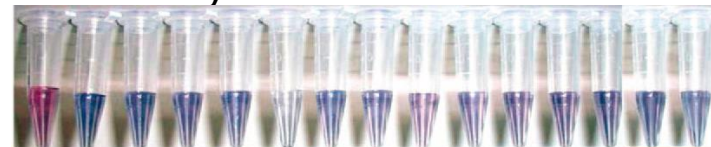


Sensitivity (DL: 1 nM)



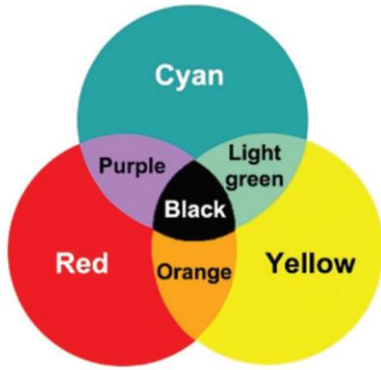
0nM 1nM 10nM 20nM 50nM 100nM 200nM 500nM 750nM 1µM 2µM UO_2^{2+}

Selectivity



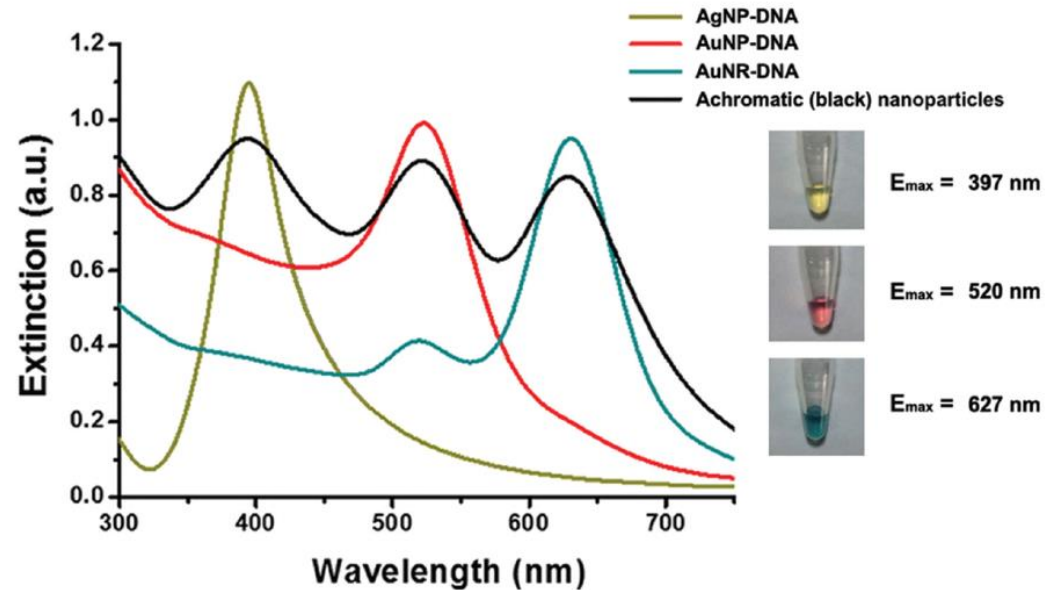
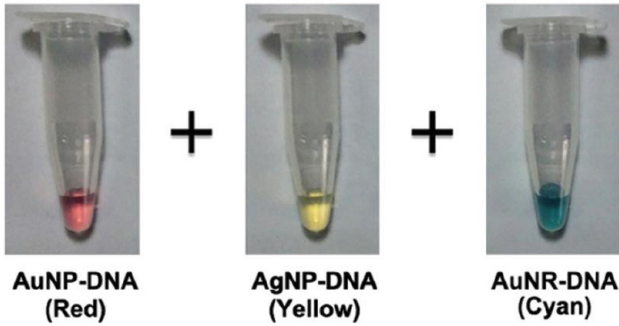
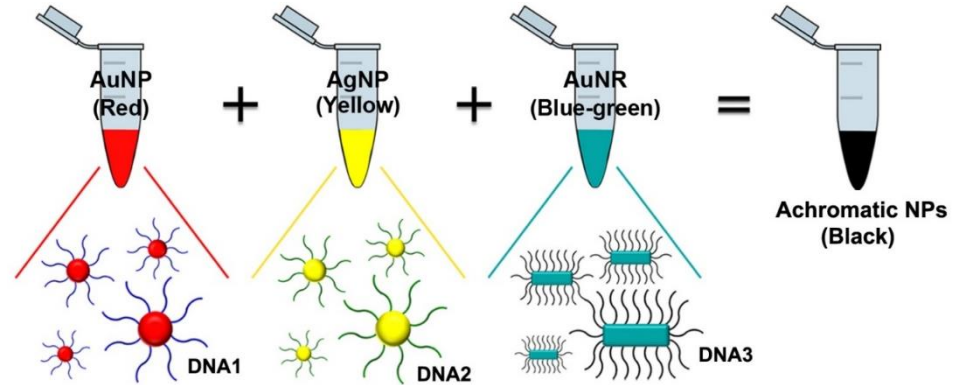
UO_2^{2+} Th^{3+} Tb^{3+} Pb^{2+} Co^{2+} Hg^{2+} Zn^{2+} Eu^{3+} Mn^{2+} Cu^{2+} Ni^{2+} Cd^{2+} Sr^{2+} Ca^{2+} Mg^{2+} (2 µM)

Achromatic (black) nanosensors based on CRYK model

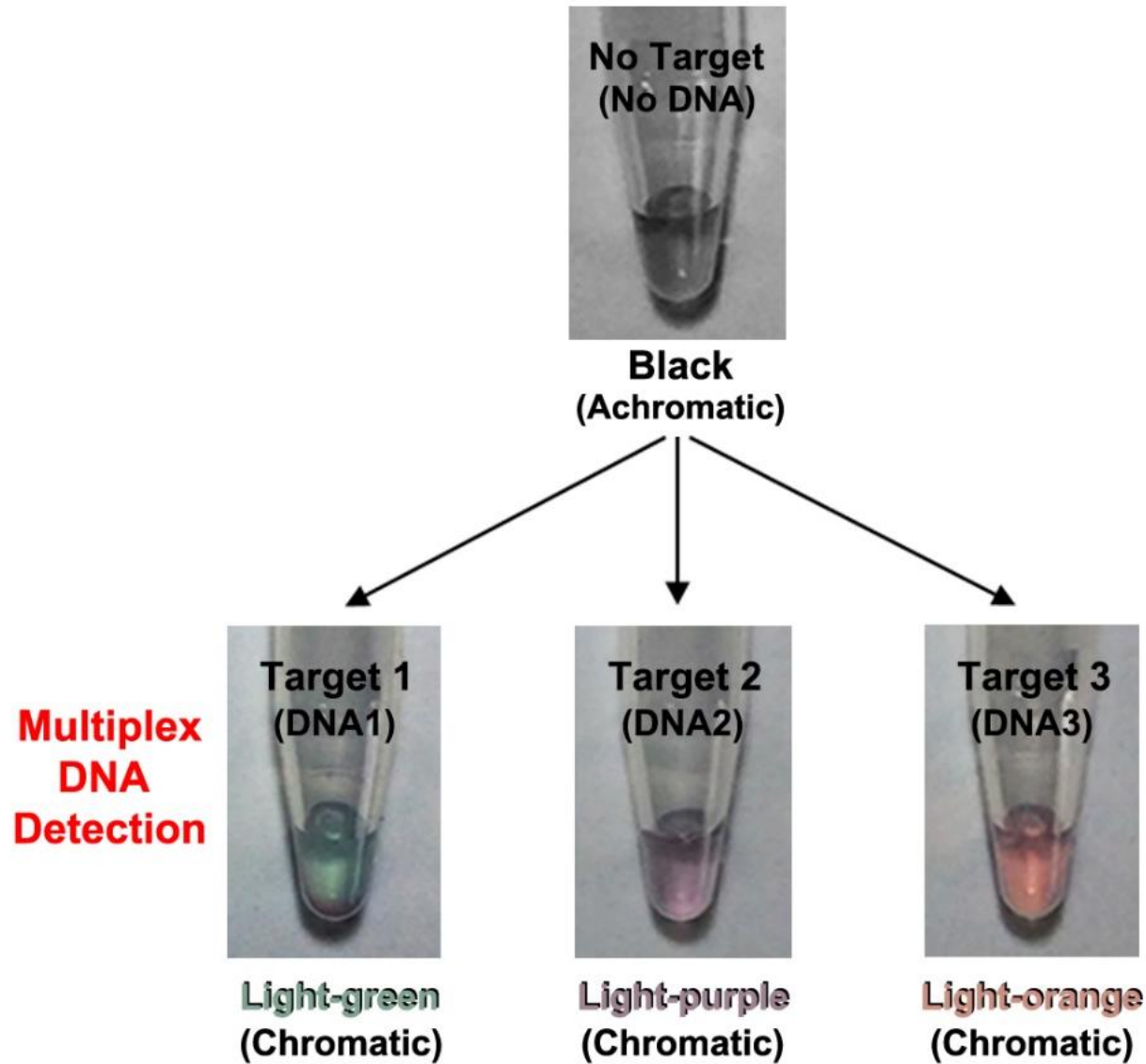


CRYK Color Model

Achromatic nanoparticles (Achromatic NPs)

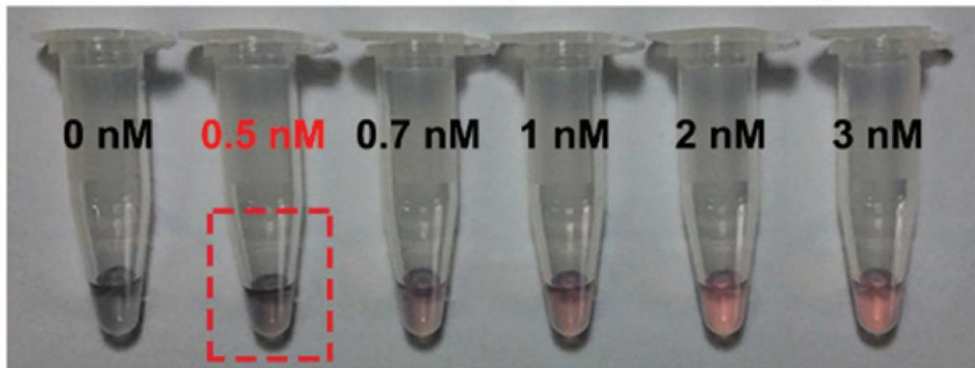


Achromatic (black) nanosensor for multiplexed detection

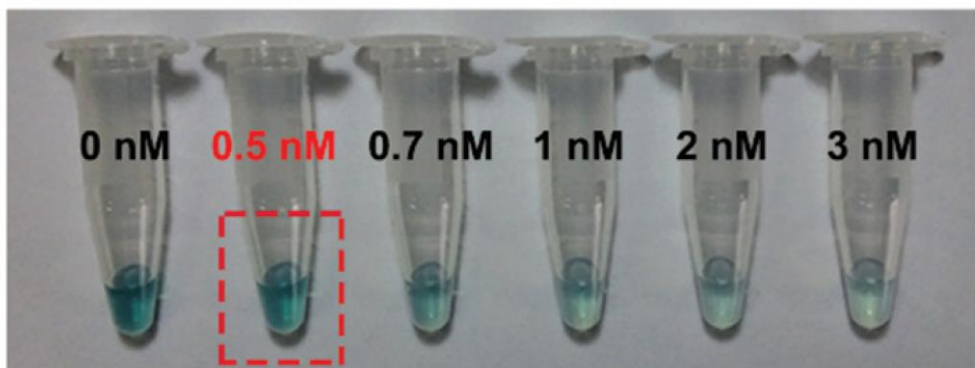


Enhancement of color transition with naked eyes

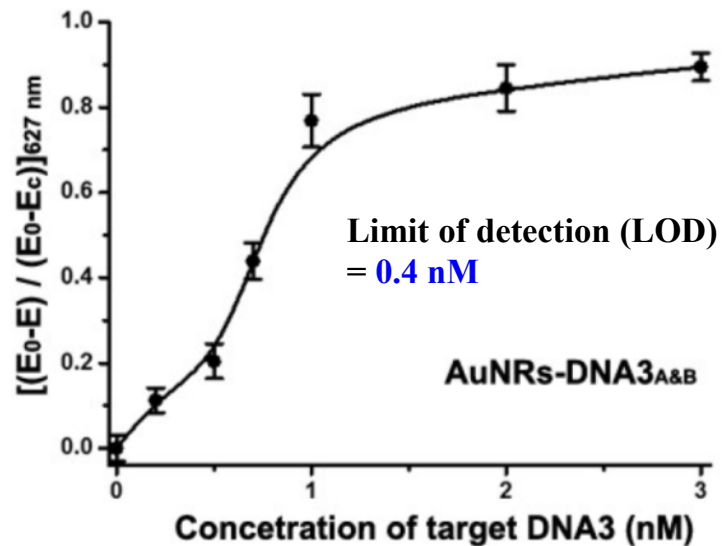
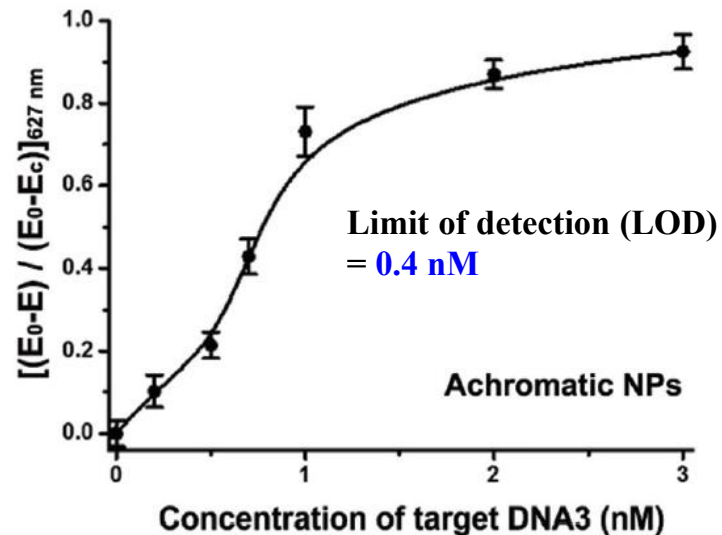
Achromatic nanoparticles (Achromatic NPs)



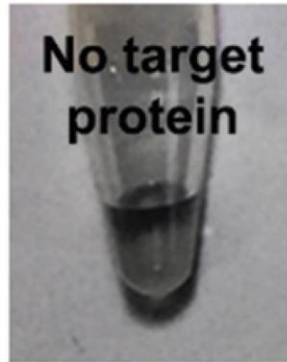
AuNRs-DNA3A&B



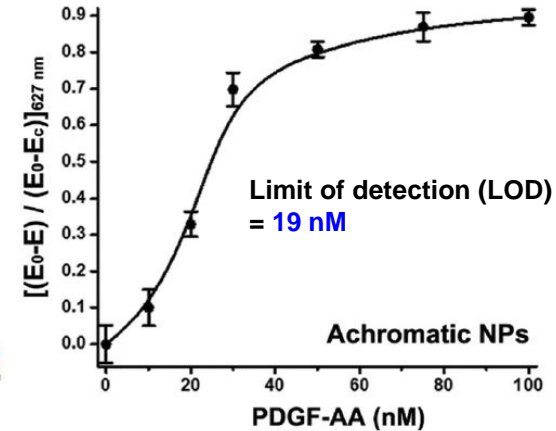
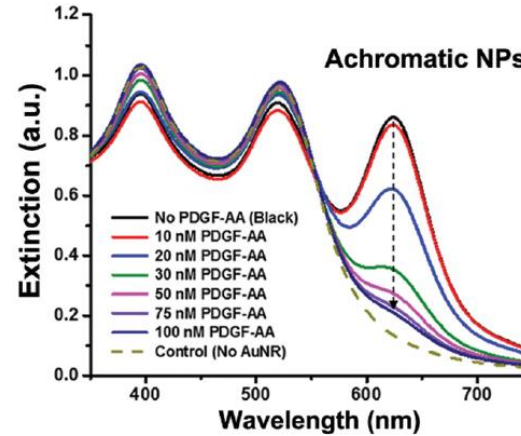
(Concentration of target DNA3 : 0 - 3 nM)



Achromatic sensors for biomarker detection



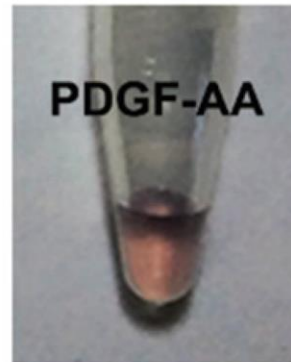
**Black
(Achromatic)**



**Multiplex Protein
Detection**

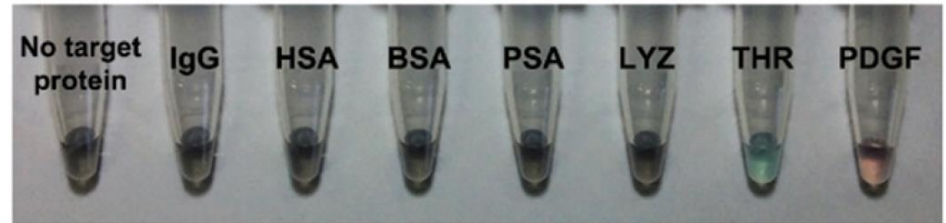


**Light-green
(Chromatic)**

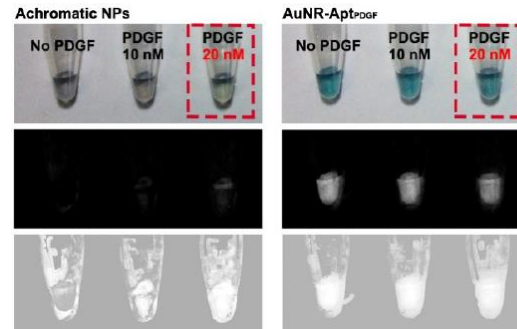


**Orange
(Chromatic)**

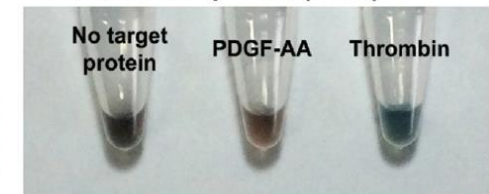
Selectivity



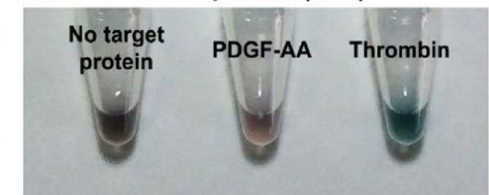
Color analysis



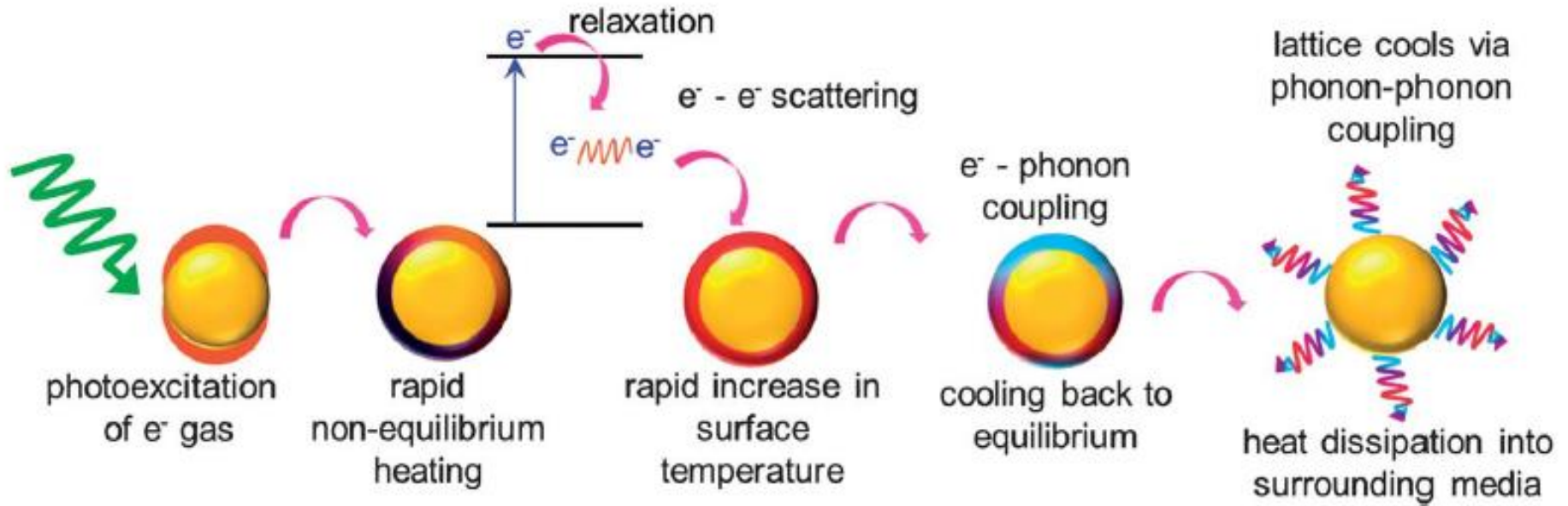
In Human blood plasma (100%)



In Human blood plasma (50%)

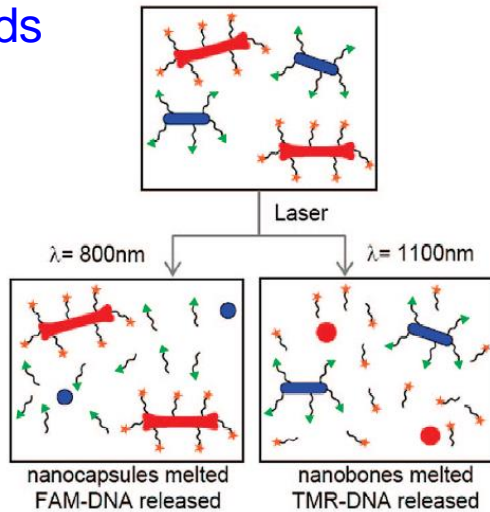


Photothermal phenomenon

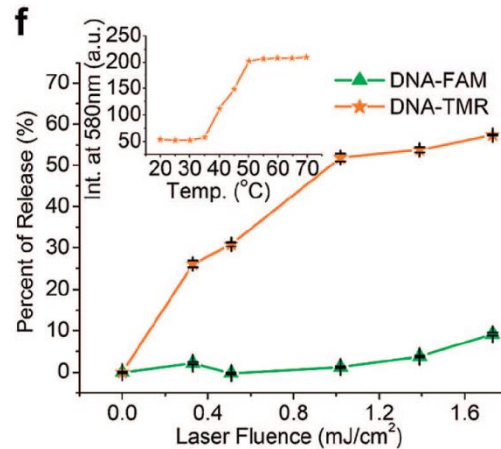
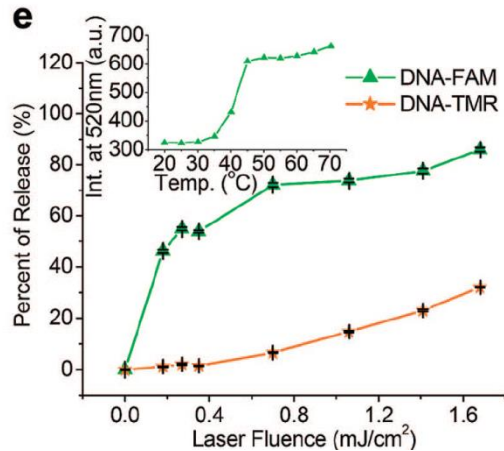
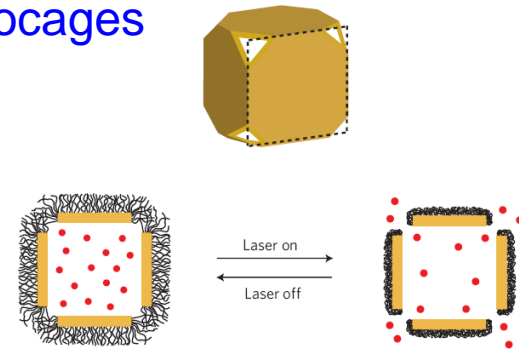


Photothermal driven release

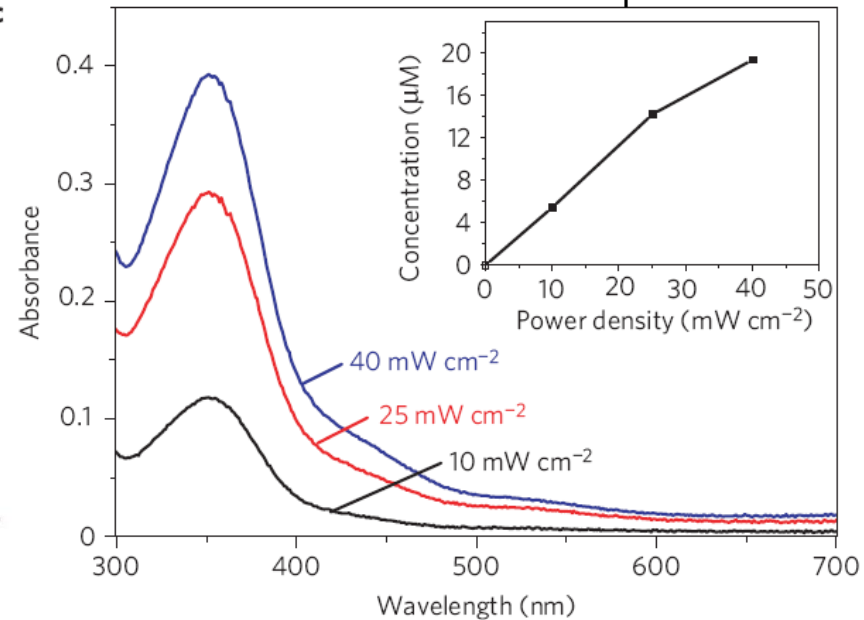
Nanorods



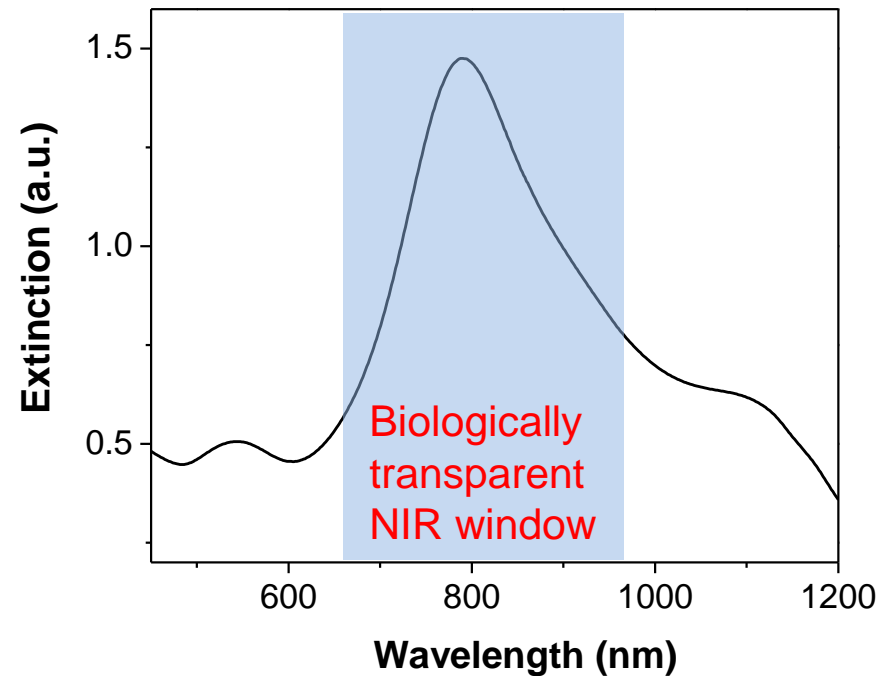
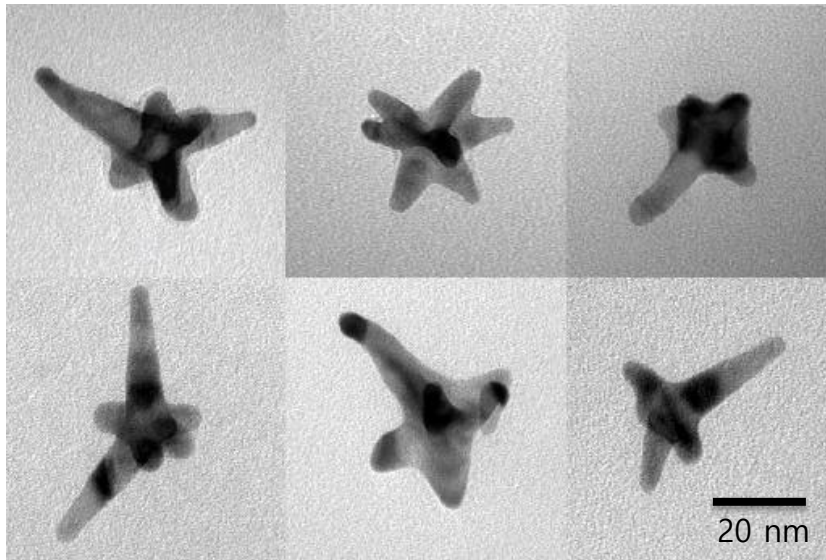
Nanocages



c 2 min irradiation with NIR pulsed laser



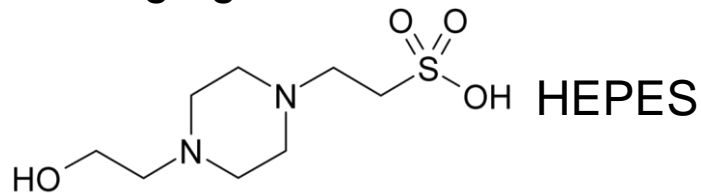
Gold nanostars



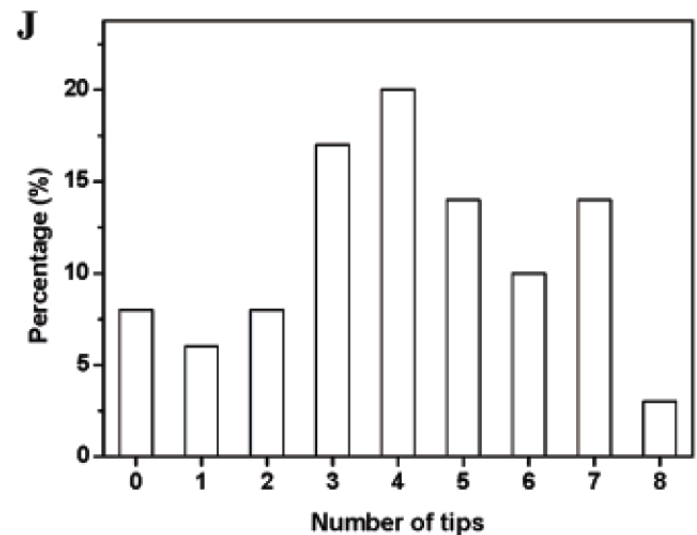
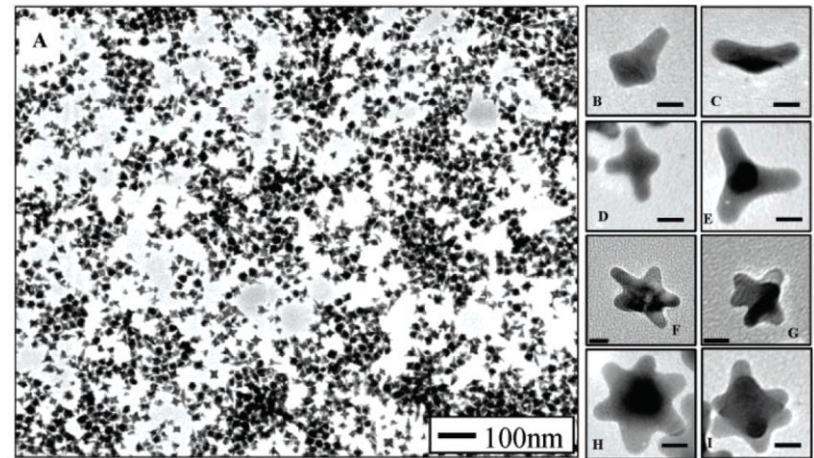
- ❑ Biocompatibility (HEPES is a buffer)
- ❑ Surface plasmon resonance at ~750-800 nm
- ❑ Small size (~30-40 nm)

HEPES gold nanostars

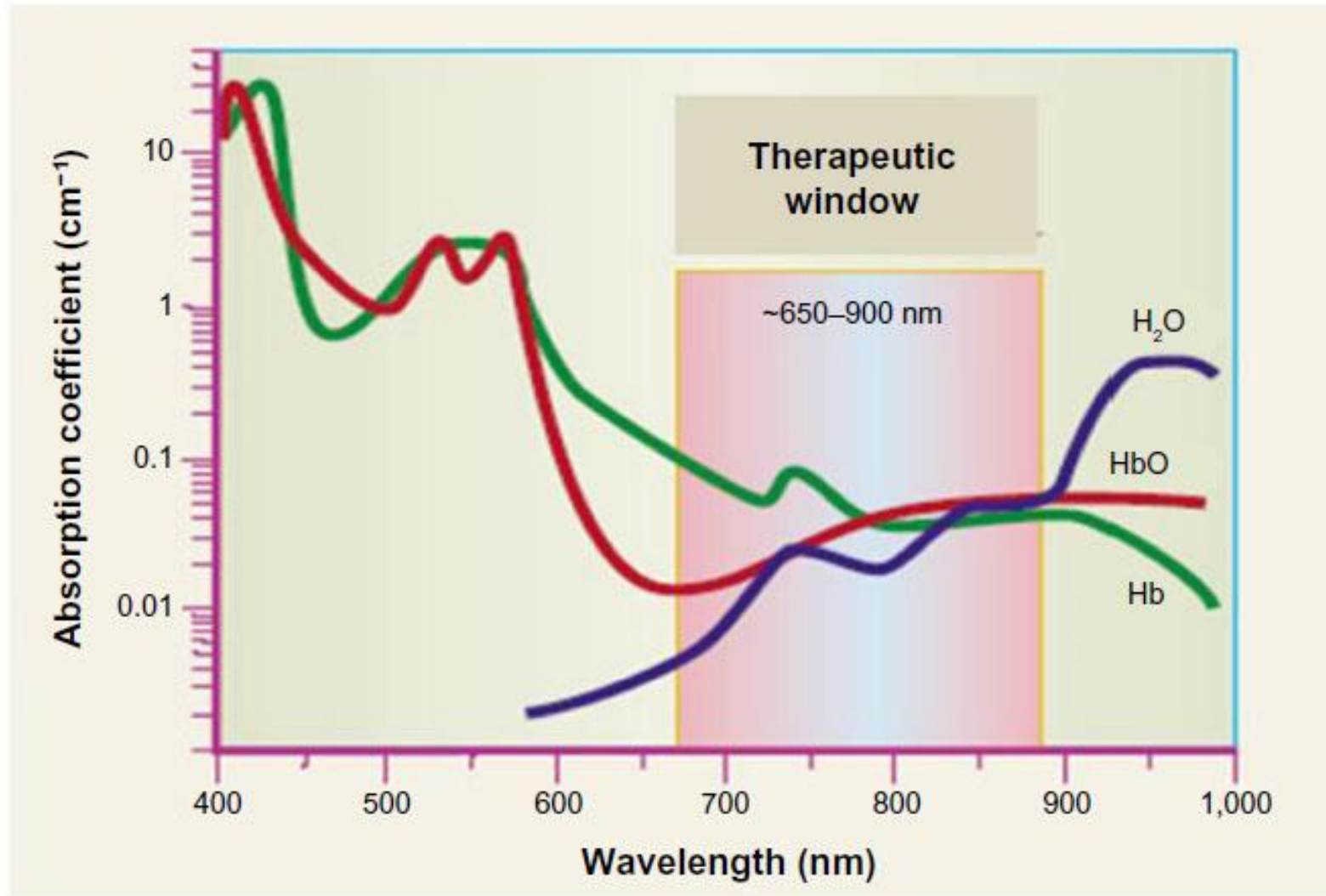
- 2-[4-(2-hydroxyethyl)-1-piperazinyl]ethanesulfonic acid (HEPES) is used as both reducing and shape-directing agents.



- Branched gold nanocrystals were formed by selective tip growth in the $\langle 111 \rangle$ directions, suggesting relatively weak or no adsorption of HEPES on the $\{111\}$ planes compared with the $\{100\}$ planes
- Among three functional groups in the HEPES molecule (hydroxyl, sulfonate, and piperazine), piperazine is responsible for generating branched structure

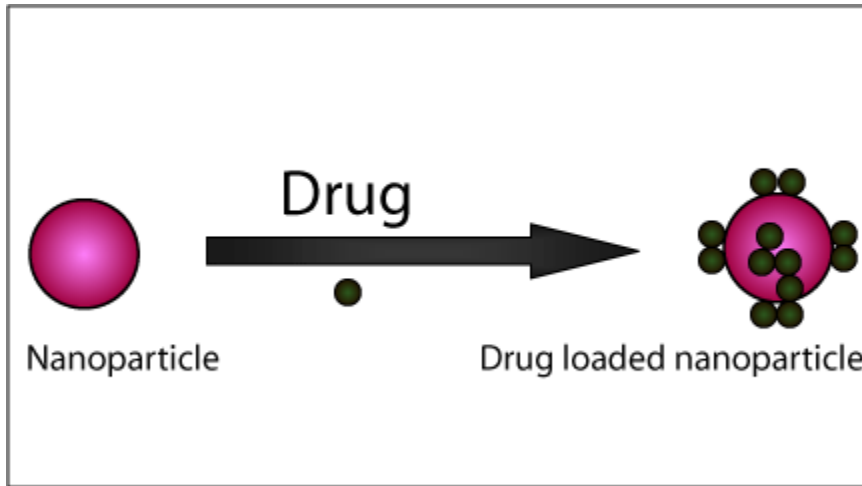


Biologically transparent window

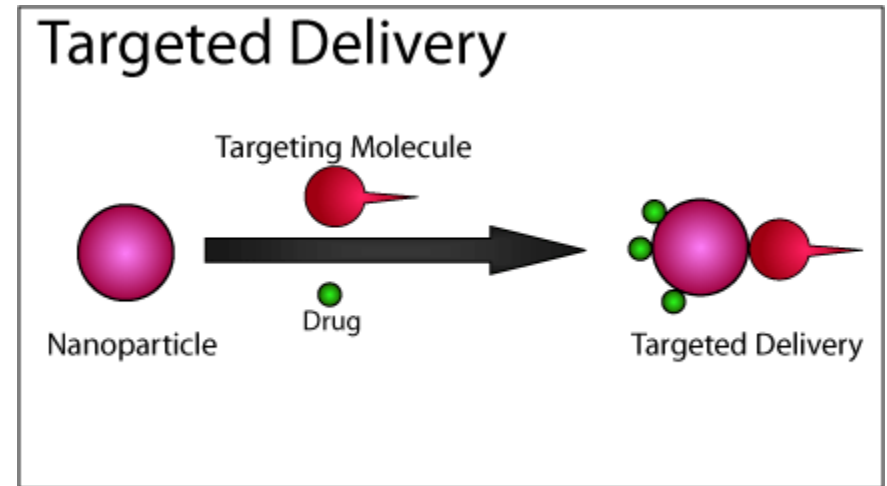


- Wavelengths of light from 650 nm to 900 nm are minimally absorbed and preferentially scattered upon interaction with tissue allowing for deeper light penetration than possible at other optical wavelengths

Targeted drug delivery



Nonspecific



Targeted to the specific part to treat in the body

Type and property of various targeting molecules for NPs. Modified from ref. [242].

Type	MW (kDa)	Diameter (nm)	Features
Monoclonal antibodies			
Whole antibodies	150	15–20	High affinity, divalent, many clinically approved examples, contains biologically active constant (Fc) region, long circulation
Engineered fragments (monovalent)			
ScFv	25	3–5	Lowered affinity, rapid clearance from circulation, renal retention, reduced stability, reduced immunogenicity
Fab'	50	5–10	Can be produced genetically or enzymatically by cleavage of monoclonal antibodies
Nanobody	15	2–3	Smallest antigen-binding fragment, single domain, can bind cryptic epitopes
Engineered fragments (divalent)			
F(ab') ₂	100	10–15	Improved affinity, can be engineered to a variety of sizes and arrangements of protein domains
Diabodies	50–80	5–10	Mono-specific or bi-specific dimer of ScFv
Minibodies	80	10	Can be produced genetically
Aptamers			
RNA/DNA	10–30	2–3	Rapid clearance, automated chemical synthesis, susceptible to nucleases without chemical modification
Receptor-ligands			
Whole proteins	30–150	Variable	Produced using recombinant DNA technologies, can be biologically active, susceptible to proteases
Peptides	0.5–10	Variable	Facile synthesis and modification, diverse libraries and screening technologies, susceptible to peptidases, renal retention
Small molecules	0.1–1.0	0.5–2.0	Chemical synthesis, simple modification and coupling chemistries, can be biologically active, highly variable affinities

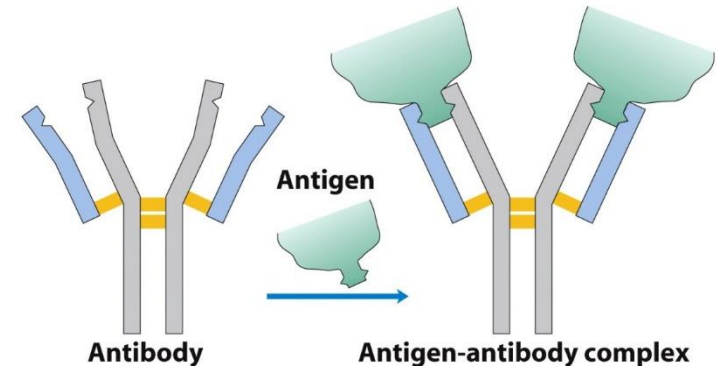
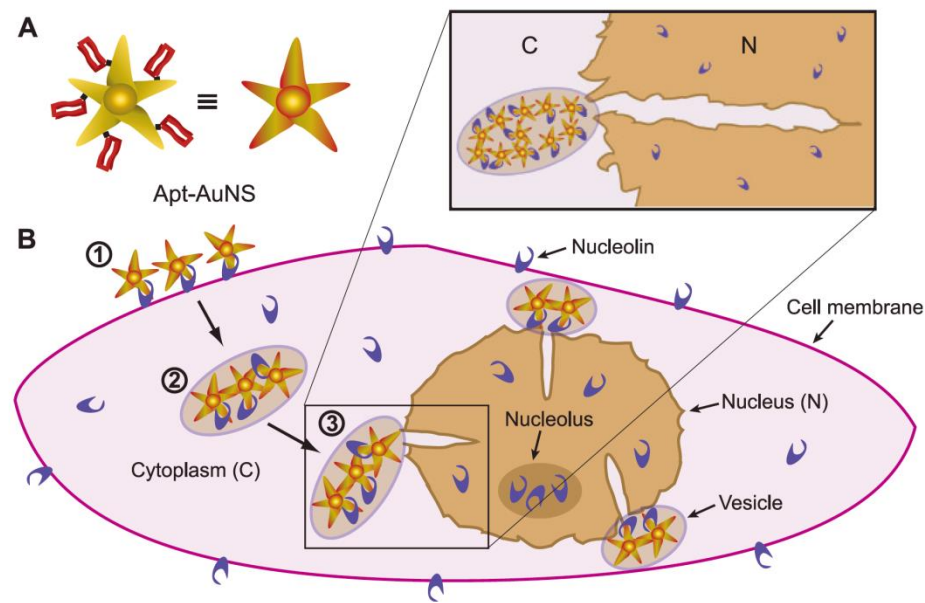
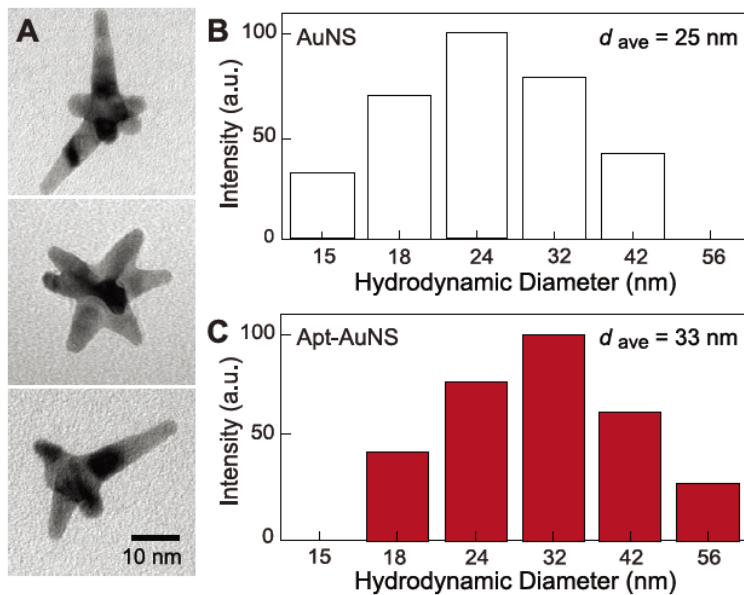
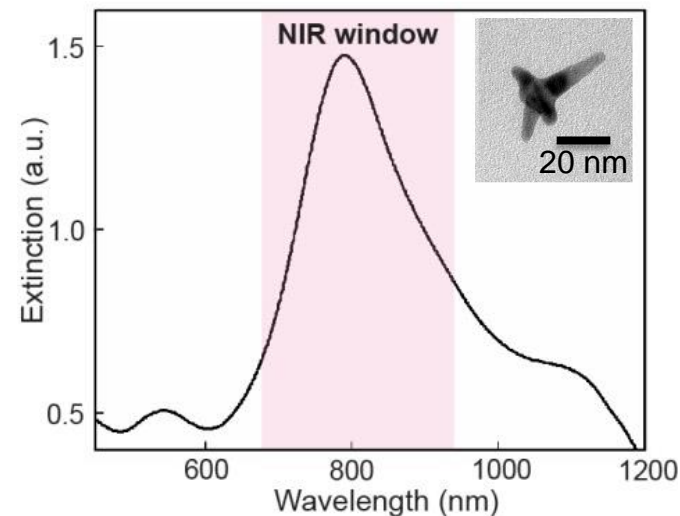
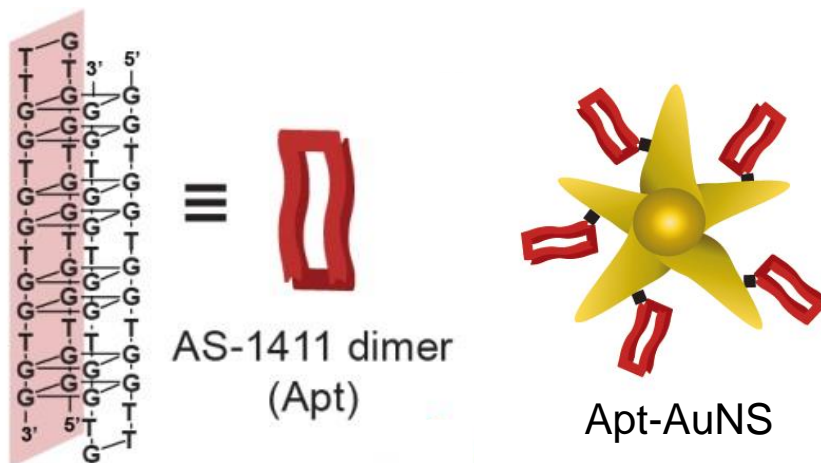


Figure 5-22
Lehninger Principles of Biochemistry, Fifth Edition
© 2008 W.H. Freeman and Company

Gold nanostar based DDS-release of drug



Cell image

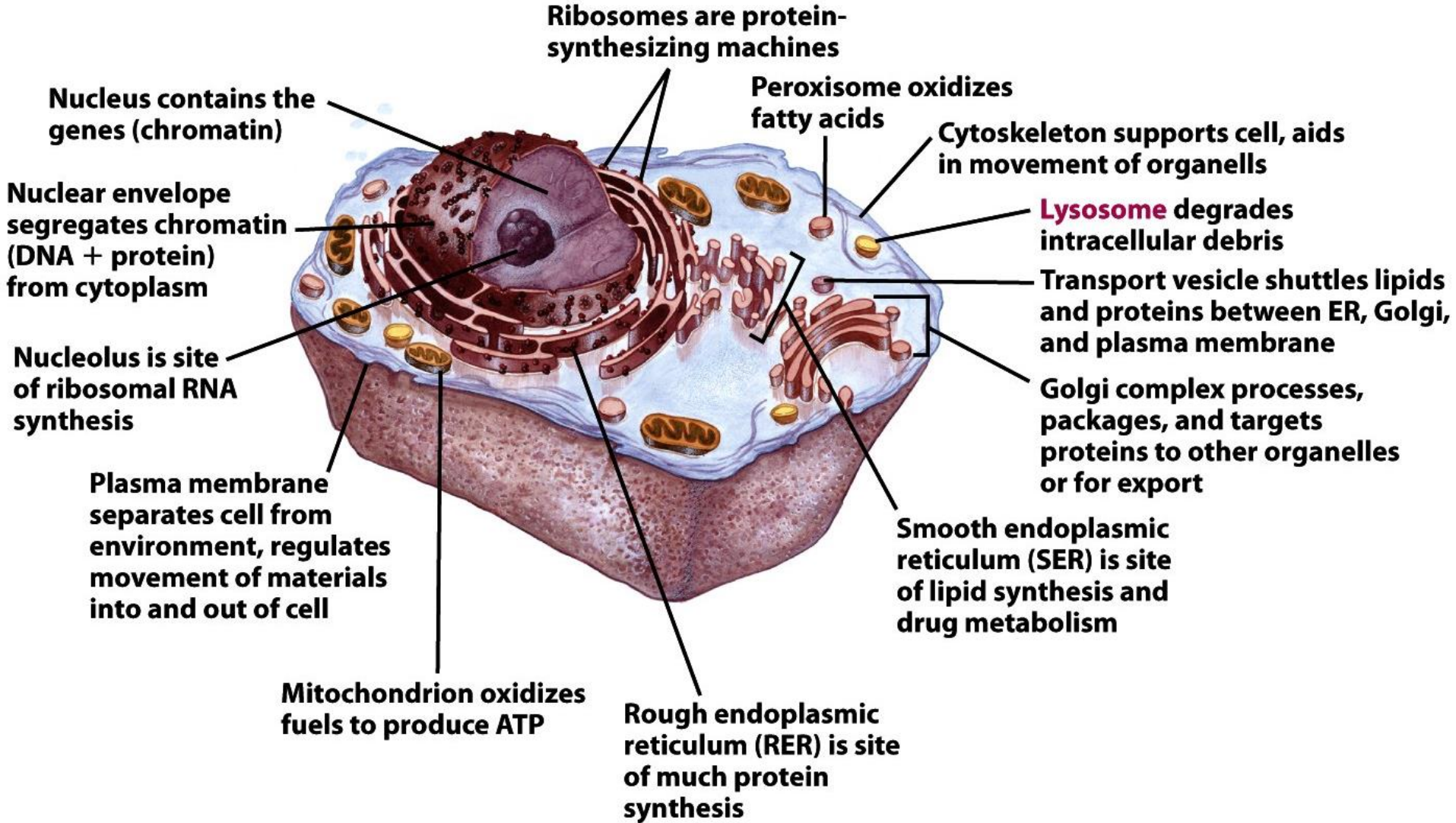
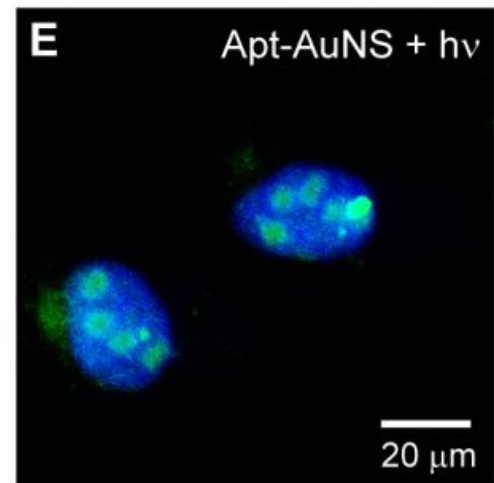
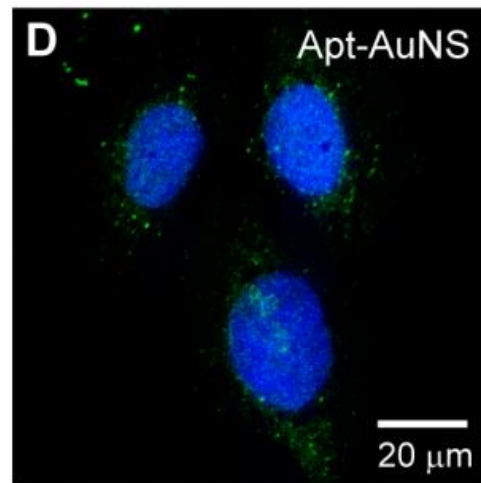
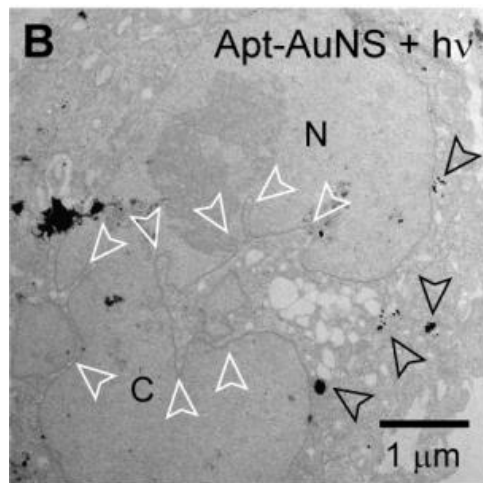
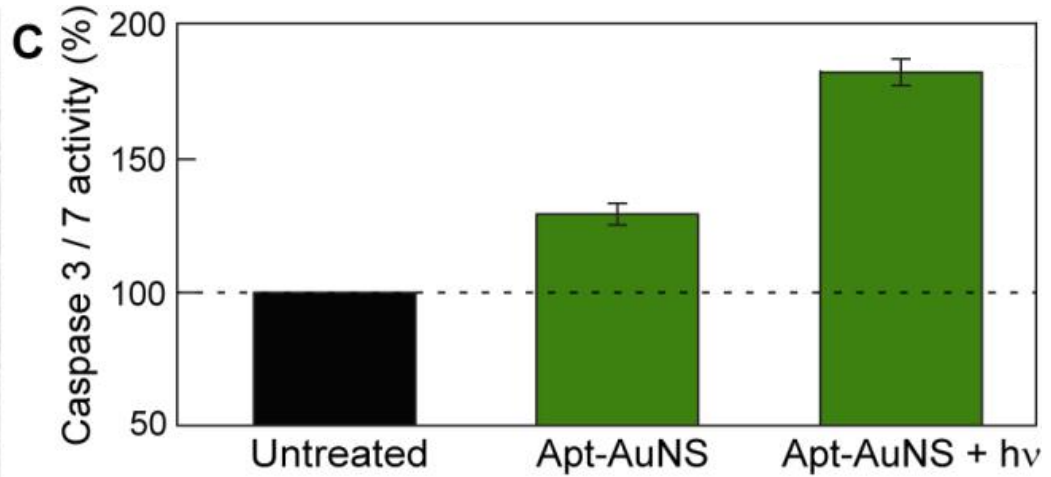
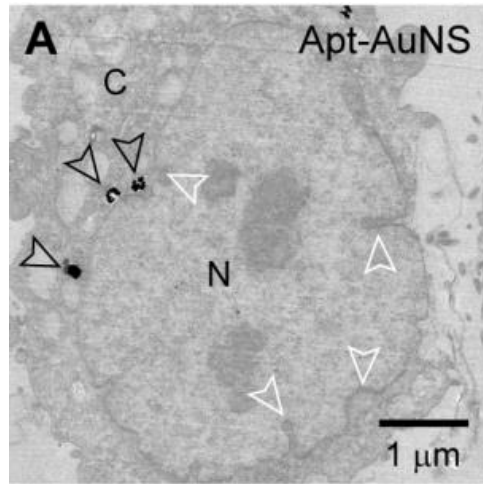
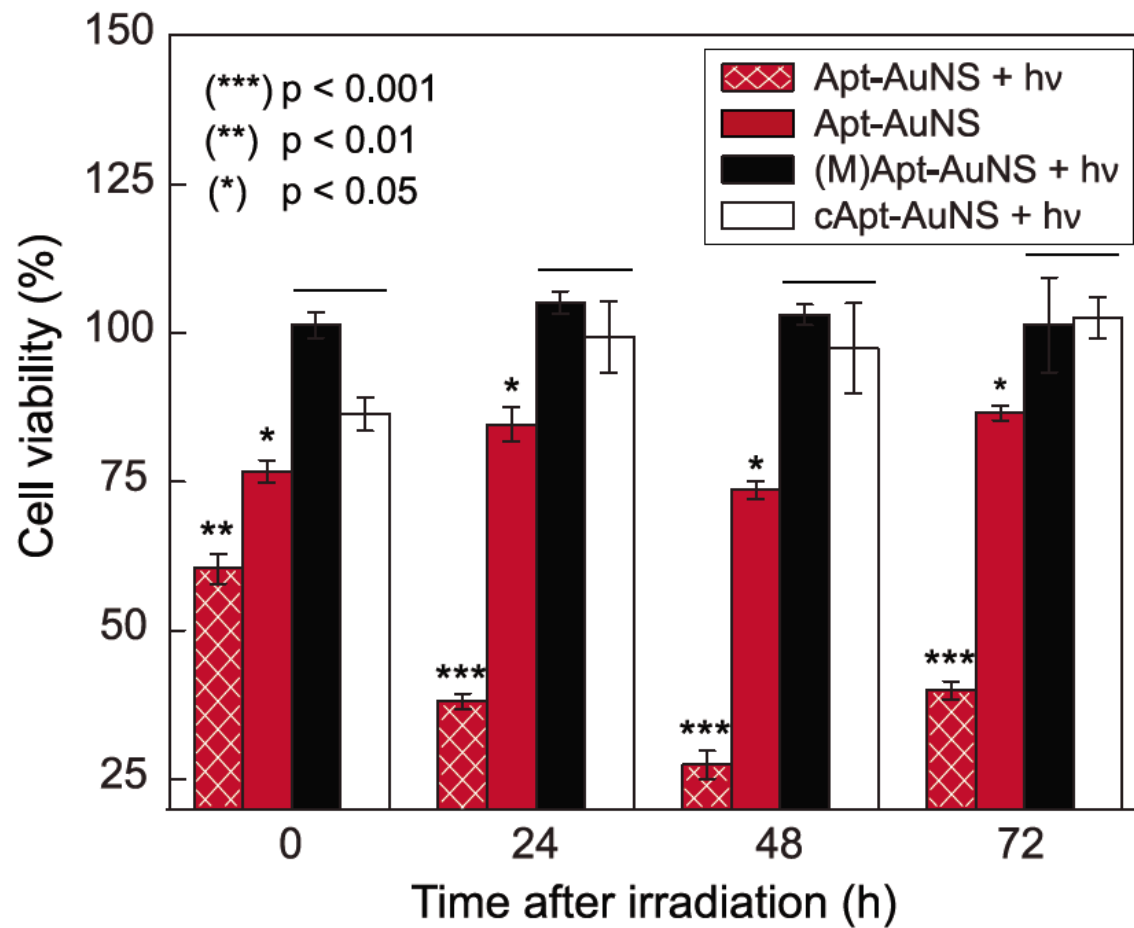


Figure 1-7a
Lehninger Principles of Biochemistry, Fifth Edition
© 2008 W.H. Freeman and Company

Gold nanostar based cancer treatment



Cell viability assay

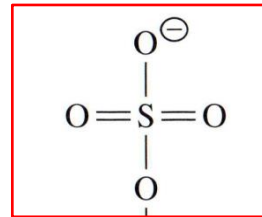


Self assembly of surfactants

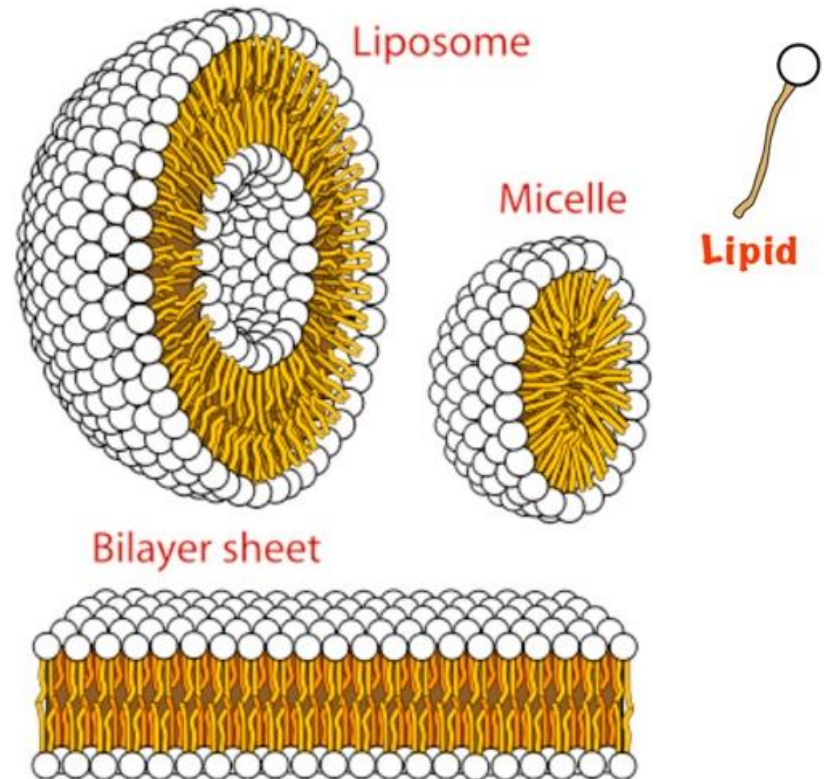
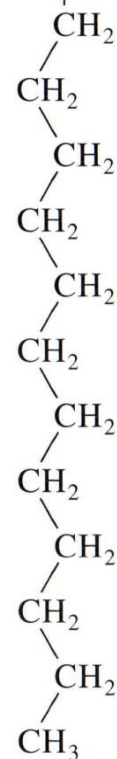
Synthetic detergent commonly used in biochemistry: Sodium dodecyl sulfate (SDS)

Polar "sulfate" group

Na⁺

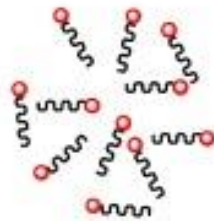
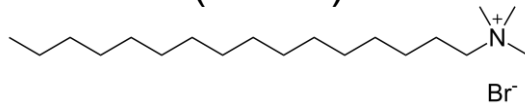


Nonpolar
“(CH₂)₁₂” chain

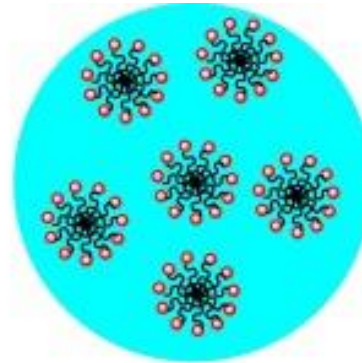


Synthesis of mesoporous materials using surfactants

Cetyl trimethylammonium bromide (CTAB)



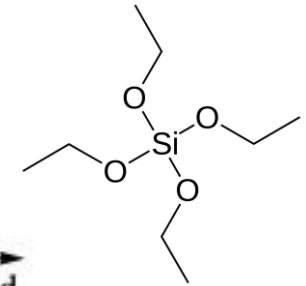
Water



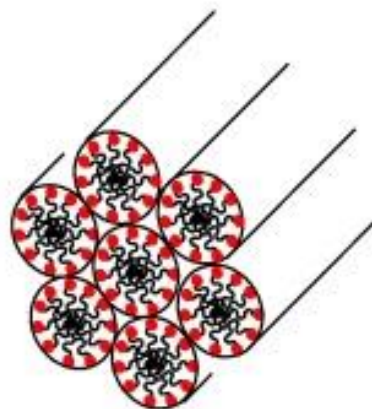
Micelles

Surfactant (e.g. CTAB)

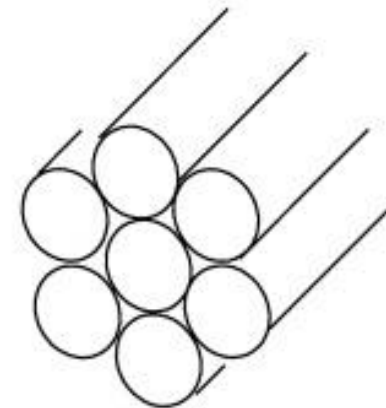
Tetraethyl orthosilicate (TEOS)



TEOS
Hydrolysis and condensation



Template removal



Mesoporous silica: e.g. MCM-41

Synthesis of AuNR/mSiO₂ core/shell structure

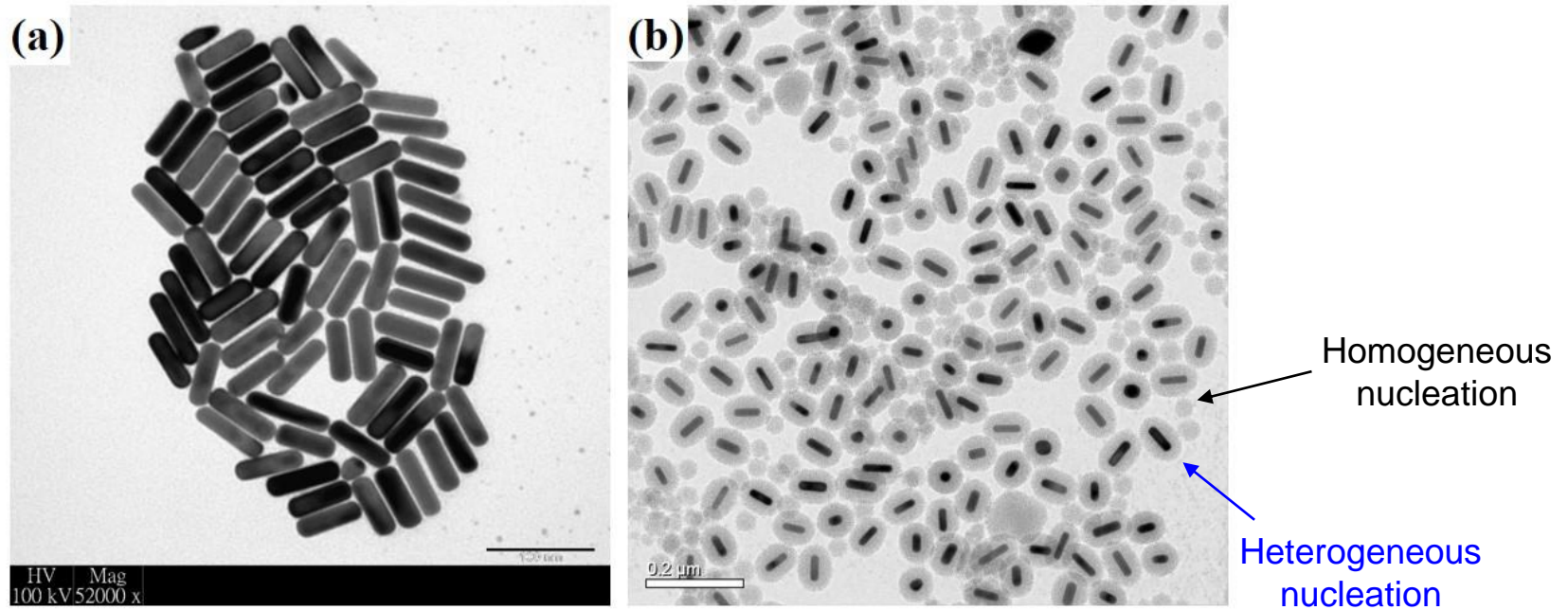
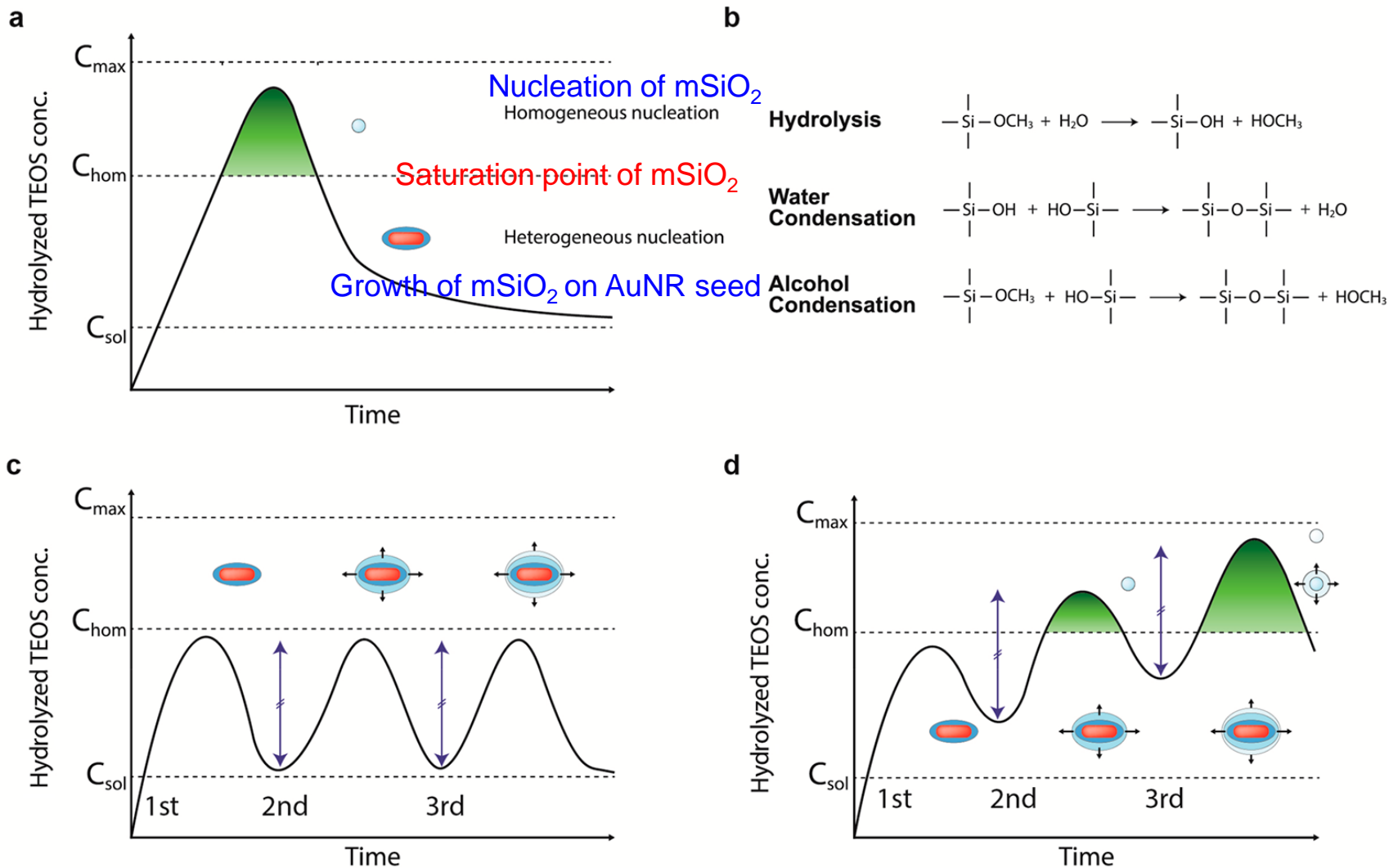


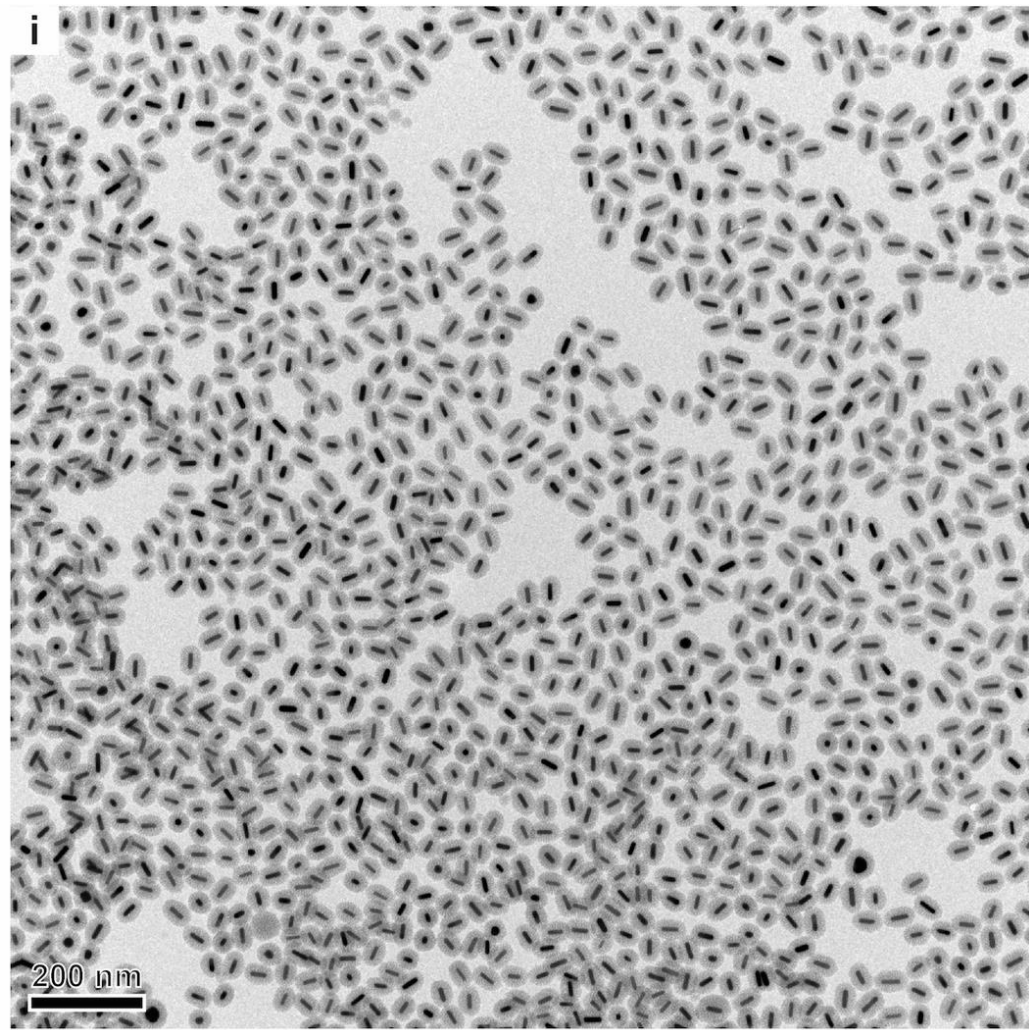
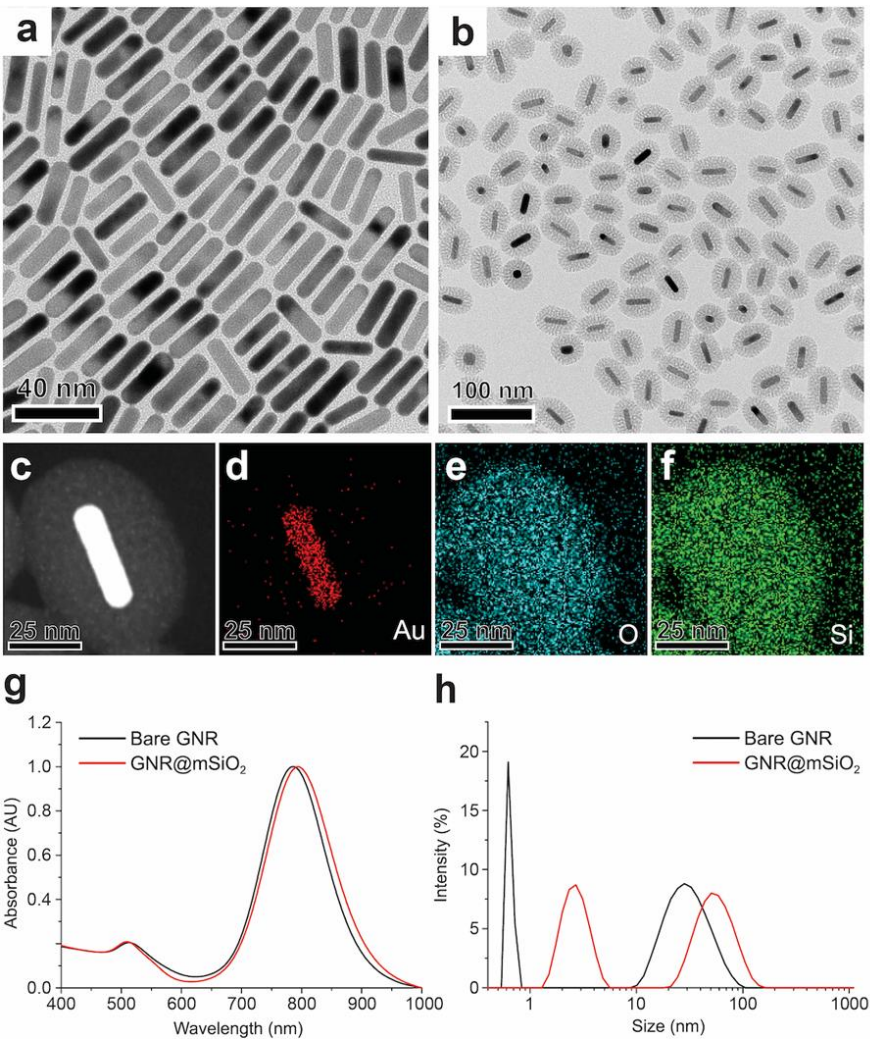
Fig. 1. TEM images of (a) AuNRs and (b) AuNR@mSiO₂ nanoparticles.

- Homogeneous nucleation: nuclei that are spontaneously generated and grow irreversibly to form a new phase.
- Heterogeneous nucleation: nuclei are formed on alien surfaces or particles, or pre-existing nuclei in the old phase.

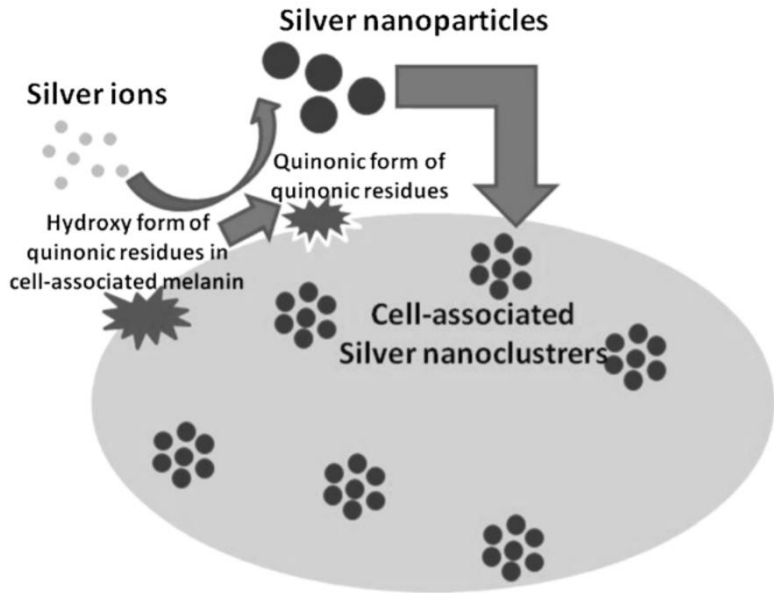
AuNR/mSiO₂ core/shell structure



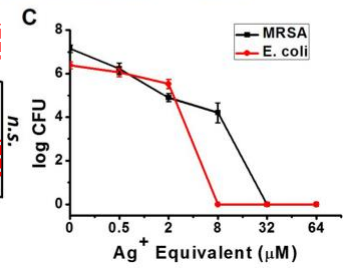
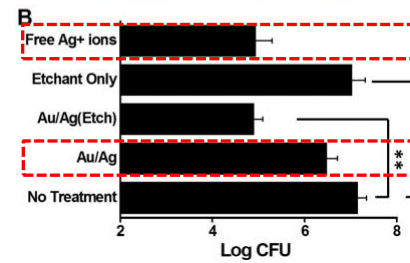
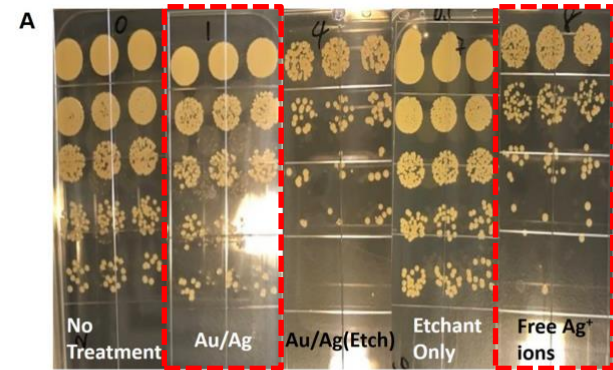
Highly monodisperse AuNR/mSiO₂



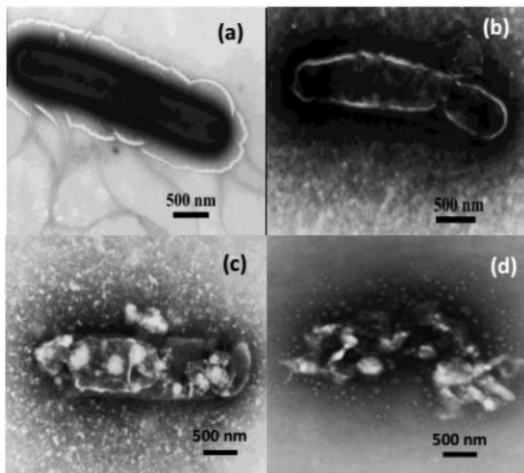
Antibacterial property of AgNP



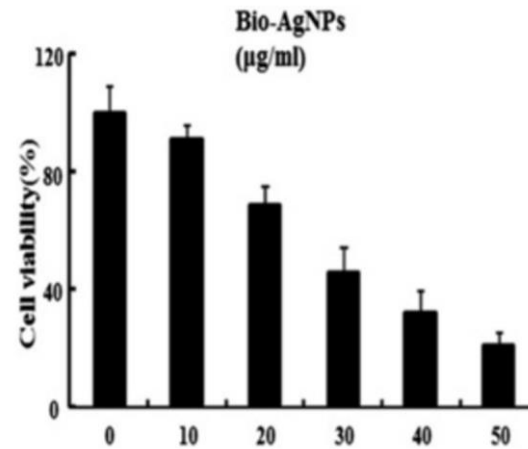
AMB Express, 2013, 3-32



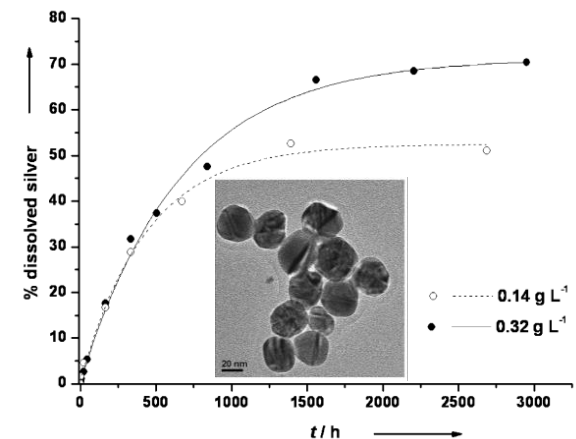
ACS Nano, 2018, 12, 5615-5625



Text. Res. J., 2017, 1377-1386

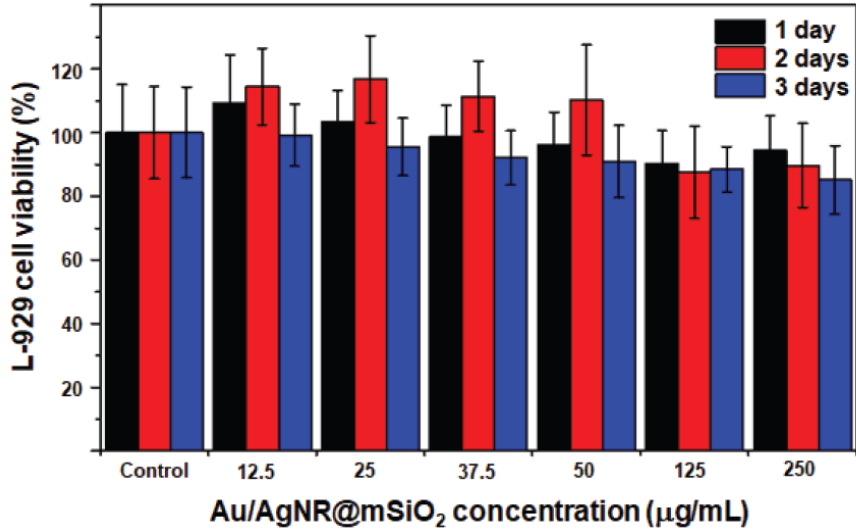
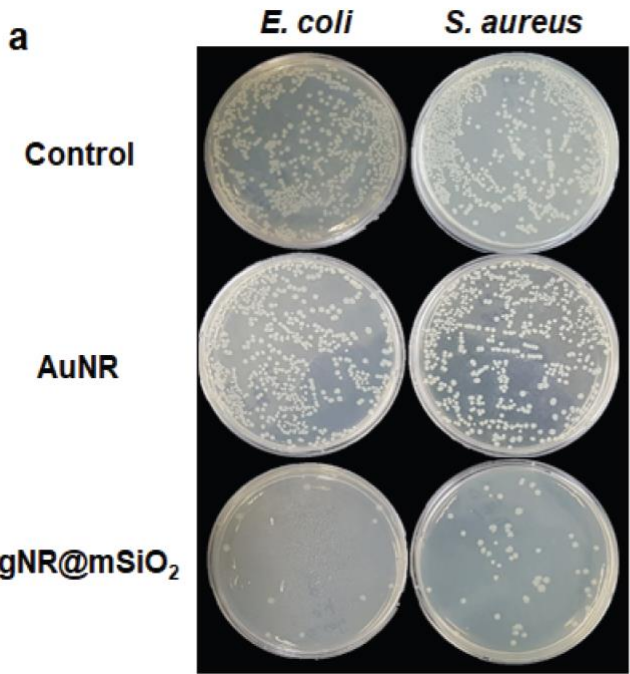
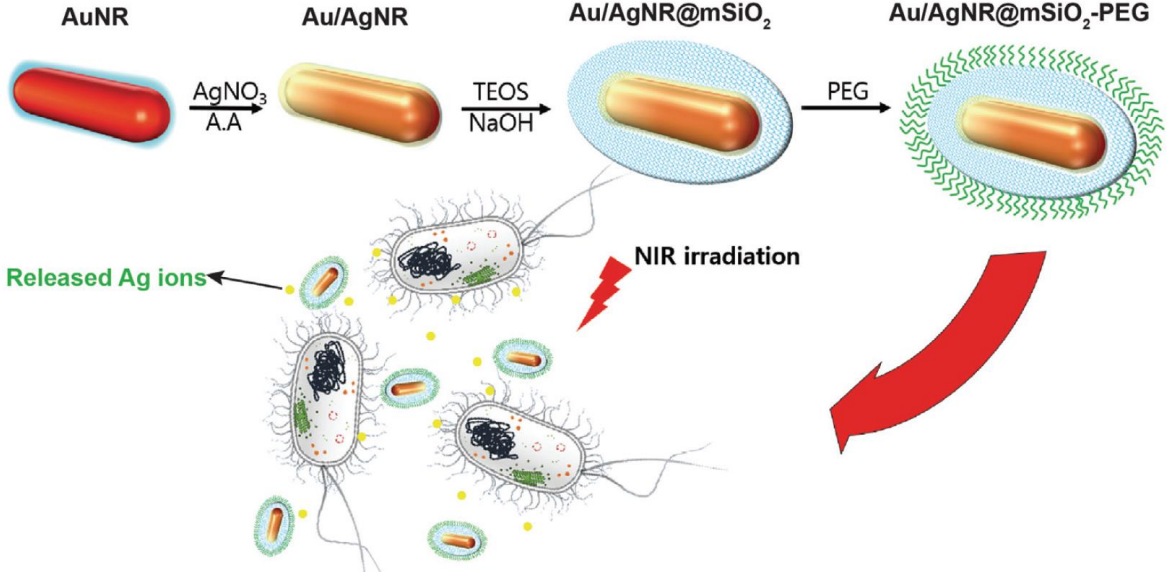


Nanoscale Res. Lett, 2014, 9, 459

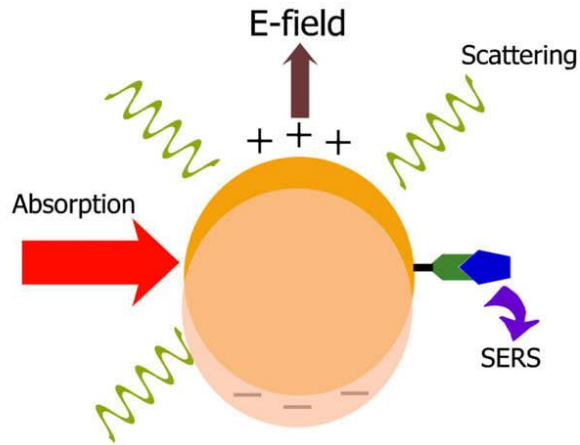


Angew. Chem, 2013, 52, 1636-1653

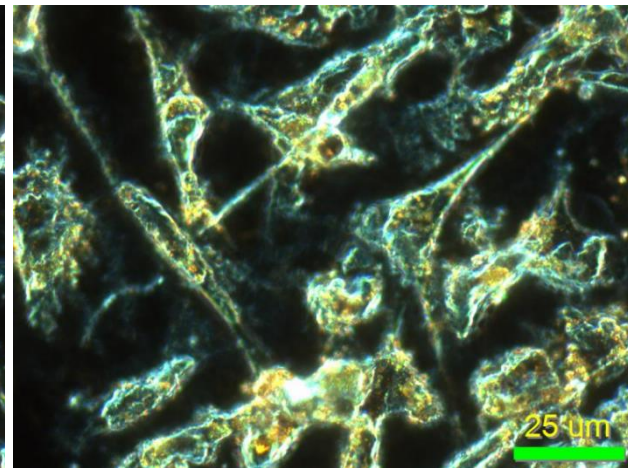
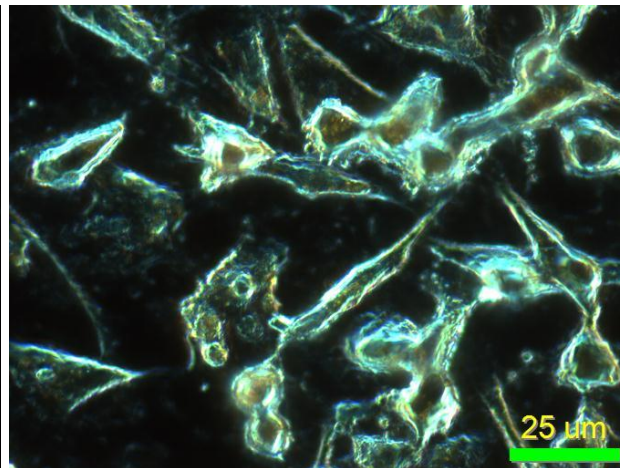
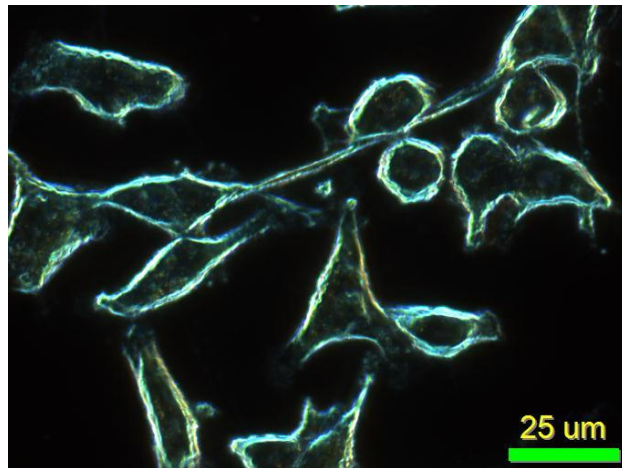
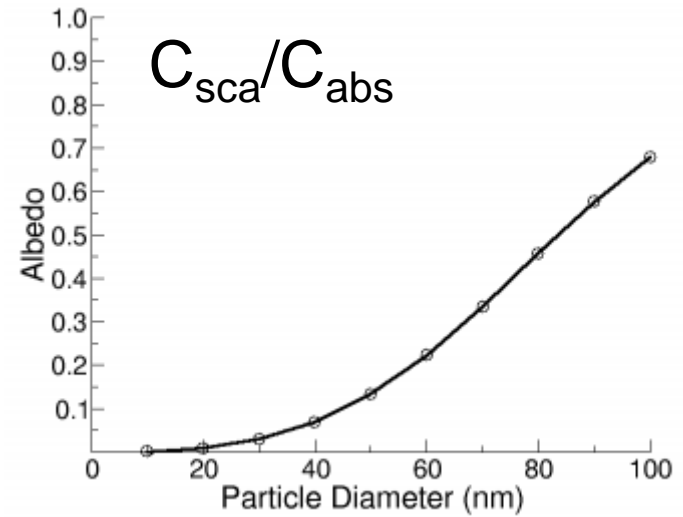
Gold/silver nanorod for anti-bacterial applications



Dark field imaging of cancer cells



Chem. Phys. Lett. **2010**, 487, 153

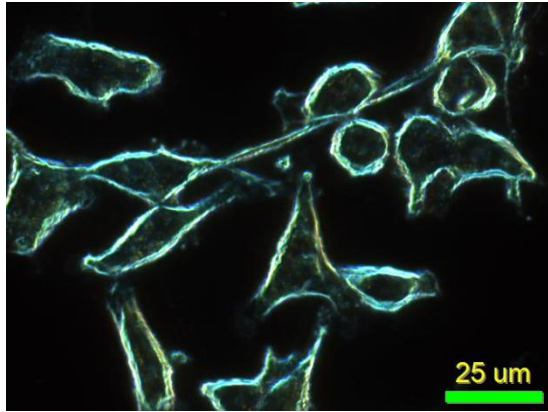


13 nm AuNP
x 20 times concentrated

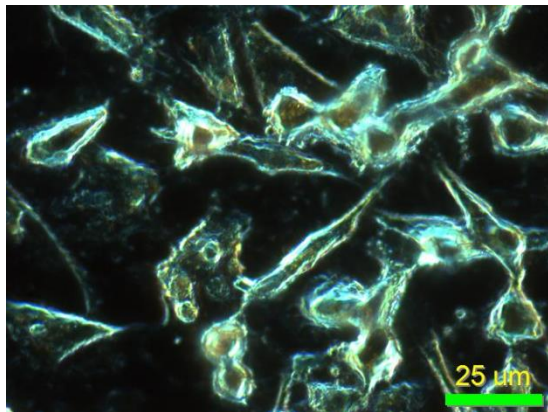
50 nm AuNP

100 nm AuNP

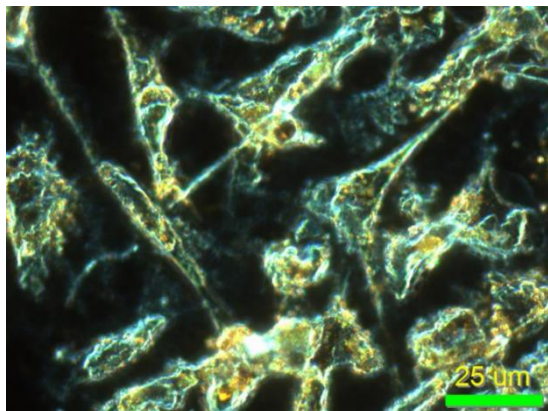
Dark field imaging of NPs



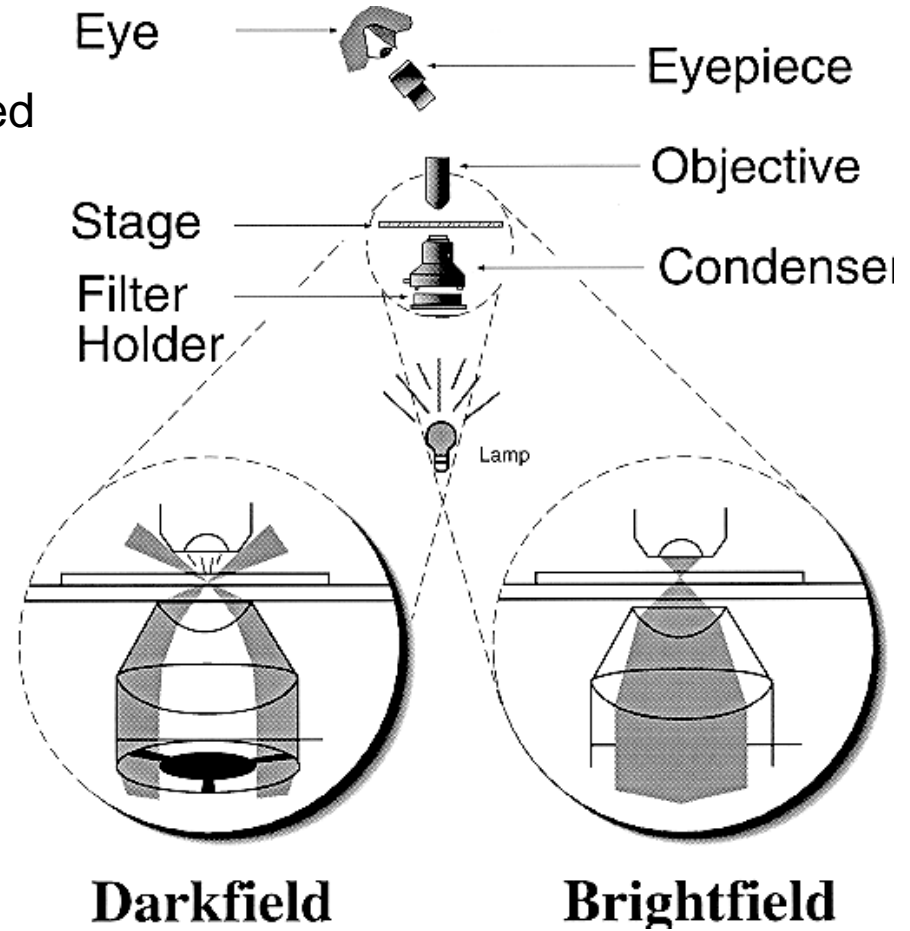
13 nm AuNP
x 20 concentrated



50 nm AuNP

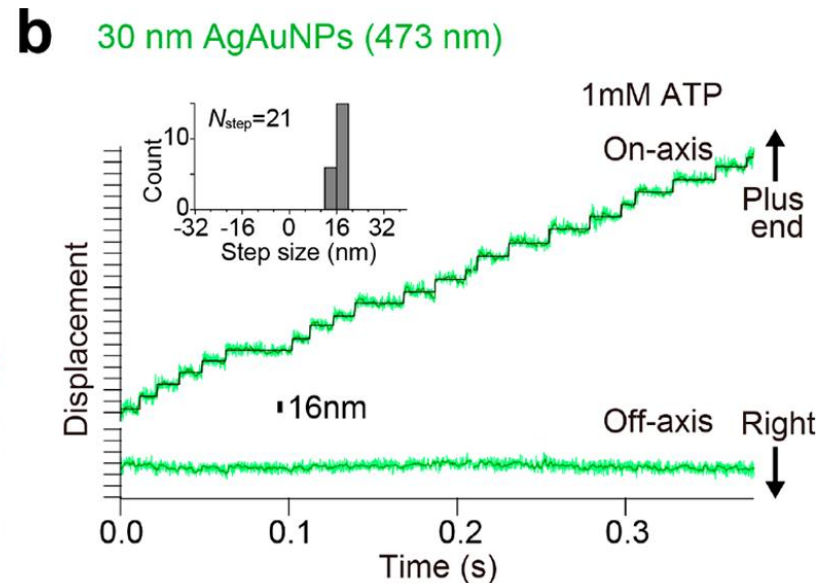
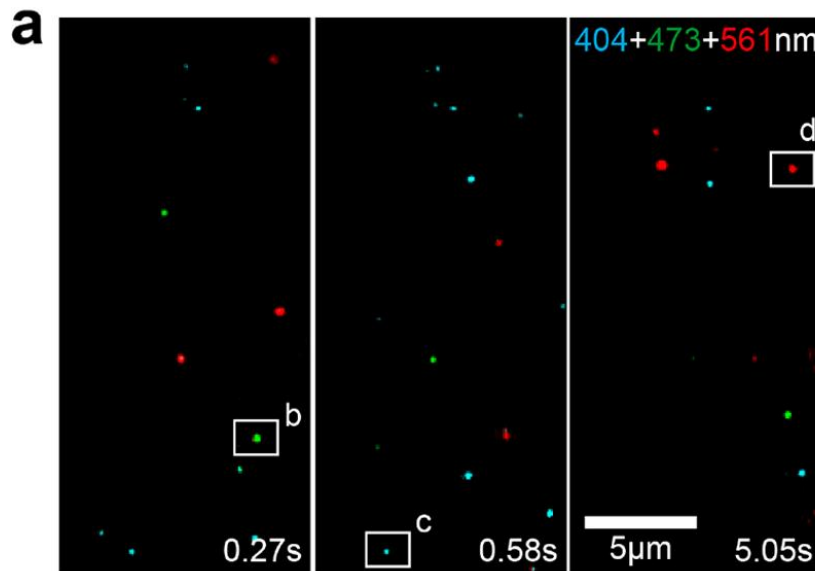
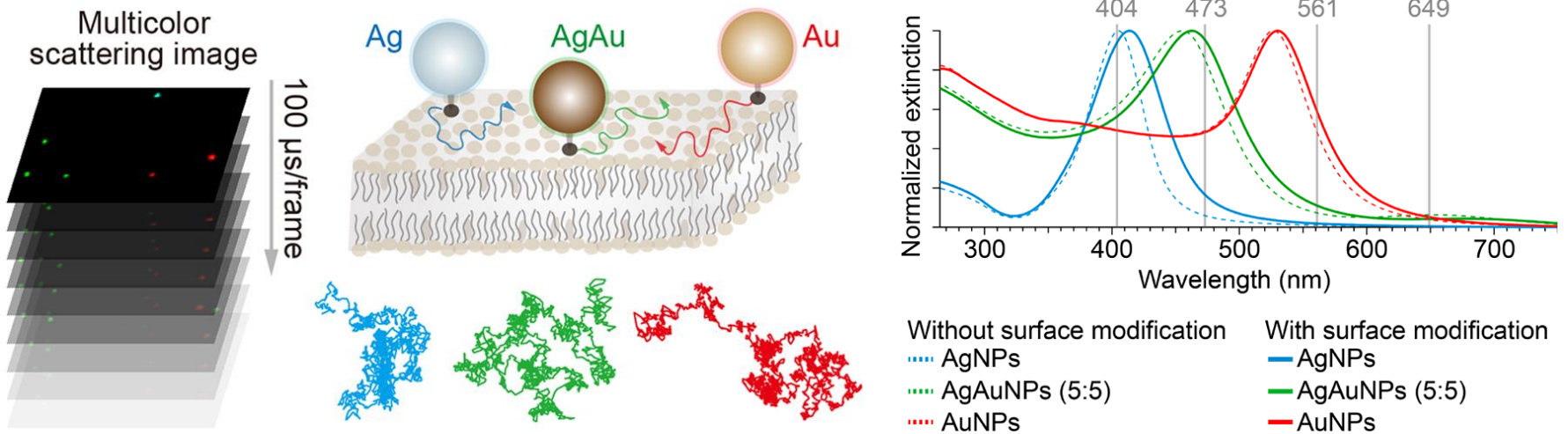


100 nm AuNP

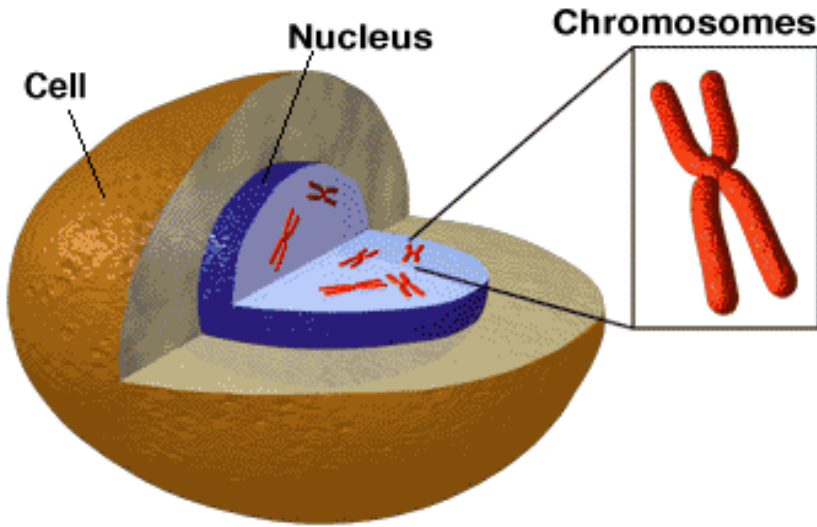


Imaging single molecule movement

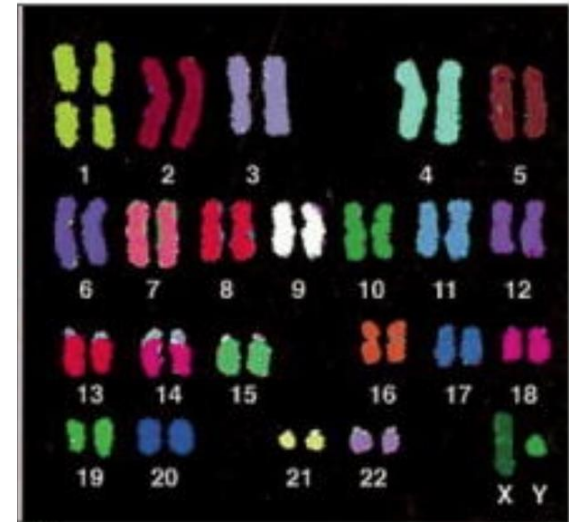
Multicolor tracking of single biomolecules using plasmonic NPs



Dimension of genomic DNA molecules

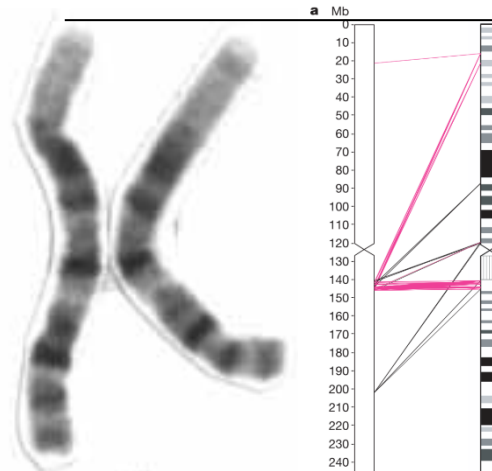


Iowa Public Television, 2004

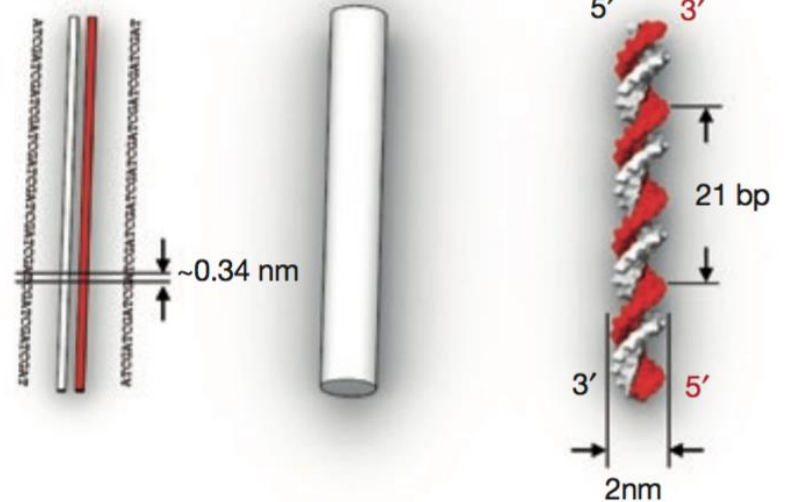


Science, 1996, 273, 494–497

Chromosome no.1
 2.5×10^8 bp = 85 mm
 (0.34 nm/1 bp)



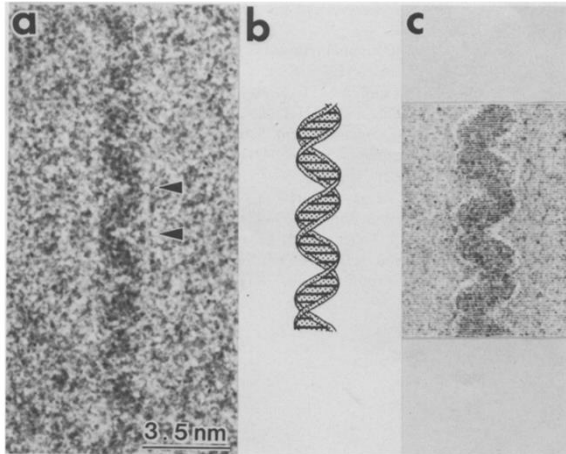
Nature. 2006, 441, 7091, 315–321



Nat. Methods, 2011, 8, 221–229

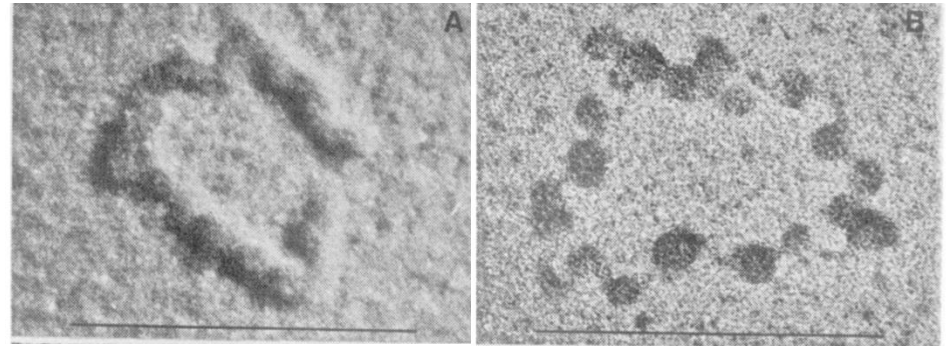
Transmission Electron Microscopy (TEM) based DNA visualization

Double helical B-form DNA structure



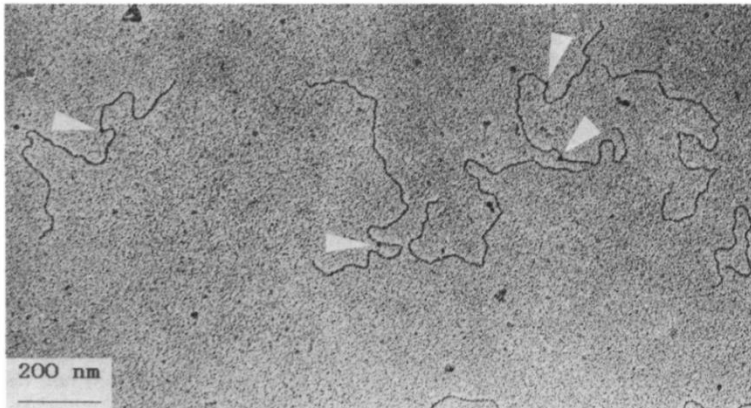
Ultramicroscopy, 1981, 7, 189

DNA-protein complex (chromatin)



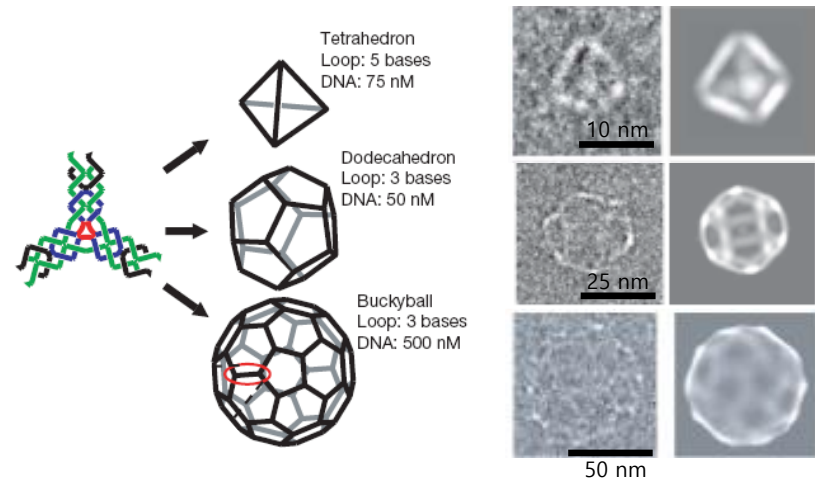
Science, 1975, 187, 1202

Sequence specific mapping



Nucl. Acids Res., 1994, 22, 5218

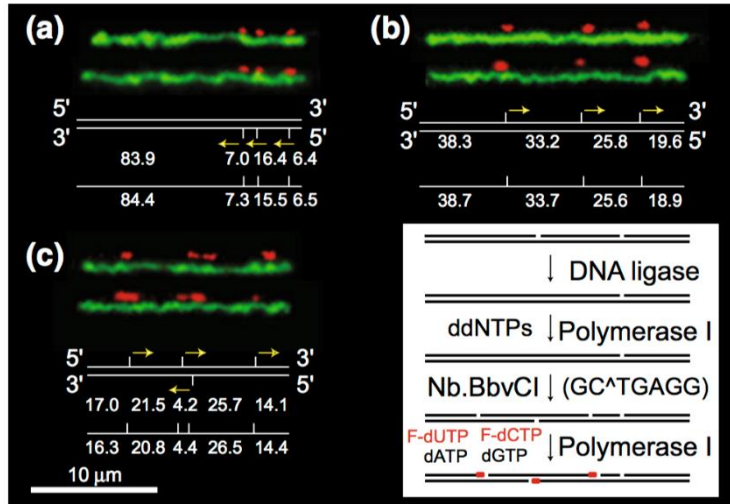
DNA nanostructures



Nature, 2008, 452, 198

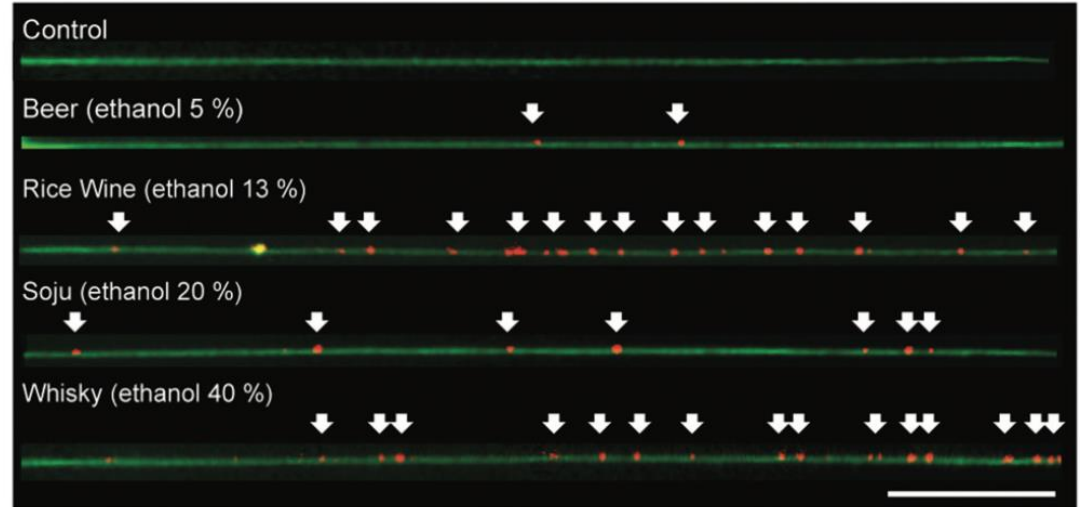
Fluorescence imaging for genomic DNA studies

High-throughput sequencing



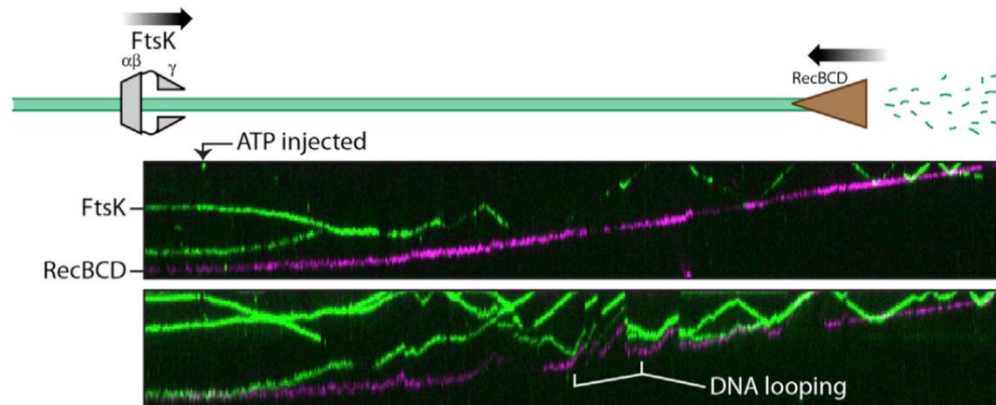
PNAS, 2007, 104, 2673-2678

DNA damages



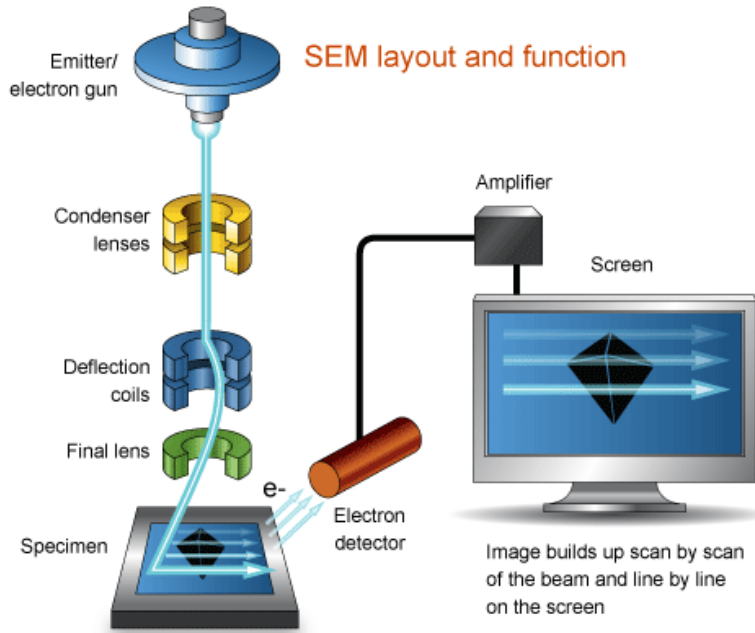
Analyst, 2016, 141, 4326-4331

DNA-protein interactions



Mol. Cell., 2014, 54, 832-843

Scanning Electron Microscopy (SEM) imaging

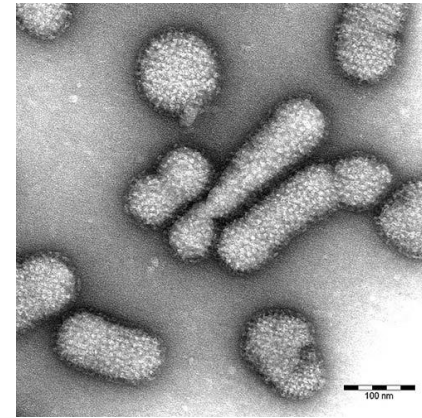


<http://www.ammrf.org.au/>

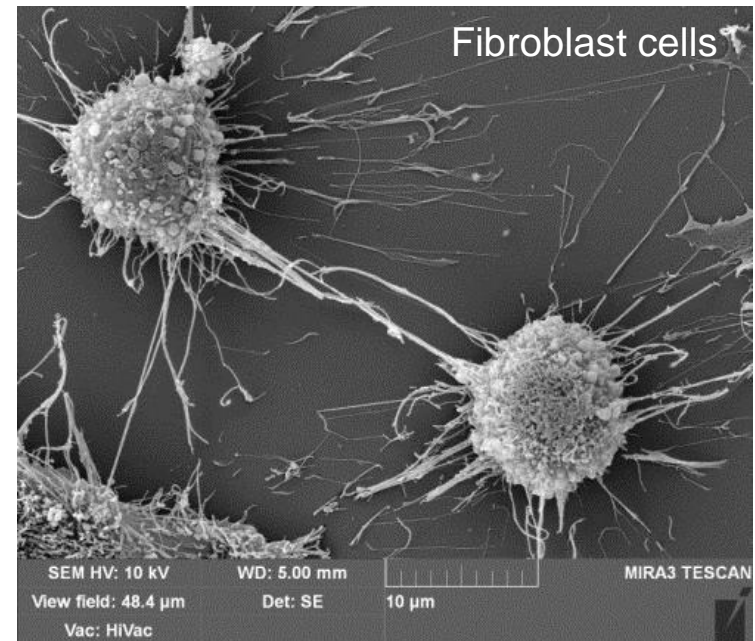


The James Hutton Institute

Avian flu virus

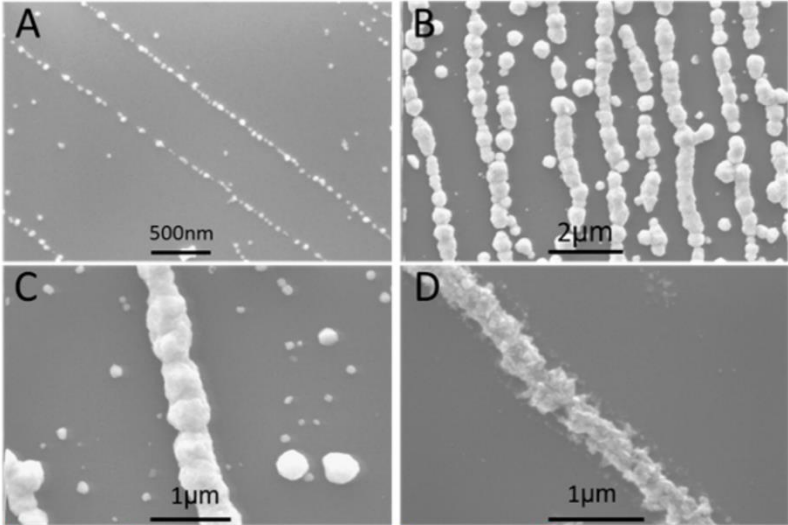


UAF Center for Distance Education

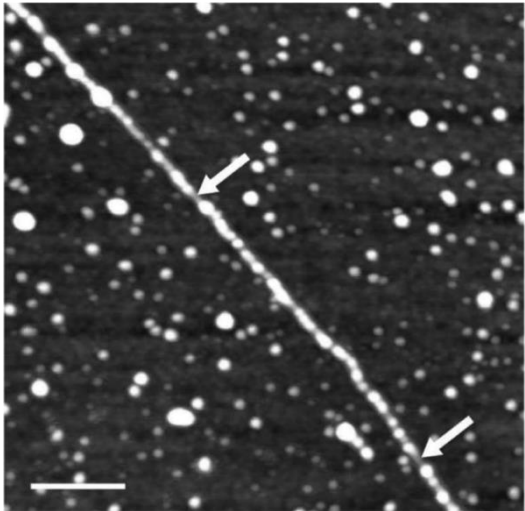


www.tescan.com

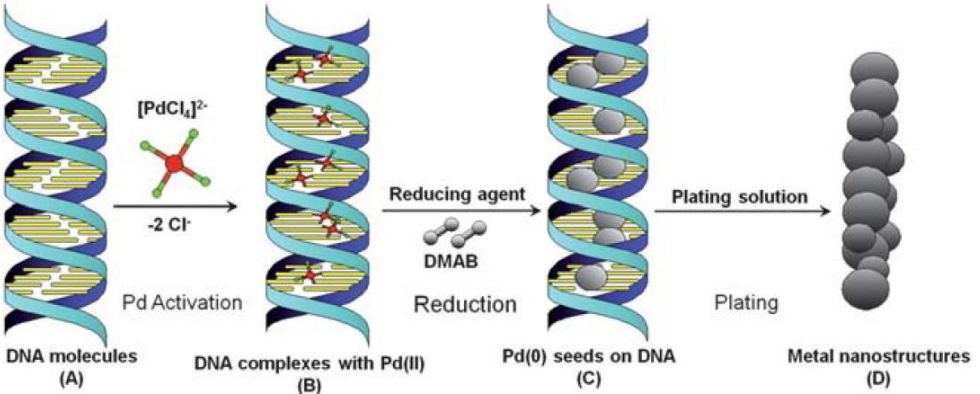
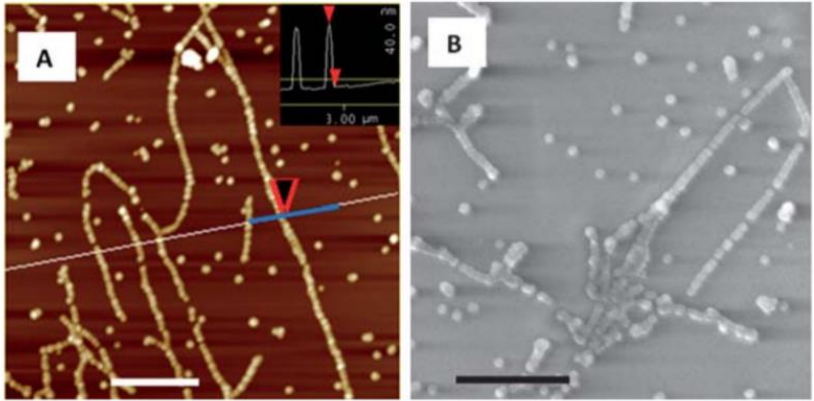
Limitation of current DNA metallization strategies



Langmuir, **2013**, 29, 11176–11184



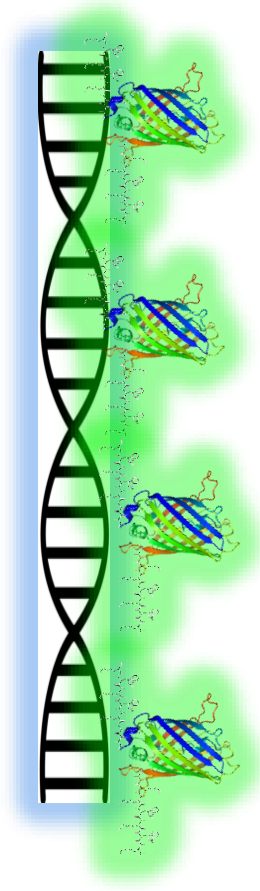
Langmuir, **2010**, 26(3), 2068–2075



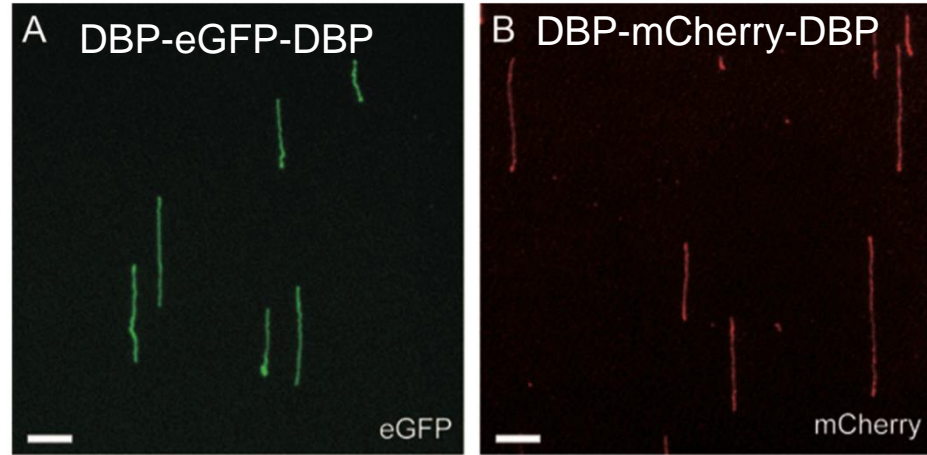
J. Mater. Chem., **2011**, 21, 12126–12131

DNA binding peptide (DBP) based DNA imaging

KWKWKKA-FP-AKKWKWK
FP: eGFP, mCherry

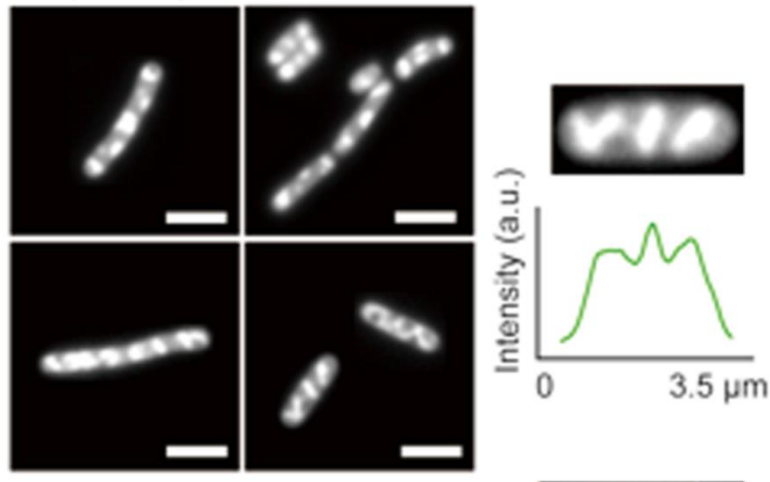


Design of DBP based fluorescence imaging

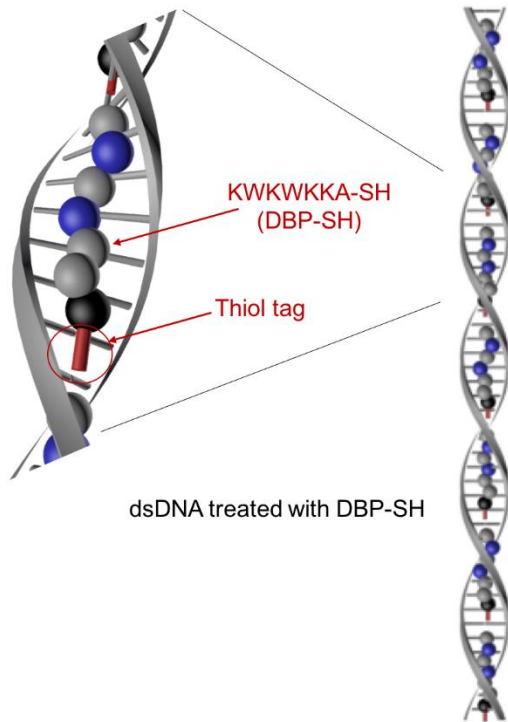


Imaging DNA of live *E. coli* cells

FP(eGFP)-DBP



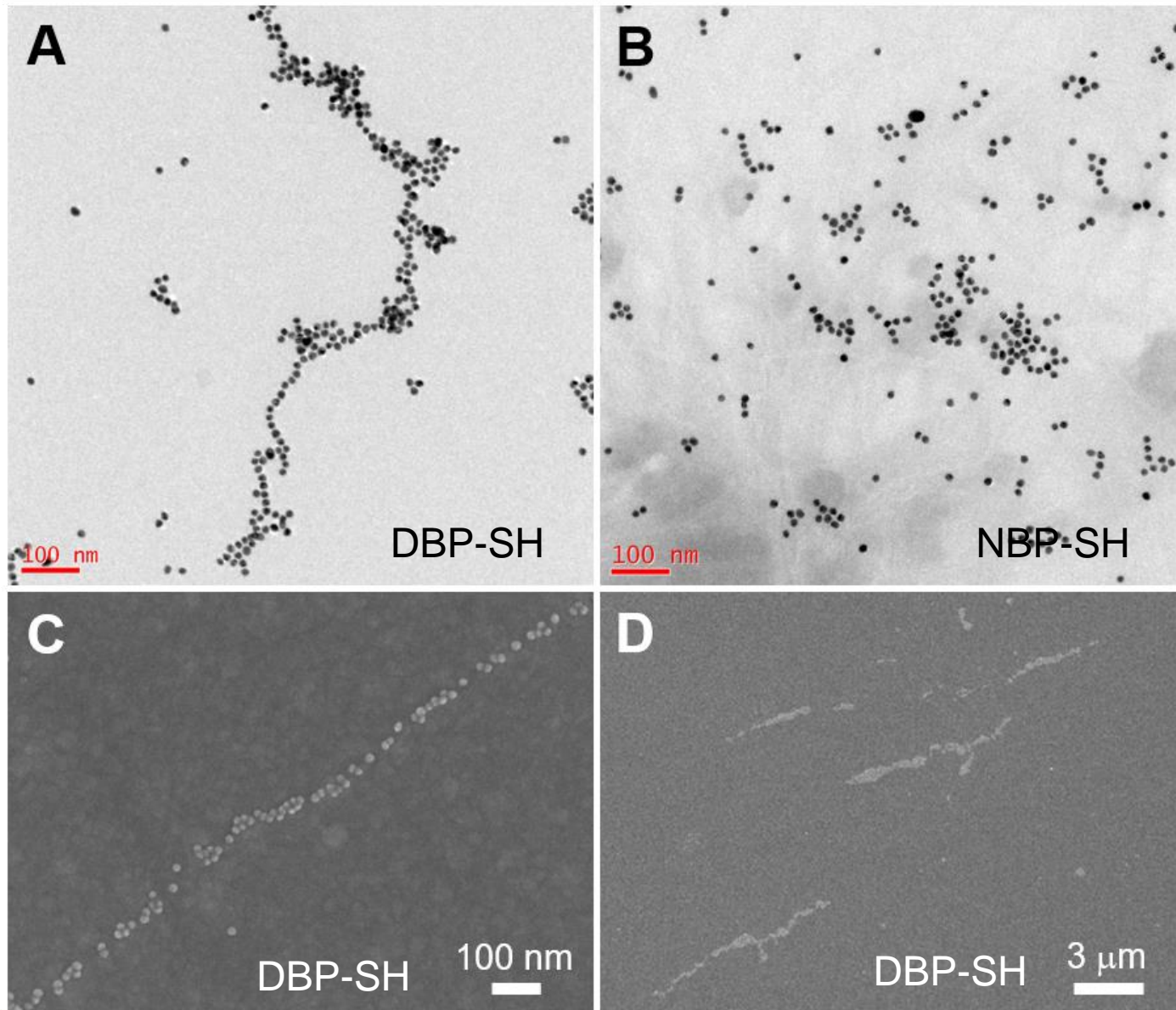
DNA metallization strategy



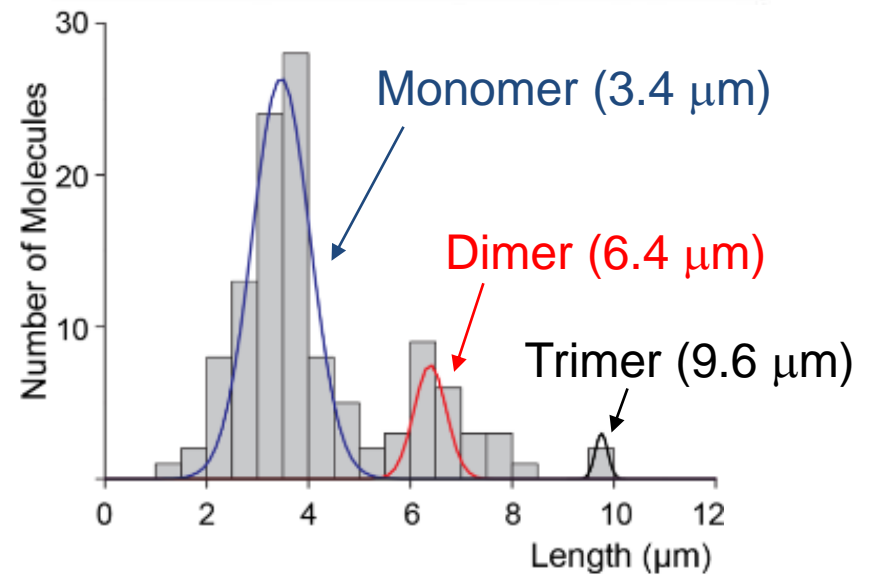
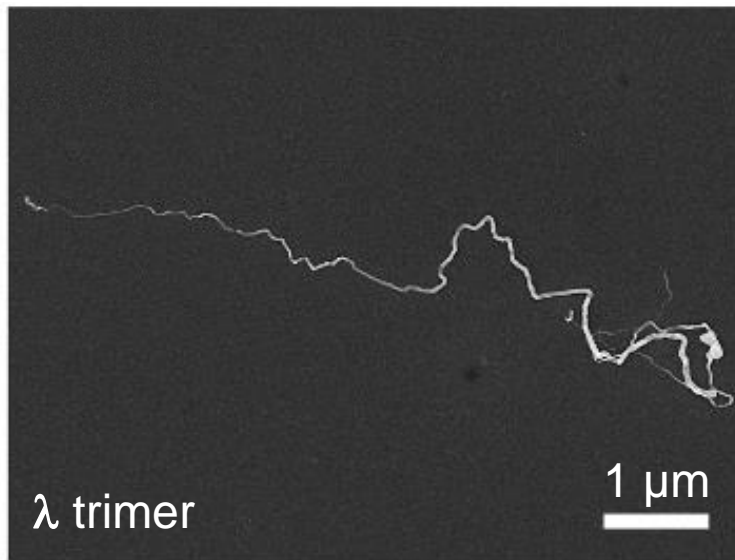
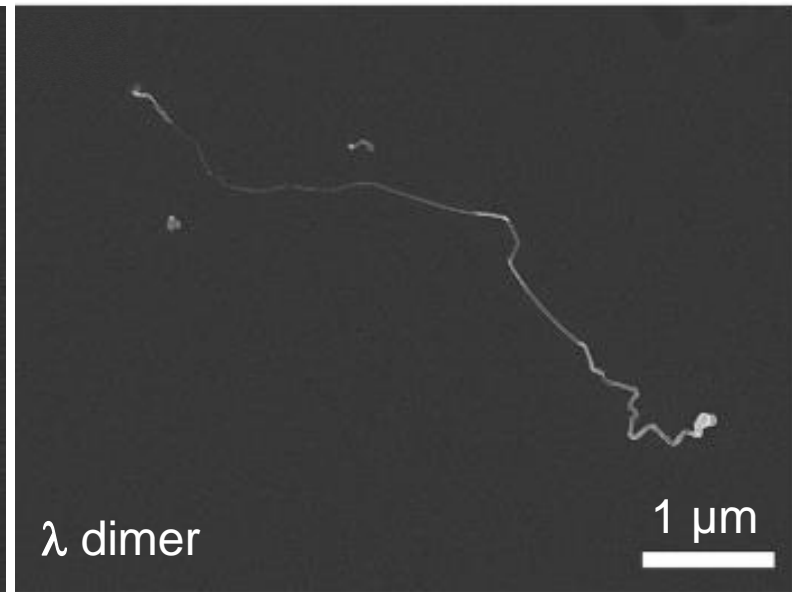
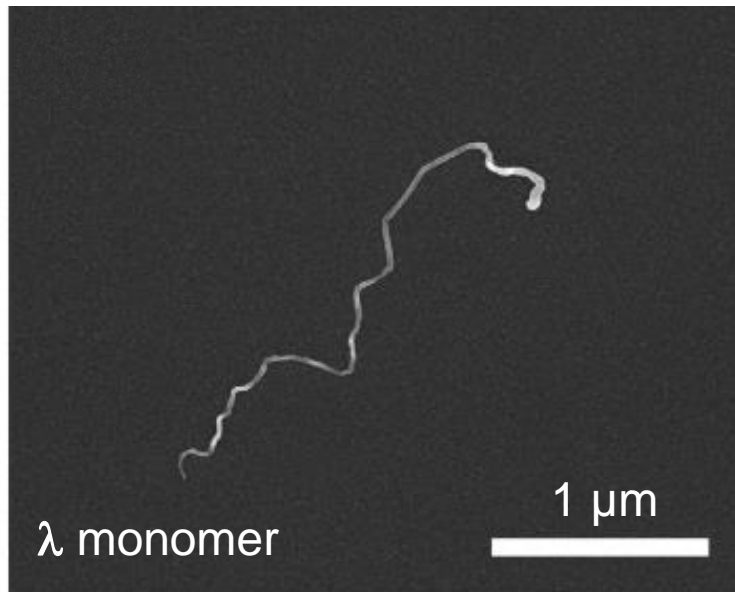
DNA with high density thiol groups

- Interaction with dsDNA and DNA binding peptide with thiol tag (DBP-SH) for the introduction of functional groups
- Covalent bonding of thiol group onto the gold surface and nanoparticle

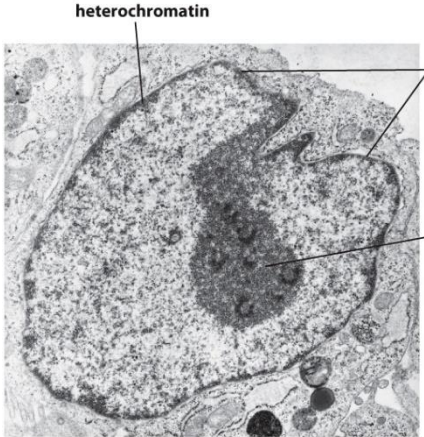
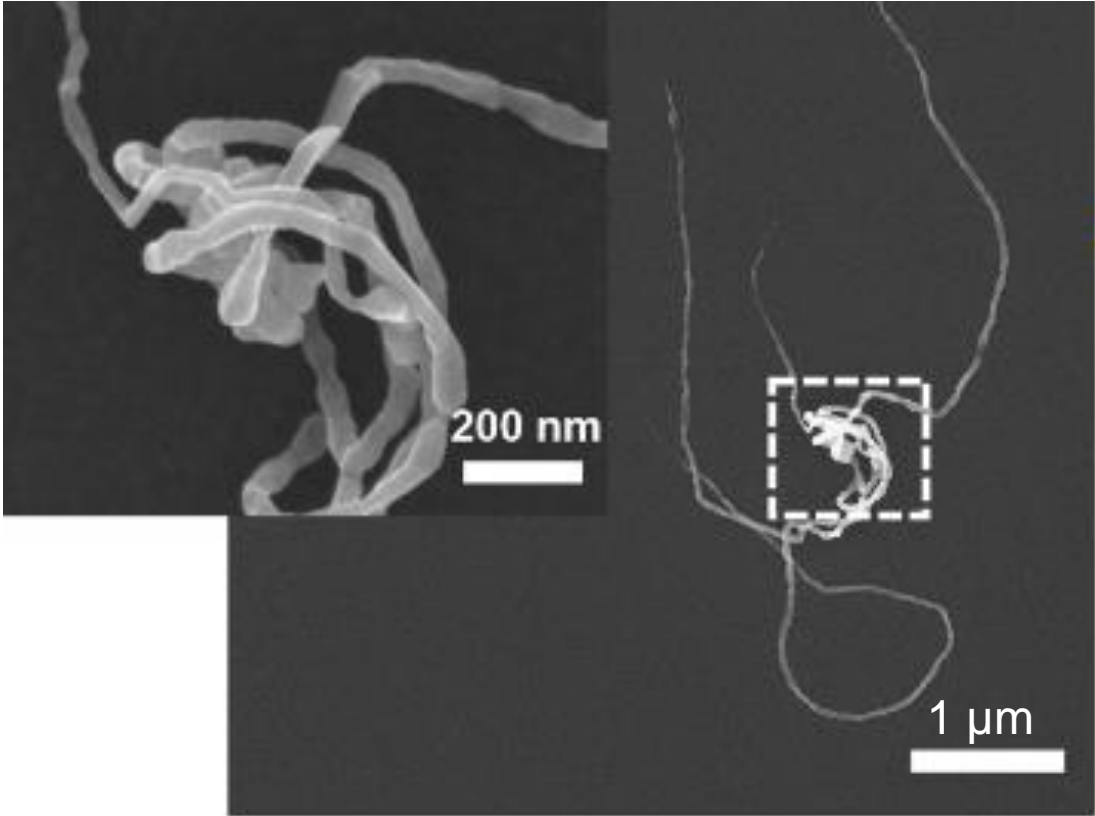
AuNP assembly on DBP-SH treated λ -DNA



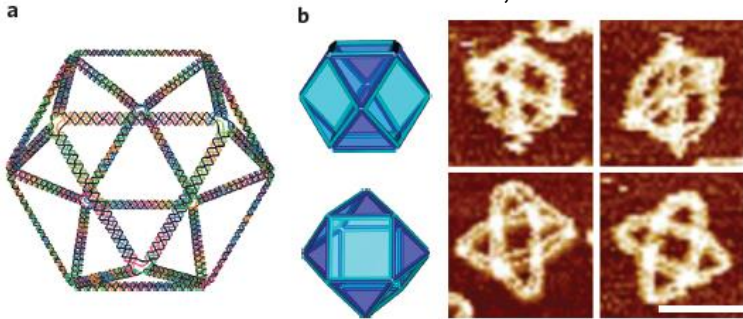
Au nanowires grown on λ -DNA templates



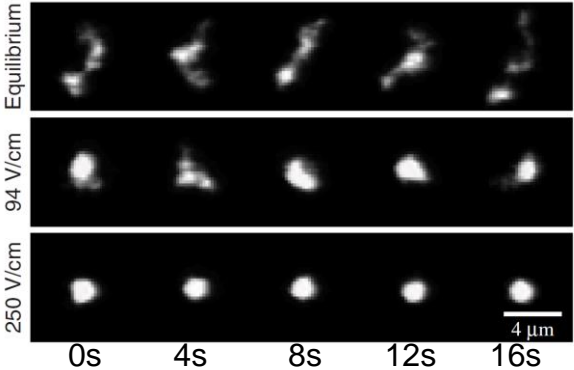
Imaging 3D entangled DNA structures



(A) *Mol. Biol. of the Cell*, 6th Ed.

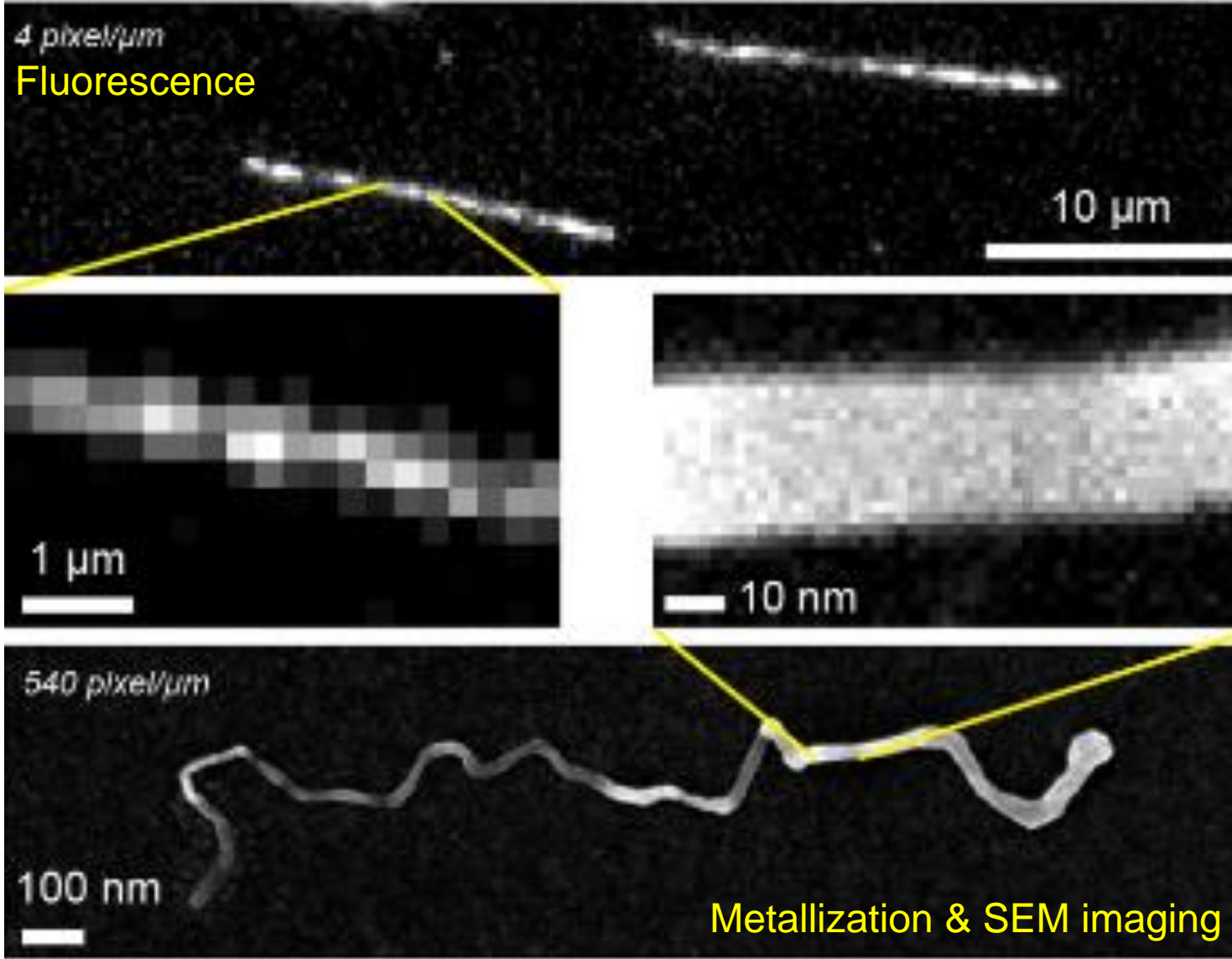


Nat. Nanotechnol., 2015, 10, 779



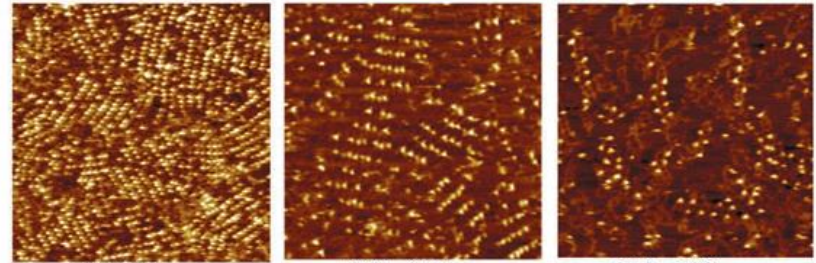
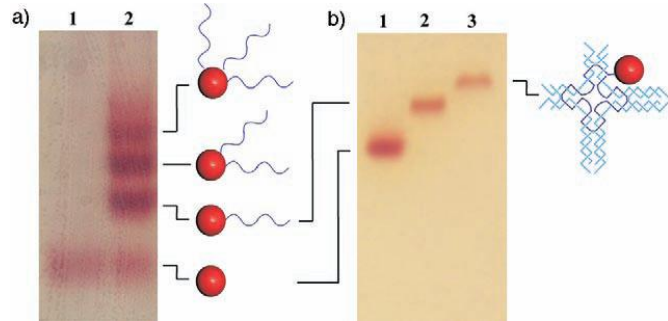
PNAS., 2011, 108 (39), 16153-16158

Comparison of fluorescence & SEM images of DNA



DNA nanostructures for NP patterning

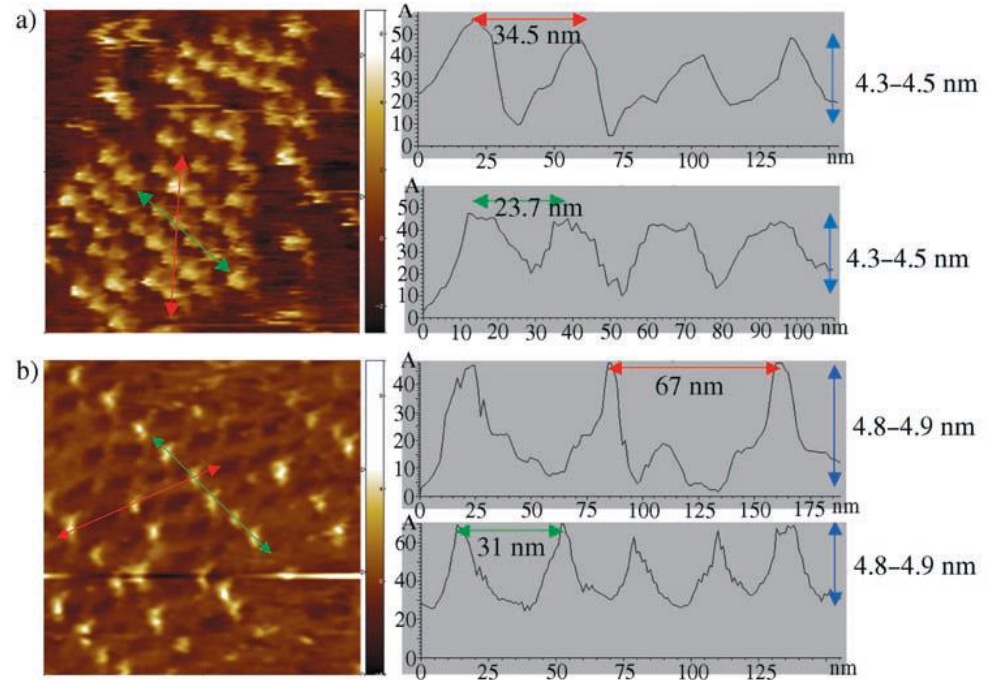
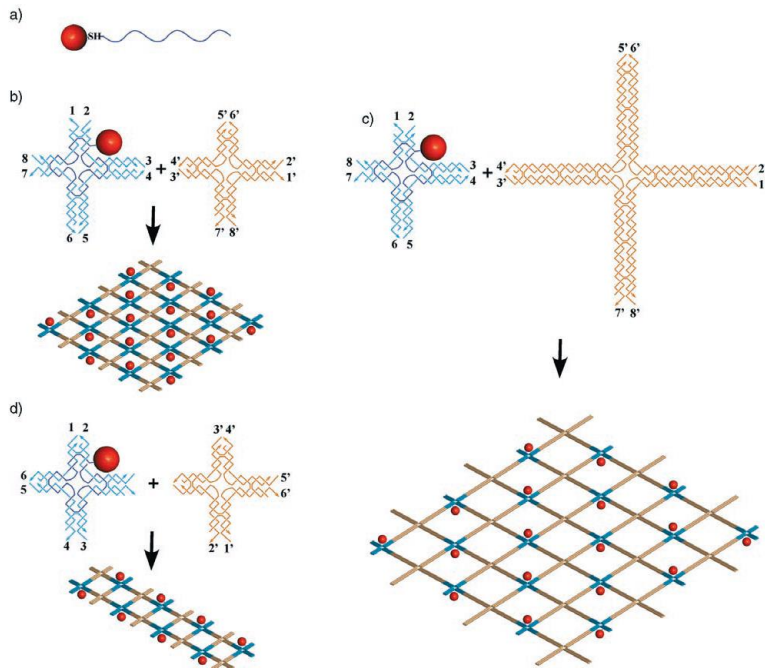
Patterning NPs on DNA nanostructures



800x800 nm

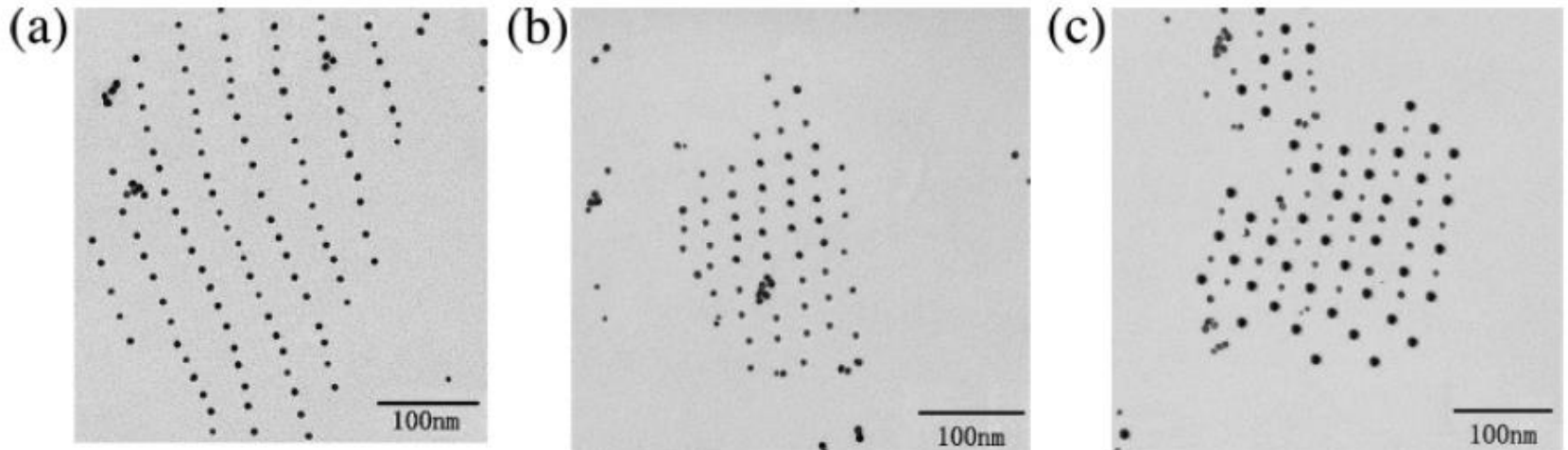
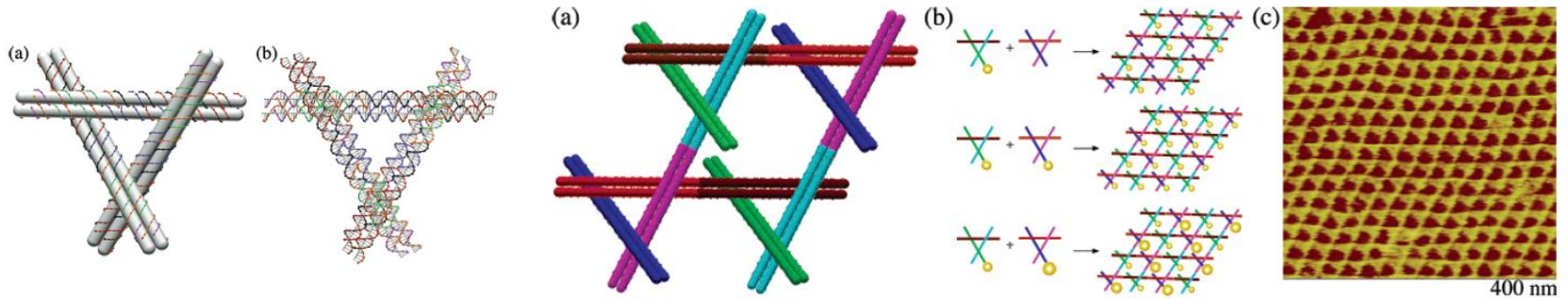
800x800 nm

800x800 nm

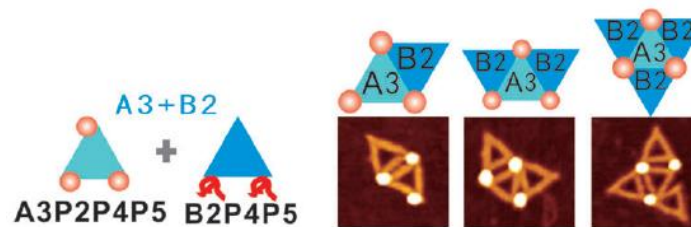
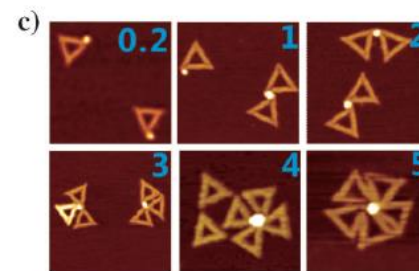
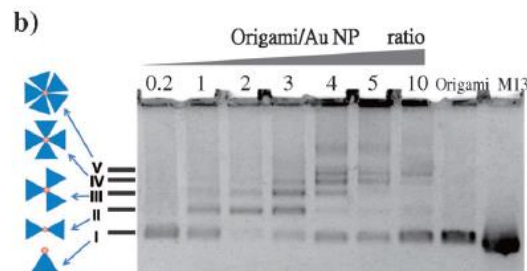
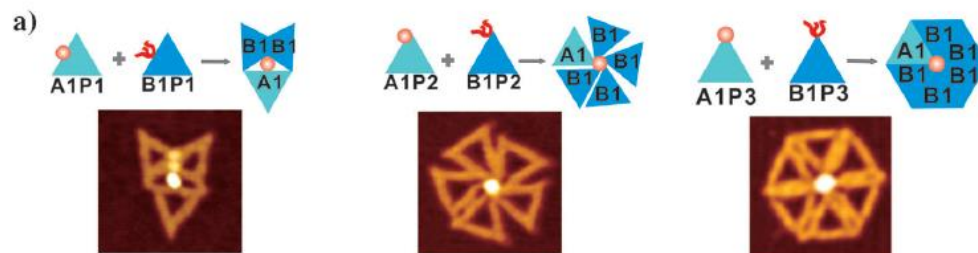
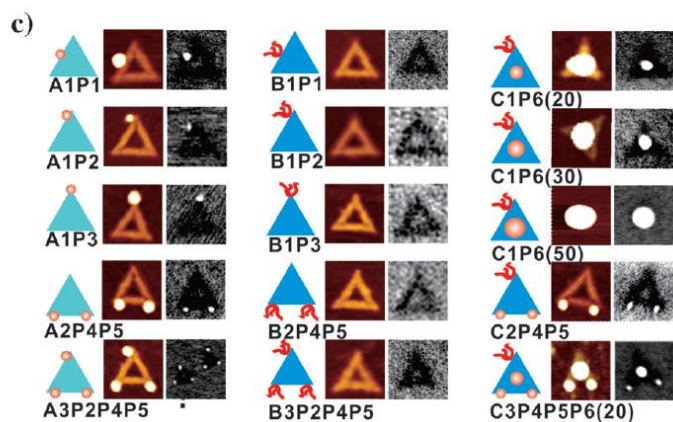
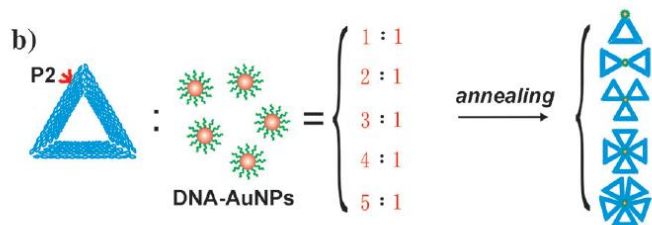
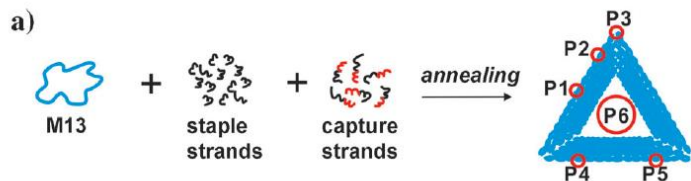


DNA nanostructures for NP patterning

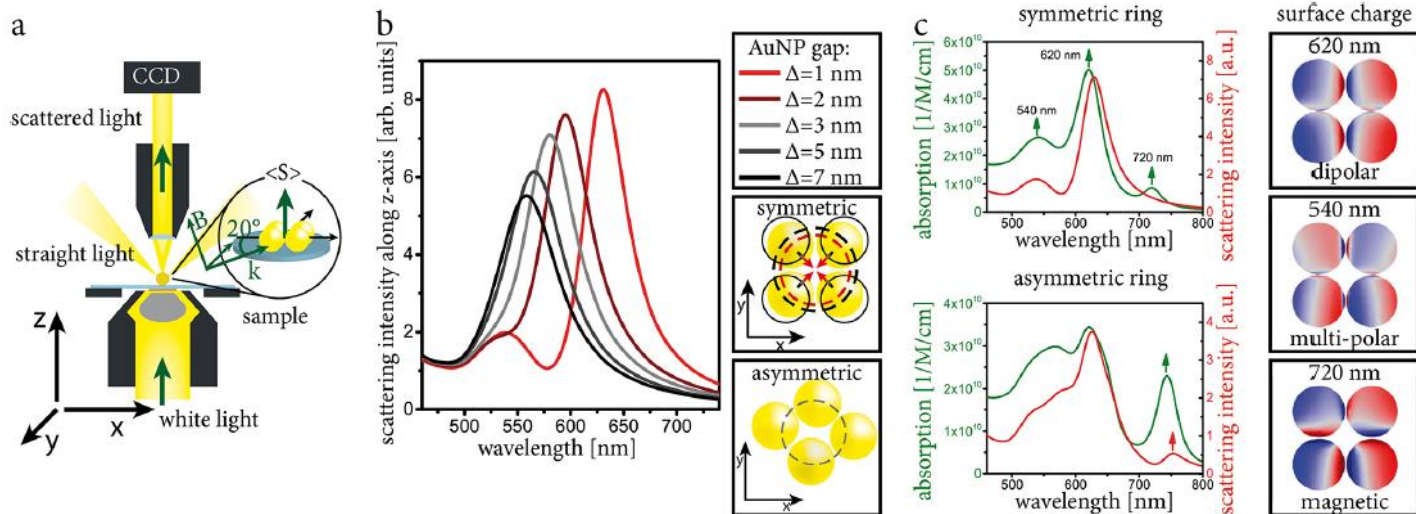
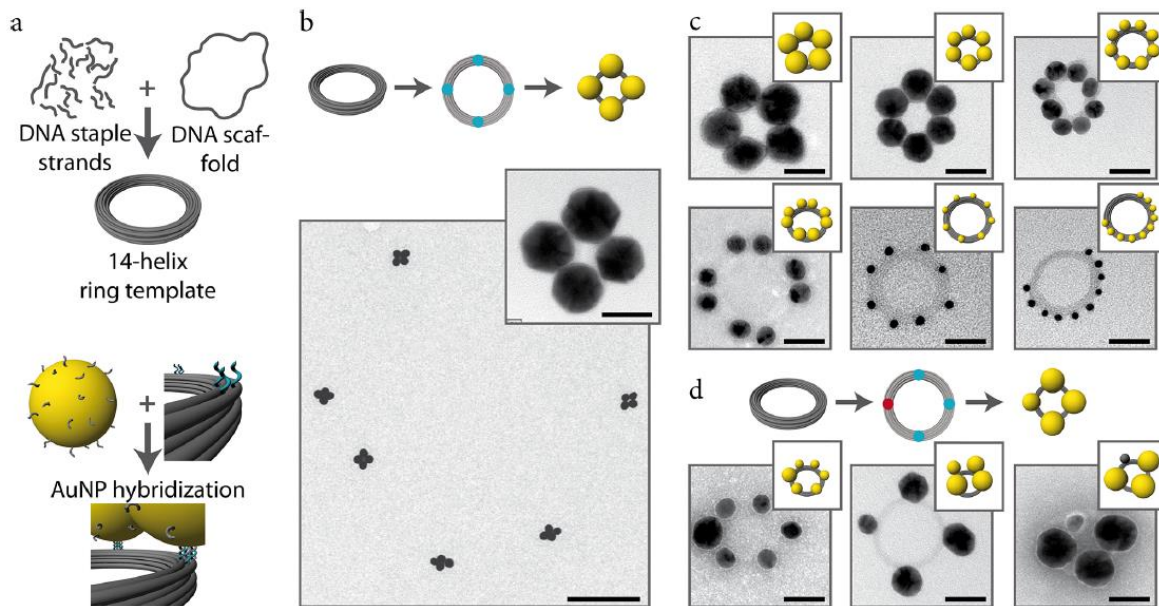
Patterning NPs on DNA nanostructures



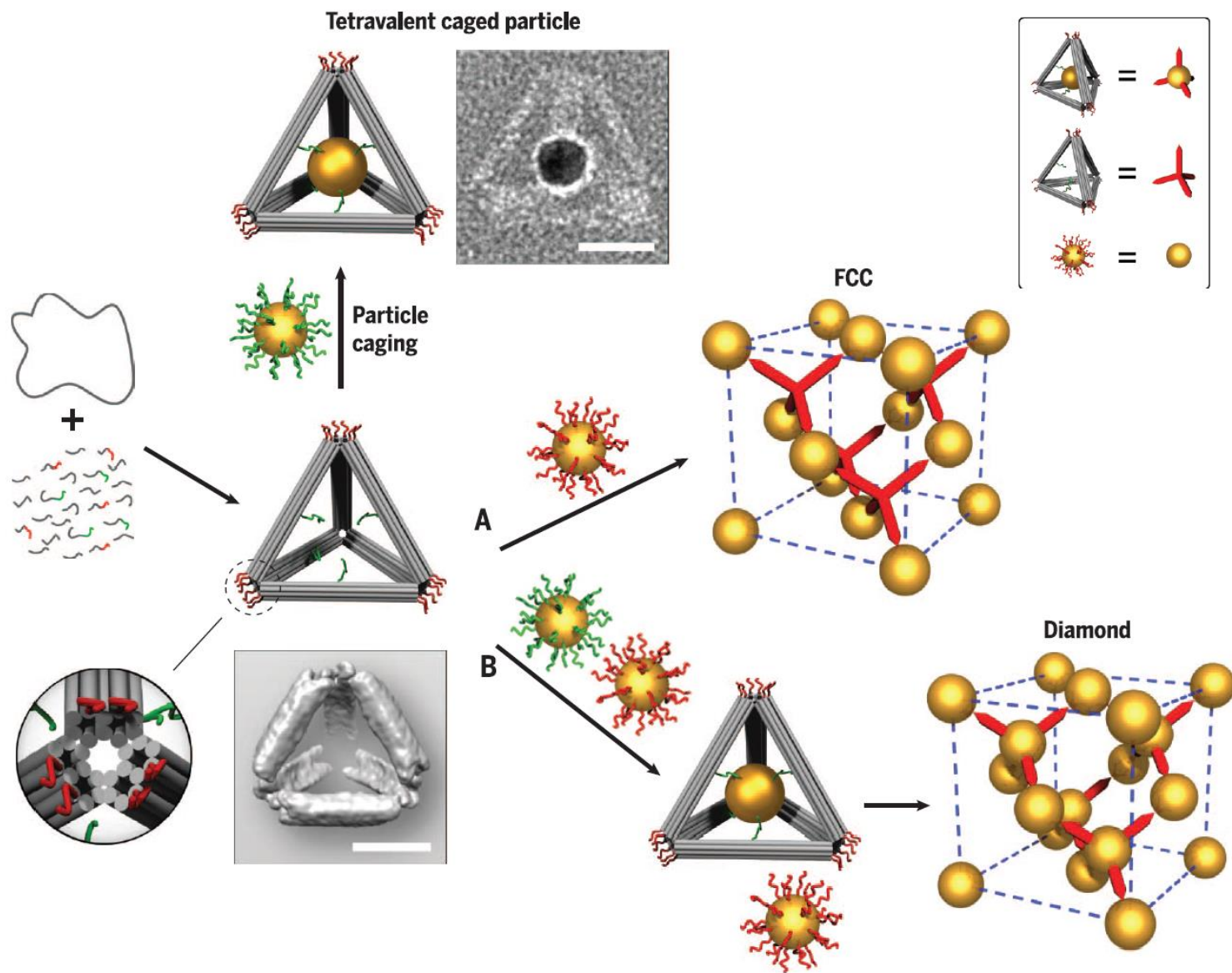
NP patterning on DNA origami



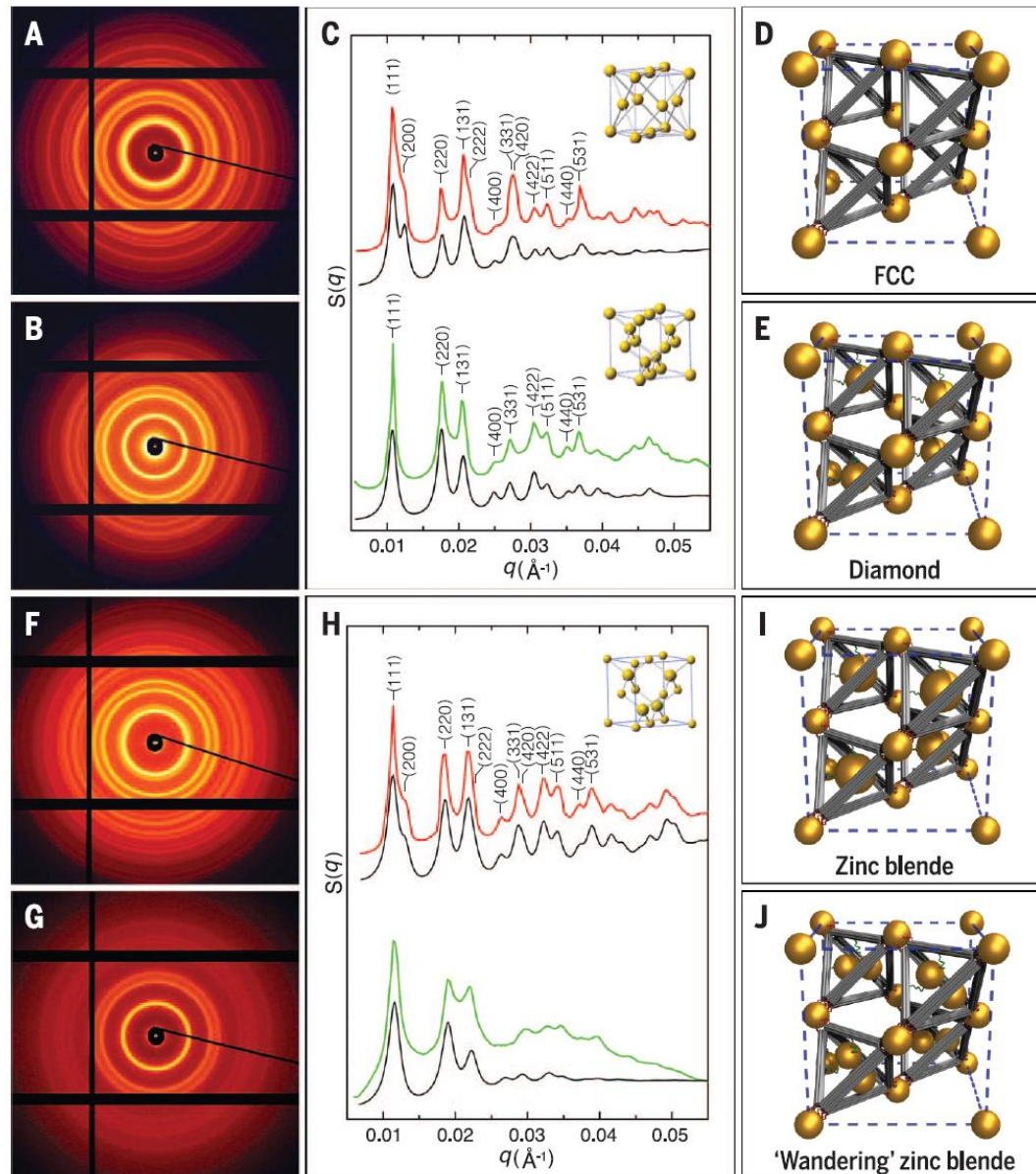
Self assembly of NPs using DNA origami



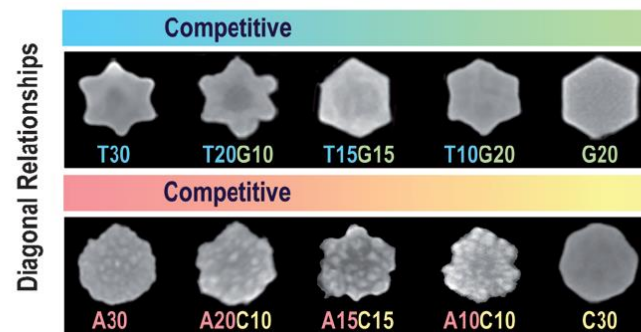
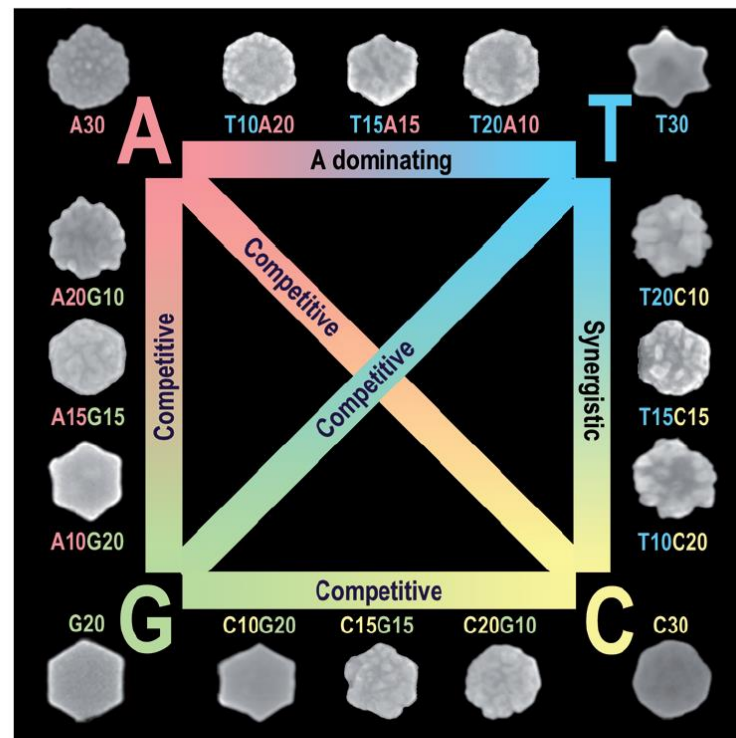
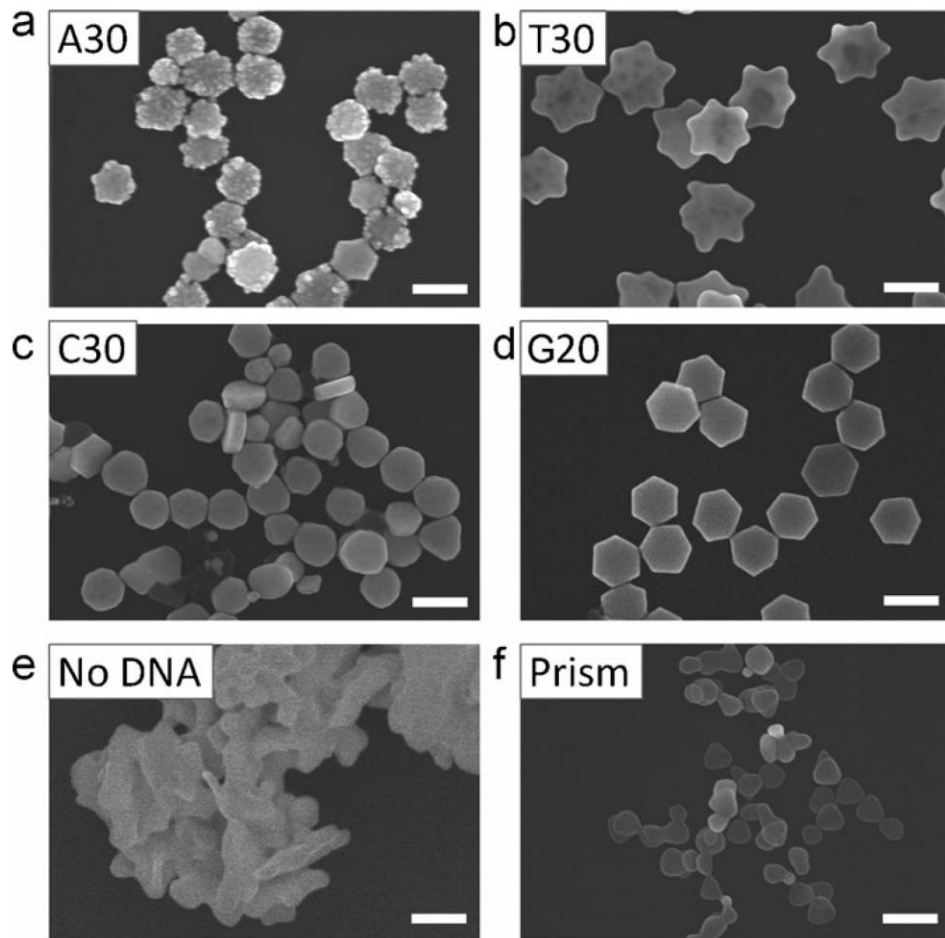
Lattice structures formed using DNA origami



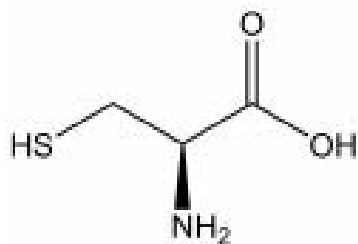
Lattice structures formed using DNA origami



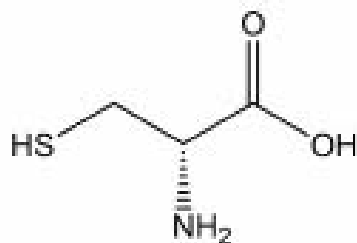
DNA base directed AuNP morphologies changes



Synthesis of chiral AuNPs using AA/peptides



L-cysteine



D-cysteine

Use chiral amino acids or peptide to guide synthesis of chiral materials

